



Comparison of Convective Heat Transfer Characteristics of Supercritical Fluid for Circular-Pipes in Horizontal Flow

A. Bamido (Texas A&M University, College Station, TAMU)

A. Thyagarajan (Texas A&M University College Station, TAMU)

V. K. Dhir (Texas A&M University College Station, TAMU; University of California, Los Angeles, UCLA)

V. Prasad (University of North Texas, UNT)

D. Banerjee (Texas A&M University, College Station, TAMU)

Introduction of Authors and Co-Authors



Alaba Bamido, PhD



Ashok Thyagrajan



Vijay K. Dhir, PhD



Vish Prasad, PhD



Debjyoti Banerjee, PhD

Outline

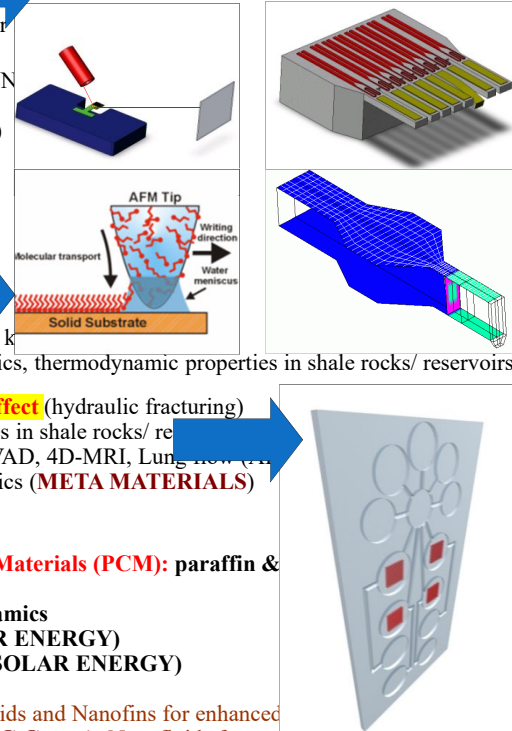
- INTRODUCTION: RESEARCH GROUP
- INTRODUCTION
- MOTIVATION AND GOAL OF STUDY
- METHODOLOGY (INVESTIGATION OF THERMOPHYSICAL PROPERTIES AND PIPE FLOW ANALYSIS)
- RESULTS AND DISCUSSION
- CONCLUSION
- ACKNOWLEDGEMENTS

Multiphase Flows & Heat Transfer Lab.

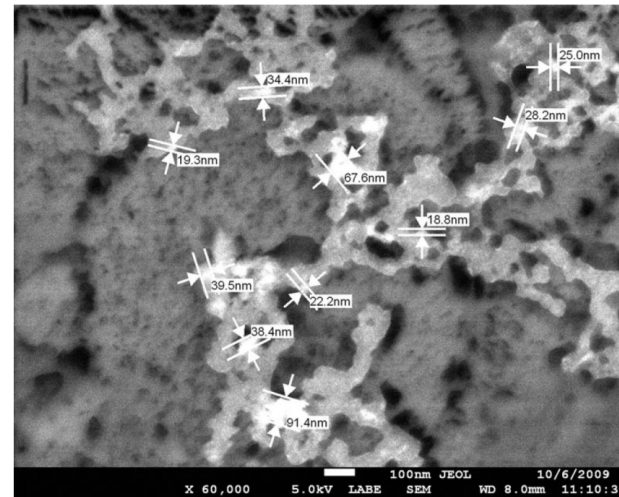
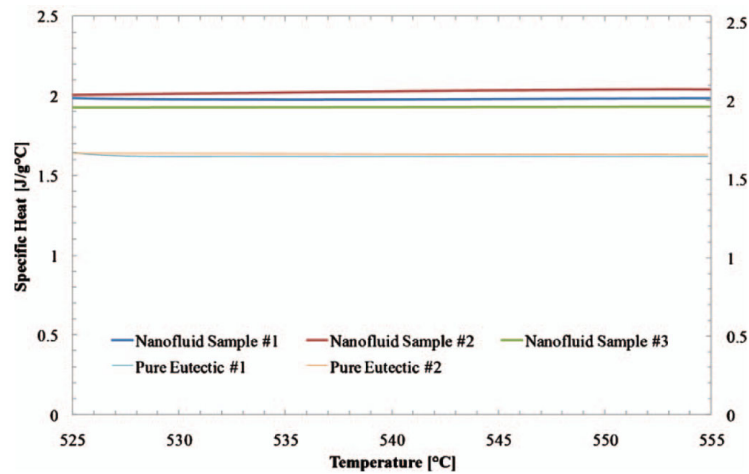
(Since Jan. 2005, 61 Extra-Mural Funded Projects in 15 years)



- **Nano-Sensors** (3 Ph.D., 1 M.S.)
 - **MEMS:**
 - **TEES:** Nano-calorimeter (**EXPLOSIVES SENSORS**)
 - **Nano-MEMS Research/NSF SBIR:** RF-MEMS, RF-Tuner
 - **DPN (Dip Pen Nanolithography):**
 - **DARPA/MTO:** chirality control, low temp. synthesis of CN
 - **ONR STTR/ ADA Tech.: Ultra-Capacitors**
 - **SPAWAR:** low temp. synthesis of Graphene (ONR/ASEE)
 - **TSGC:** Nanolithography+Microfluidics, CFD
 - **GE Research:** Silicon Nanofins
 - **Bio-Microfluidics, Lab-On-Chip:**
 - **AFRL:** Portable water quality monitor
 - **DARPA/MF³:** Micro-Chamber Filling
 - » (1) Vaccine Storage/ Paper microfluidics
 - » (2) Anthrax Detection using CD Microfluidics
 - **NASA:** Lipid bi-layer sensors for studying protein/peptide kinetics
 - **Energy Institute, Hagler Institute, T3:** Micro/nano-fluidics, thermodynamic properties in shale rocks/ reservoirs, desalination
 - **Crisman Institute:** Micro/nano-fluidics, **confined fluid effect** (hydraulic fracturing)
 - **Marathon:** Micro/nano-fluidics, thermodynamic properties in shale rocks/ reservoirs
 - **ENMED:** Neural Organoids (organoid-on-Chip/ OoC), LVAD, 4D-MRI, Lung flow (AFRL)
 - **AFOSR:** Reconfigurable microfluidic device for nano-optics (**META MATERIALS**)
- **Thermal Management** (3 Ph.D., 1 M.S.)
 - **Nanostructures**
 - **ARPA-E (ARID):** Energy-water nexus, **Phase Change Materials (PCM):** paraffin & organics
 - **ONR:** Flow Boiling on Carbon Nanotubes
 - **NSF:** Pool Boiling on Silicon Nanofins, Molecular Dynamics
 - **DOE:** Nanofluids for Thermal Energy Storage (**SOLAR ENERGY**)
 - **GE-Alstom:** Nanofluids for Thermal Energy Storage (**SOLAR ENERGY**)
 - **AFOSR/AFRL(ASEE-SFFP):** Nano-Fluids
 - **Qatar National Research Foundation (QNRF):** Nanofluids and Nanofins for enhanced heat transfer
 - **Photonics Corp./ Trianja Tech. (Si Values Partners/ B G Group):** Nanofluids for energy app.
 - **Irvine Sensors:** Micropump Design for Electronics Cooling (**AFRL SBIR Phase II**)
 - **Aspen Thermal Systems:** Compact condensers (**ONR SBIR Phase I; Phase II**)
 - **Thermo-Chemical Energy Storage (TCES), Thermal Energy Storage (TES)**
 - **AFRL:** Chevron Plate Heat Exchangers (PHE) for rapid thermal management
 - **Biomedical Device (Surgical Sterilizer)**
 - **Lynntech/ ARO SBIR Phase II:** Portable sterilizer for surgical tools using steam.



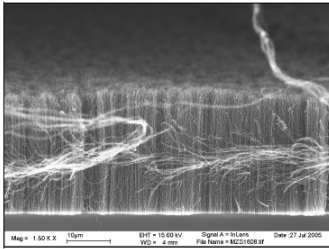
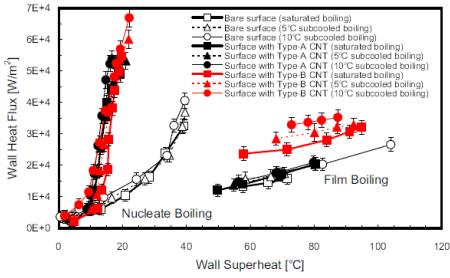
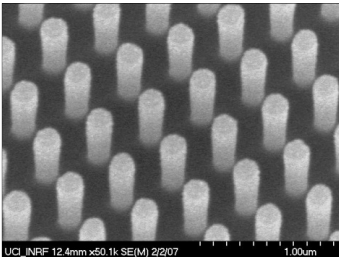
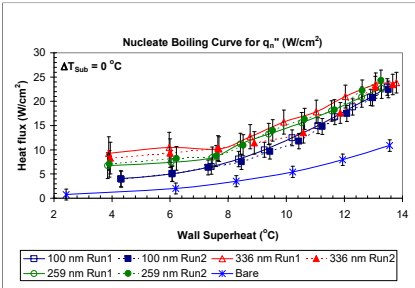
Molten salt nanofluids



Carbonate salt eutectic + SiO₂
26 % enhancement

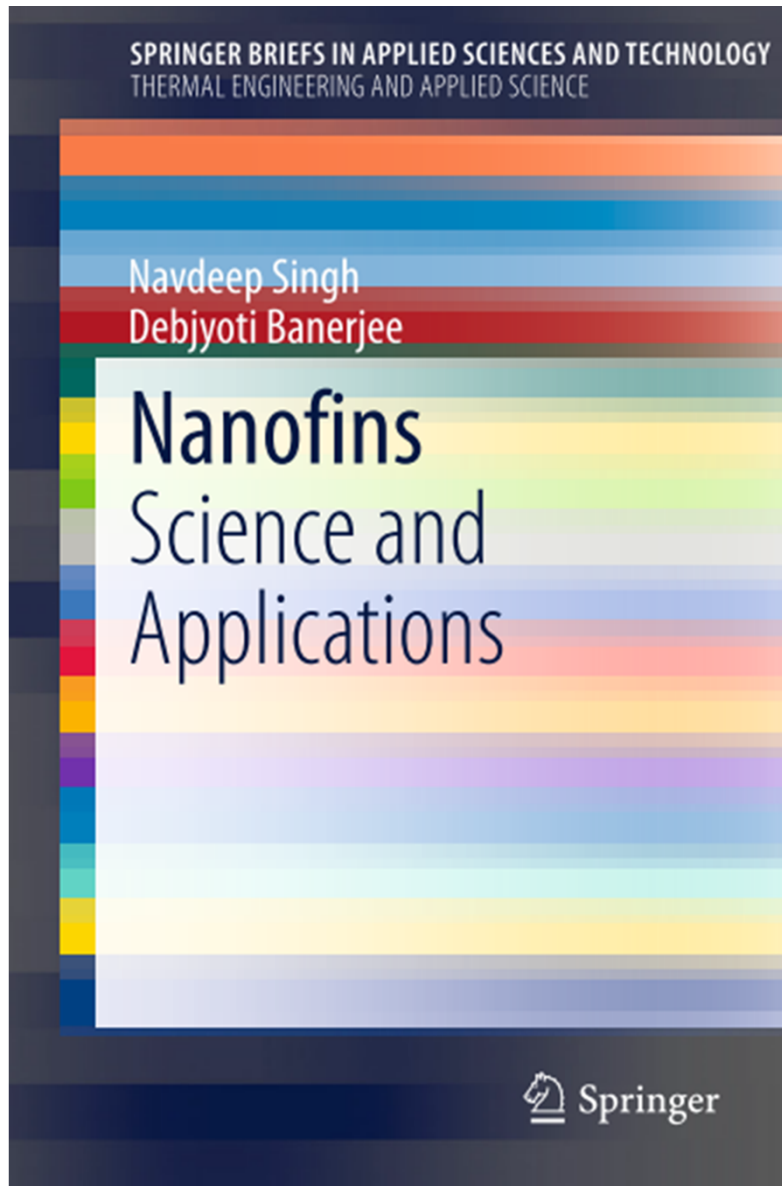
Shin and Banerjee, J Heat Trans., 2011

Pool Boiling on Nano-Fins

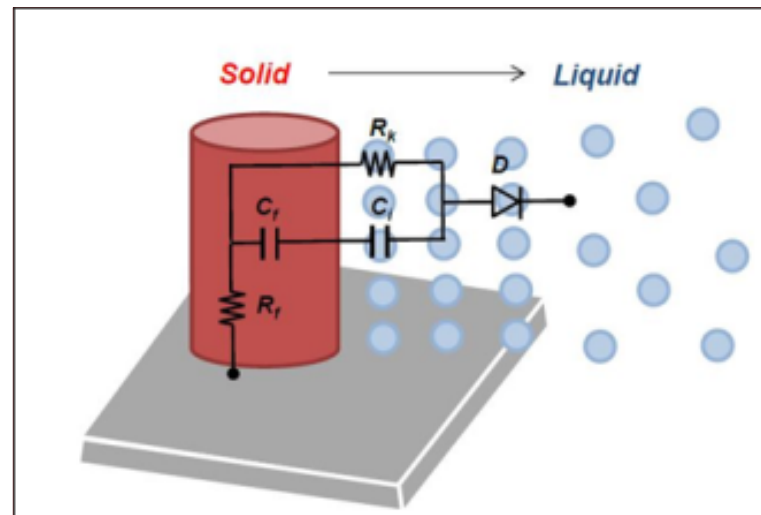
<i>Nanostructure</i>	<i>Boiling curve (large heater)</i>	<i>CHF enhancement</i>
<p style="text-align: center;">CNT</p>  <p style="font-size: 8px; margin-top: 5px;">Mag = 1.00 K X 10µm HV = 15.00 kV Signal = 111.0kV Date = 27 Jul 2006 WD = 4.0mm File Name = M2210281F</p>	 <p style="font-size: 8px; margin-top: 5px;">Legend: -□- Bare surface (saturated boiling) -○- Bare surface (5 °C subcooled boiling) -△- Bare surface (10 °C subcooled boiling) -■- Surface with Type A CNT (saturated boiling) -●- Surface with Type A CNT (5 °C subcooled boiling) -▲- Surface with Type A CNT (10 °C subcooled boiling) -◆- Surface with Type B CNT (saturated boiling) -◇- Surface with Type B CNT (5 °C subcooled boiling) -◆- Surface with Type B CNT (10 °C subcooled boiling) </p>	<p>~60 % [A]</p>
<p style="text-align: center;">Silicon</p>  <p style="font-size: 8px; margin-top: 5px;">UCI_NRF 12.4mm x50.1k SE(M) 22/07 1.00µm</p>	 <p style="font-size: 8px; margin-top: 5px;">Nucleate Boiling Curve for q_w (W/cm²) $\Delta T_{sub} = 0$ °C Legend: -□- 100 nm Run1 -■- 100 nm Run2 -▲- 336 nm Run1 -◆- 336 nm Run2 -◇- 259 nm Run1 -●- 259 nm Run2 -×- Bare </p>	<p>~120 % [B]</p>

[A] H. Ahn, Heat transfer enhancement in single-phase forced convection with blockages and in two-phase pool boiling with nano-structured surface, Ph. D dissertation, Texas A&M University, 2007

[B] Sriraman, Pool boiling on nano-finned surfaces, Master thesis, Texas A&M University, 2007



“*nano-Fin Effect*” (nFE)



Corrosion reduced by 50~100%

Lee, J. *, Kuchibhotla, A. *, Banerjee, D., and Berman, D.,
“*Silica Nanoparticles as copper corrosion inhibitors*”,
Materials Research Express, May, 2019.

<https://doi.org/10.1088/2053-1591/ab2270>

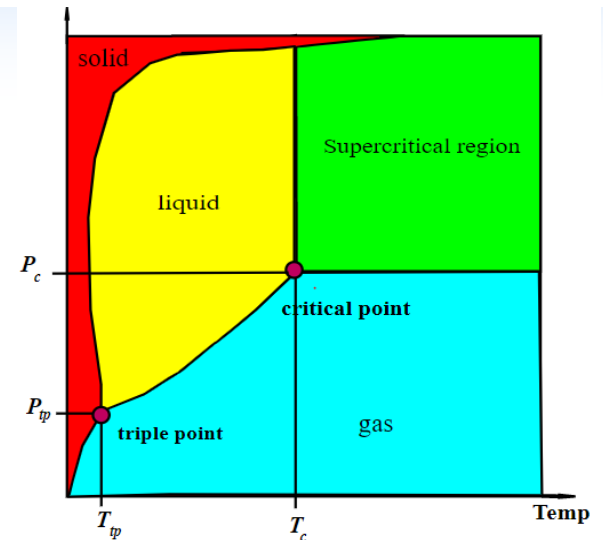
Introduction

Supercritical fluids exist at temperatures and pressures exceeding the critical point.

They can exist as “liquid-like” or “gas-like” states, and the differentiator between the two states is also known as the “*Widom Line*” / “*Frenken Line*”.

Critical properties of CO_2

Critical pressure	7.38 [MPa]
Critical Temperature	304.1 [K]



P-T Phase diagram

Fomin, Y. D., Ryzhov, V. N., Tsiok, E. N. & Brazhkin, V. V. Thermodynamic properties of supercritical carbon dioxide: Widom and Frenkel lines. Phys Rev E 91,2015.

<https://doi.org/10.1103/PhysRevE.91.022111>

Introduction



Cleaning

Ref: The News



Biomedical

Ref: Xtalks



Aerogel production

Ref: NASA Stardust Website



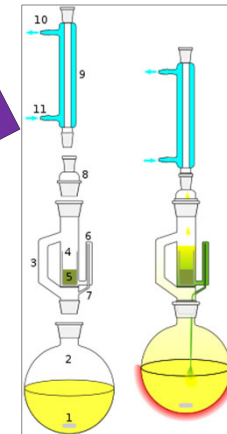
Power Generation

Ref: Flyability



Manufacturing

Market Business News



Extraction

Ref: Soxhlet extractor

sCO_2

The 7th International Supercritical CO_2 Power Cycles • February 21 – 24, 2022 • San Antonio, TX, USA

Motivation and Goal of Study

➤ Motivation

The motivation of this study was to explore the efficacy of heat exchangers involving forced convective heat transfer of supercritical fluids (tube side) integrated with air cooling (i.e., in free convection).

➤ Goal

The goal of this study was to determine the forced convective heat transfer characteristics of supercritical carbon dioxide (sCO₂) in air-cooled tube heat-exchangers.

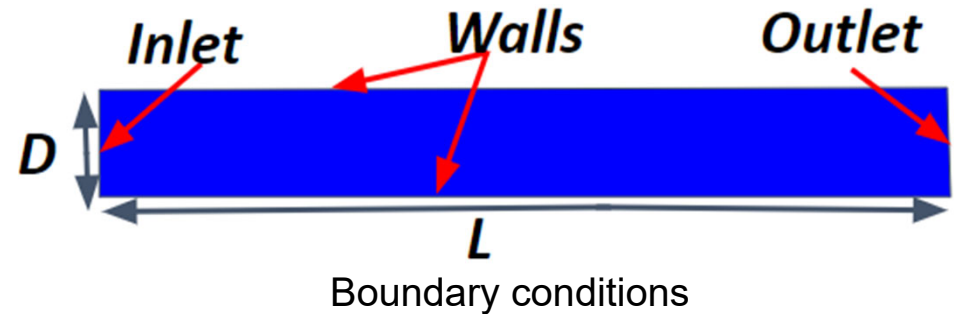
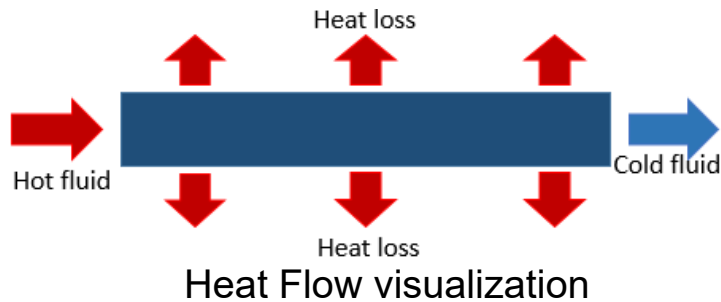
Methodology (Investigation of Thermophysical properties of sCO₂)

The thermophysical properties of sCO₂ was obtained from NIST database for pressure, $P=8.12\text{MPa}$ and temperature variation, $T=450 - 700[\text{K}]$. Properties investigated are:

- Specific heat at constant volume, c_v
- Specific heat at constant pressure, c_p
- Density, ρ
- Thermal conductivity, k
- Dynamic viscosity, μ
- Kinematic viscosity, ν
- Thermal diffusivity, α
- Prandtl number, Pr

Methodology (Pipe Flow Analysis)

2D Numerical and Analytical investigation of the hydrodynamic and thermal characteristics of sCO_2 flowing in a circular horizontal pipe.



Boundary conditions and fluid properties

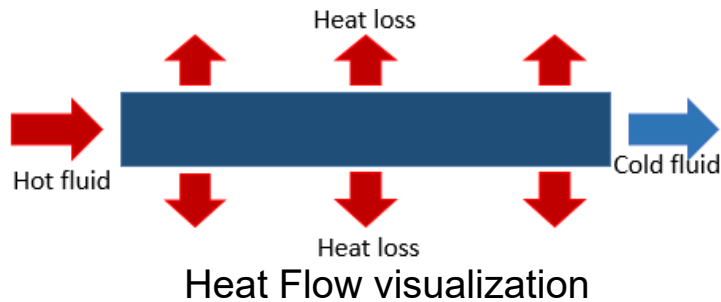
<i>Re</i>	10, 100, 1000, 1500, 5000, 10 000
<i>Reduced Press., P_r</i>	1.1
<i>Reduced Temp., T_r</i>	1.64
<i>Inlet Temp., T_{in}</i>	700 [K]
<i>Ambient Temp., T_∞</i>	300 [K]
<i>External convection coefficient, h_o</i>	43.2 [W/m ² K]
<i>Density, ρ</i>	90.817 [kg/m ³]
<i>Specific heat cap., c_p</i>	1132.2 [J/kgK]
<i>Thermal cond., k</i>	0.036361 [W/mK]
<i>Dynamic visc., μ</i>	$2.5e^{-5}$ [Pa.s]

Dimensions

Diameter (D)	1, 2, 5 and 10 [mm]
Length (L)	1 [m]

Methodology (Pipe Flow Analysis)

2D Numerical and Analytical investigation of the hydrodynamic and thermal characteristics of sCO_2 flowing in a circular horizontal pipe. Numerical simulation performed with ANSYS Fluent 2019R19.1



Boundary conditions and fluid properties

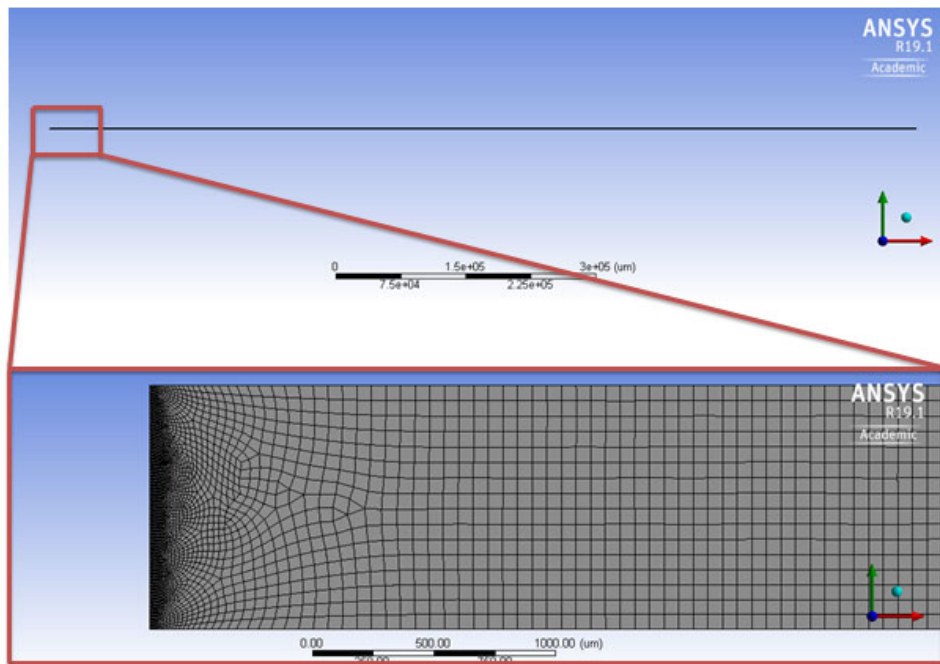
<i>Re</i>	10, 100, 1000, 1500, 5000, 10 000
<i>Reduced Press., P_r</i>	1.1
<i>Reduced Temp., T_r</i>	1.64
<i>Inlet Temp., T_{in}</i>	700 [K]
<i>Ambient Temp., T_∞</i>	300 [K]
<i>External convection coefficient, h_o</i>	43.2 [W/m ² K]
<i>Density, ρ</i>	90.817 [kg/m ³]
<i>Specific heat cap., c_p</i>	1132.2 [J/kgK]
<i>Thermal cond., k</i>	0.036361 [W/mK]
<i>Dynamic visc., μ</i>	$2.5e^{-5}$ [Pa.s]

Model and Assumptions

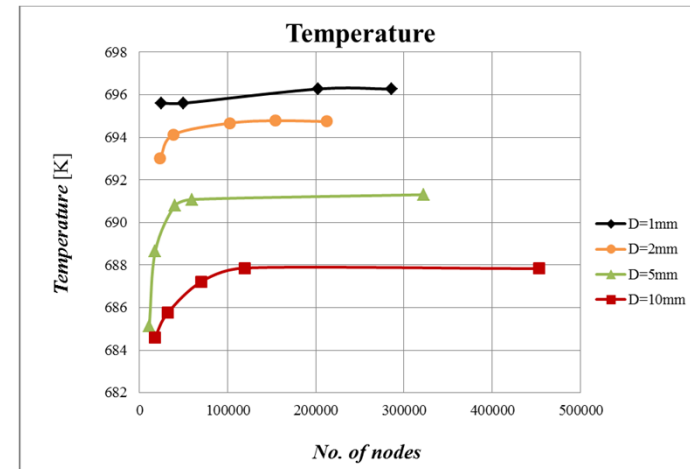
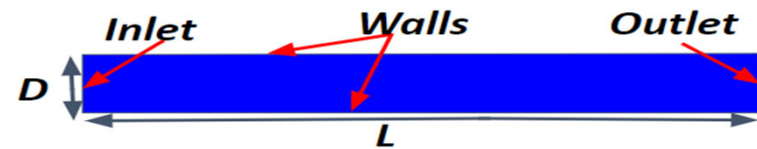
- Pressure based solver
- Steady State
- Constant properties (Incompressible flow)
- Constant ambient condition (synonymous to constant surface temp condition)
- Laminar flow for Re (10, 100, 1000 and 1500);
Turbulent flow for Re (5000 and 10000) with $k-w$ SST (shear stress transport) model
- Energy on

Methodology (Pipe Flow Analysis)

2D Numerical and Analytical investigation of the hydrodynamic and thermal characteristics of sCO_2 flowing in a circular horizontal pipe. Numerical simulation performed with ANSYS Fluent 2019R19.1

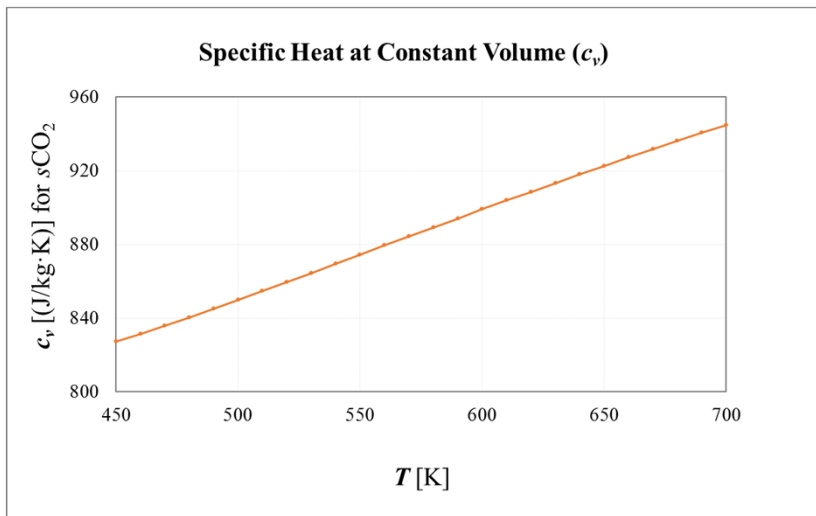


A quadrilateral mesh with refinement at the inlet and outlet sections

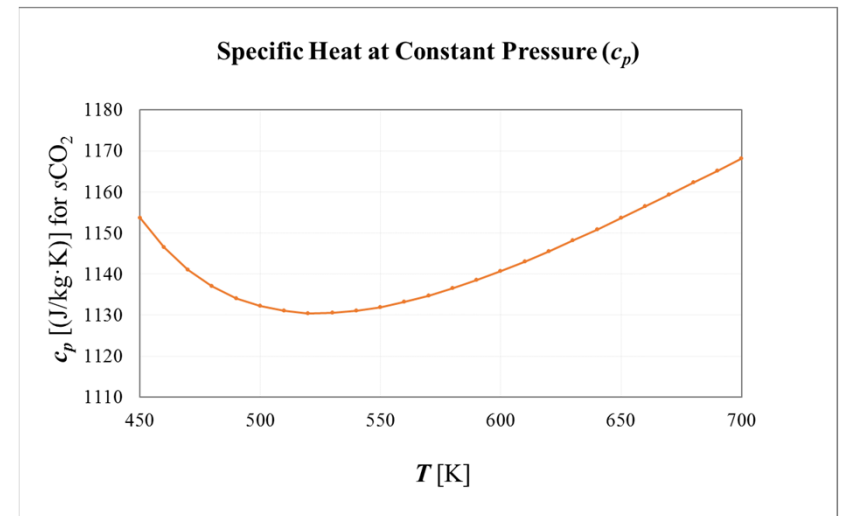


Mesh refinement was performed until there was no significant change to bulk mean temperatures

Results (Investigation of Thermophysical properties of sCO₂)

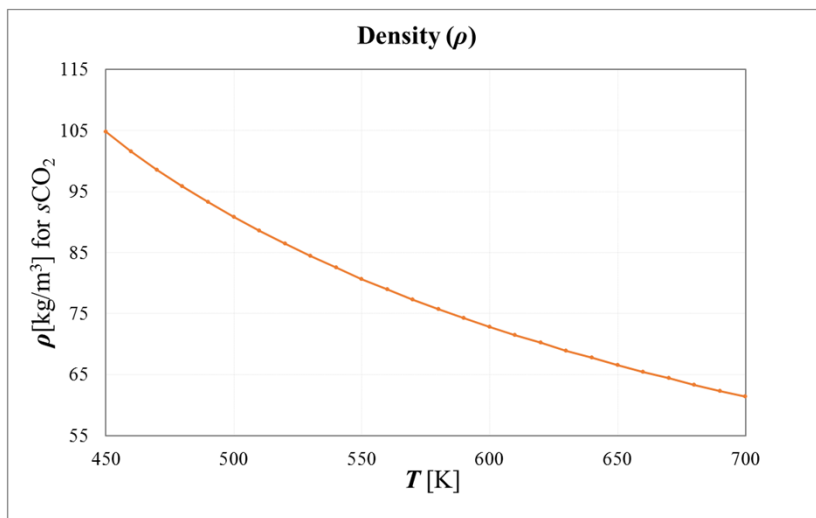


c_v ↑ as T ↑ for $450 \leq T \leq 700$

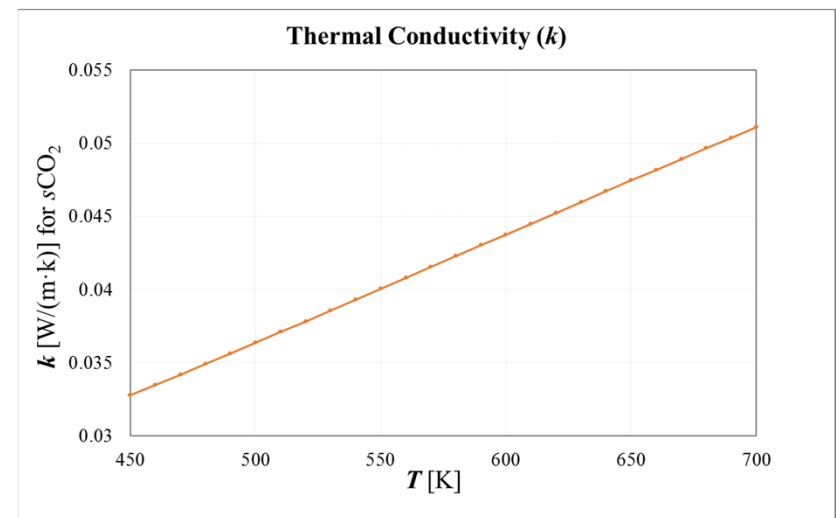


c_p ↓ as T ↑ for $450 \leq T \leq 520$
 c_p ↑ as T ↑ for $520 \leq T \leq 700$

Results (Investigation of Thermophysical properties of sCO₂)

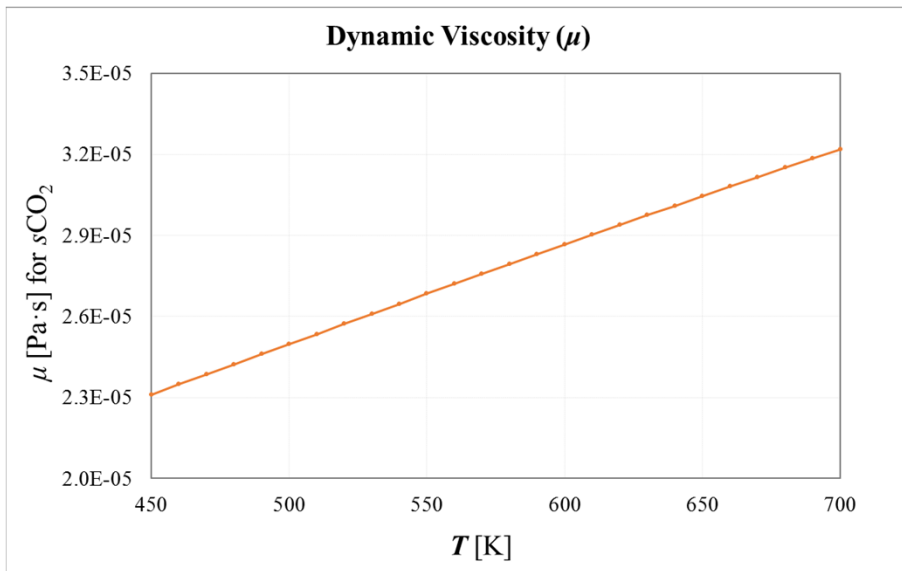


ρ ↓ as T ↑ for $450 \leq T \leq 700$

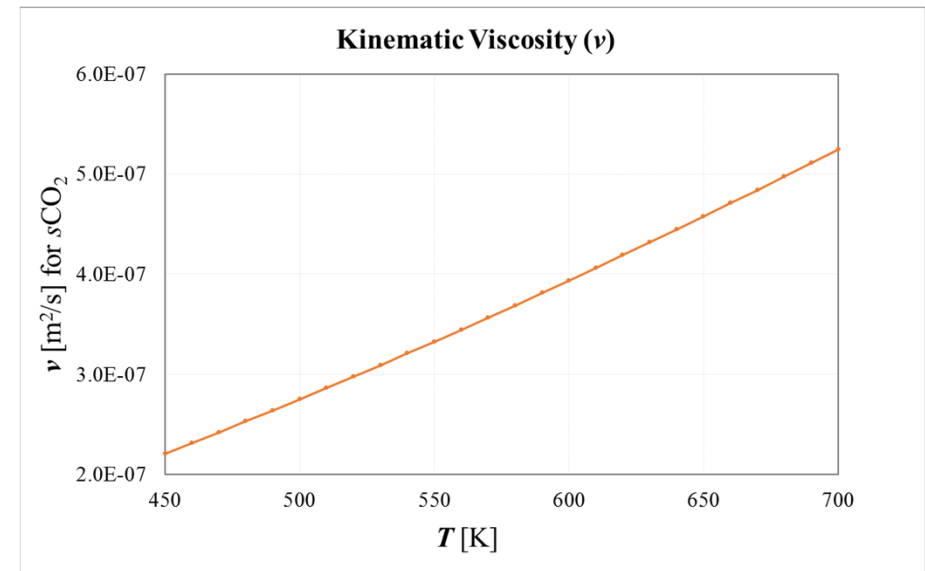


k ↑ as T ↑ for $450 \leq T \leq 700$

Results (Investigation of Thermophysical properties of sCO₂)

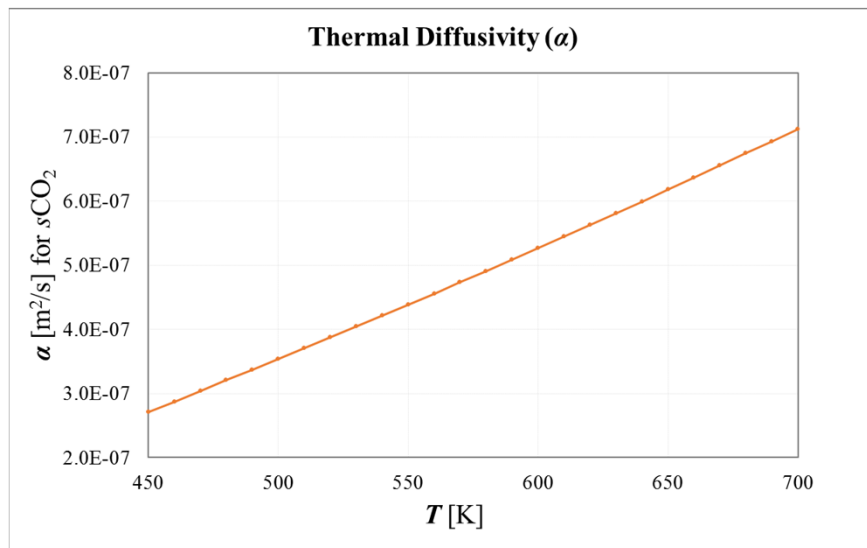


$\mu \uparrow$ as $T \uparrow$ for $450 \leq T \leq 700$

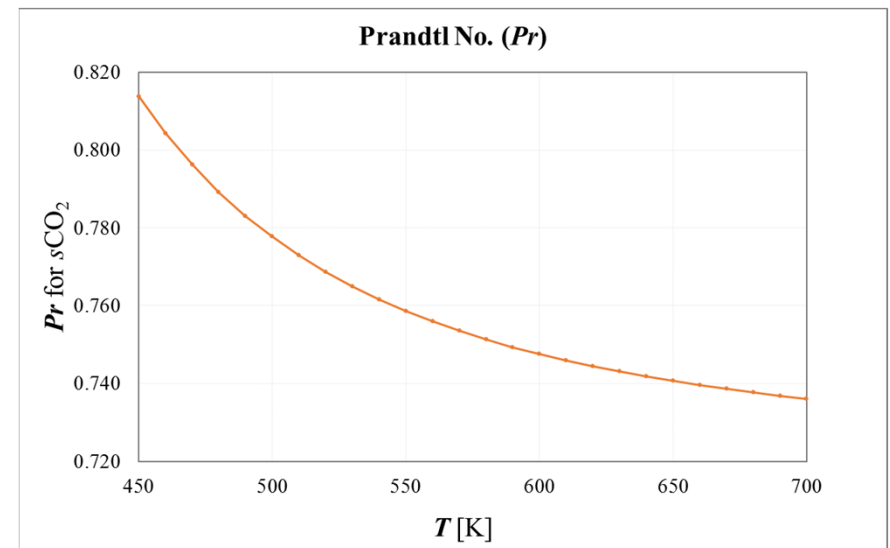


$\nu \uparrow$ as $T \uparrow$ for $450 \leq T \leq 700$

Results (Investigation of Thermophysical properties of sCO₂)

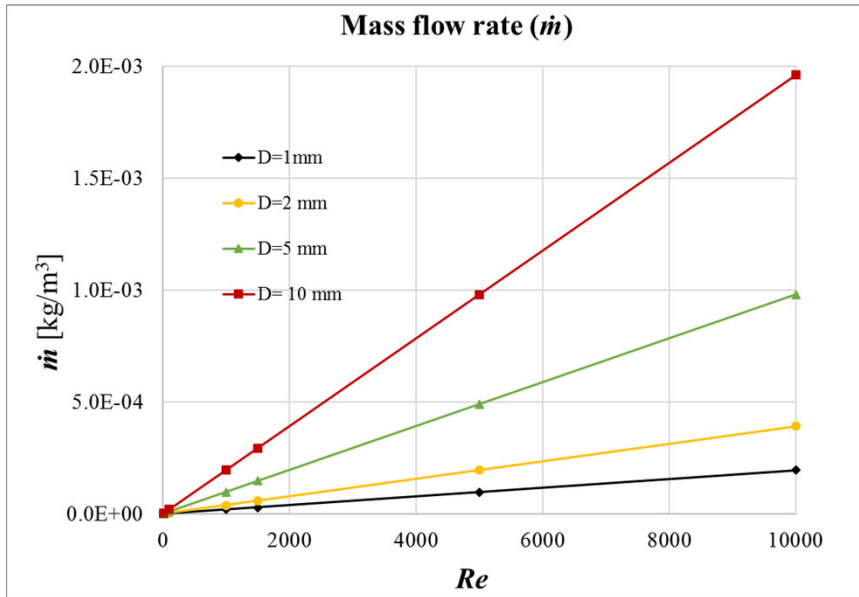


α ↑ as T ↑ for $450 \leq T \leq 700$

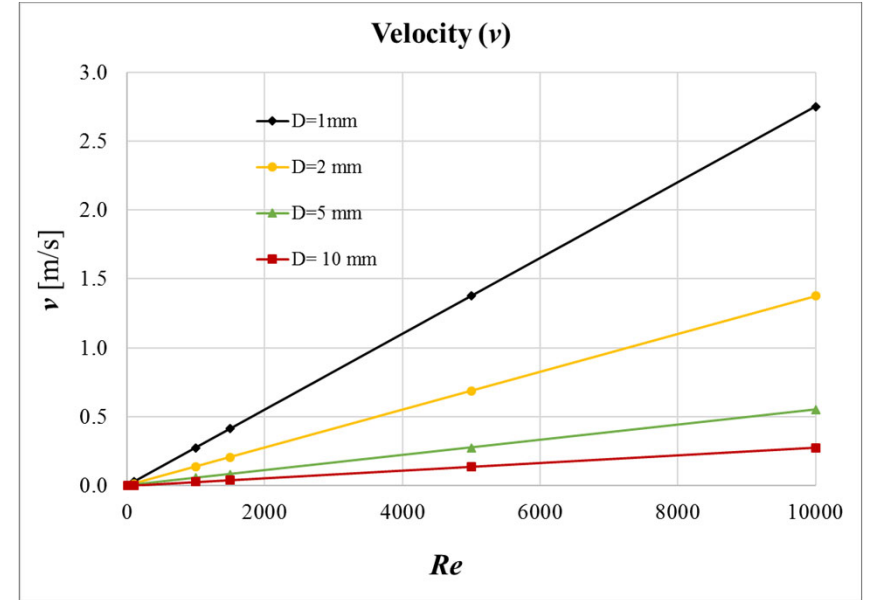


Pr ↓ as T ↑ for $450 \leq T \leq 700$

Results (Pipe Flow Analysis)



$\dot{m} \uparrow$ as $Re \uparrow$ and $D \uparrow$



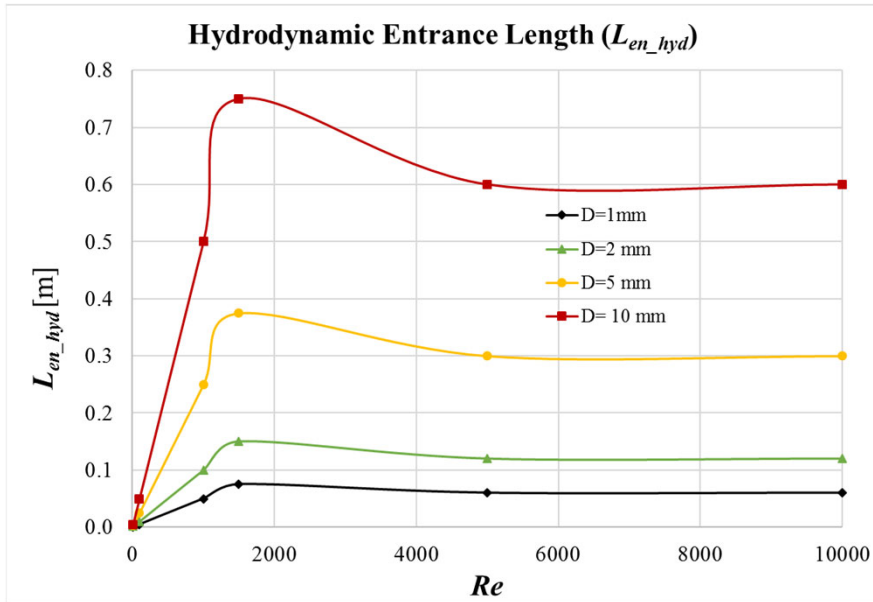
$v \uparrow$ as $Re \uparrow$ and $D \downarrow$

$$\dot{m} = Re \pi D \mu / 4$$

$$v = \dot{m} / (\rho A)$$

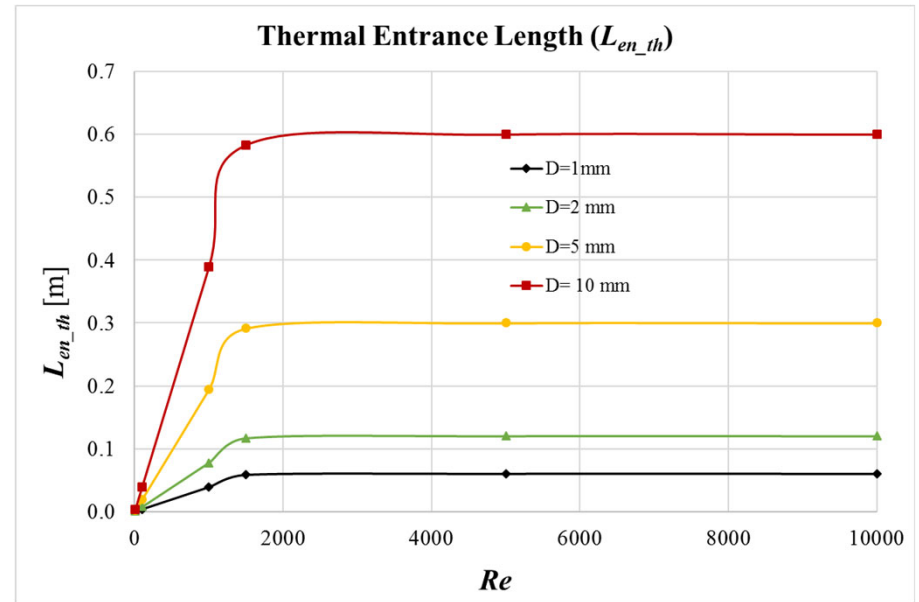
Incompressible flow
assumption

Results (Pipe Flow Analysis)



$$L_{en_hyd} = 0.05Re \cdot D \text{ (for laminar)}$$

$$L_{en_hyd} = 60D \text{ (for turbulent flow)}$$



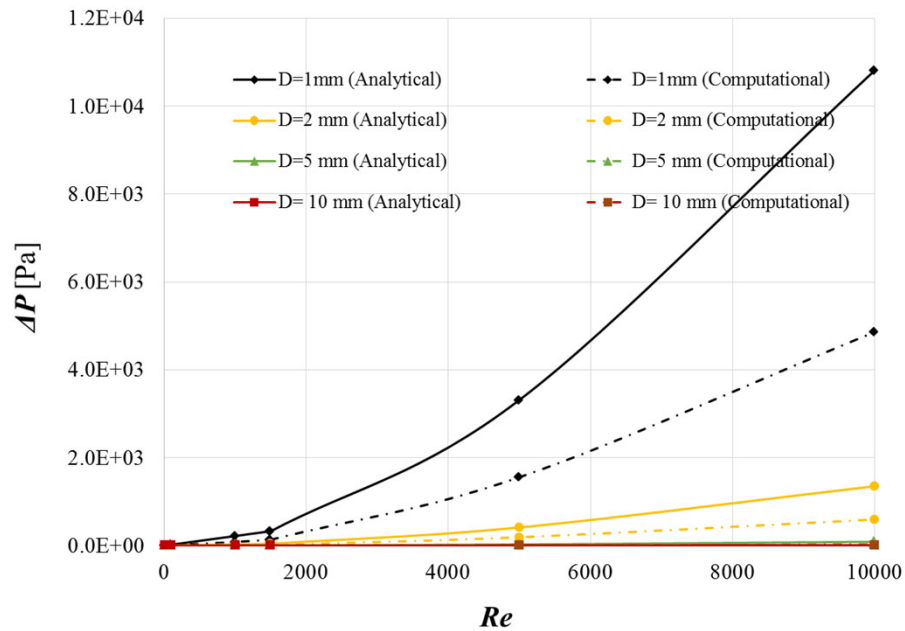
$$L_{en_th} = 0.05Re \cdot D \cdot Pr \text{ (for laminar)}$$

$$L_{en_th} = 60D \text{ (for turbulent)}$$

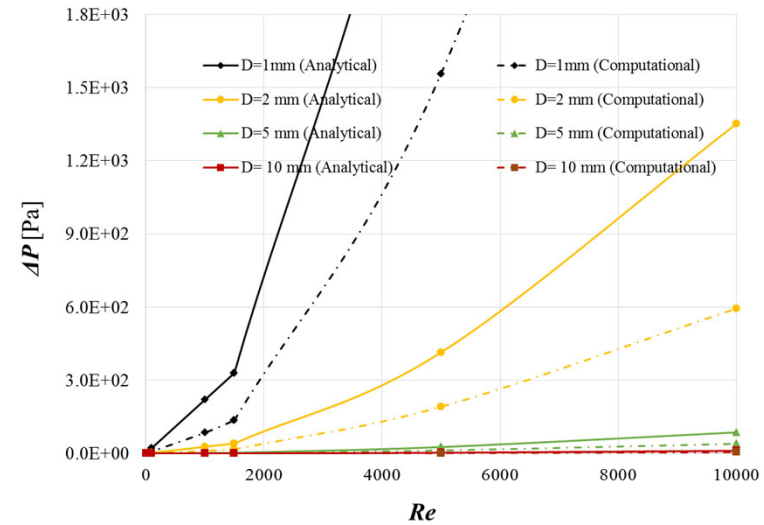
- In laminar region, $L_{en_hyd} \propto Re$ and D ; $L_{en_th} \propto Re, D$ and Pr
- In turbulent region, L_{en_hyd} and $L_{en_th} \propto D$
- In laminar region $L_{en_hyd} > L_{en_th}$, $\therefore Pr < 1$

Results (Pipe Flow Analysis)

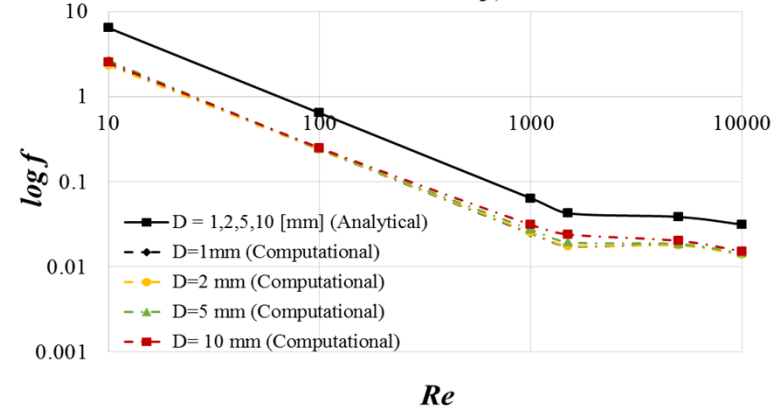
Pressure Drop (ΔP)



Pressure Drop (ΔP)



Friction factor (f)



$$f = 64 / Re, \quad Re < 2000 \text{ (for laminar flow)}$$

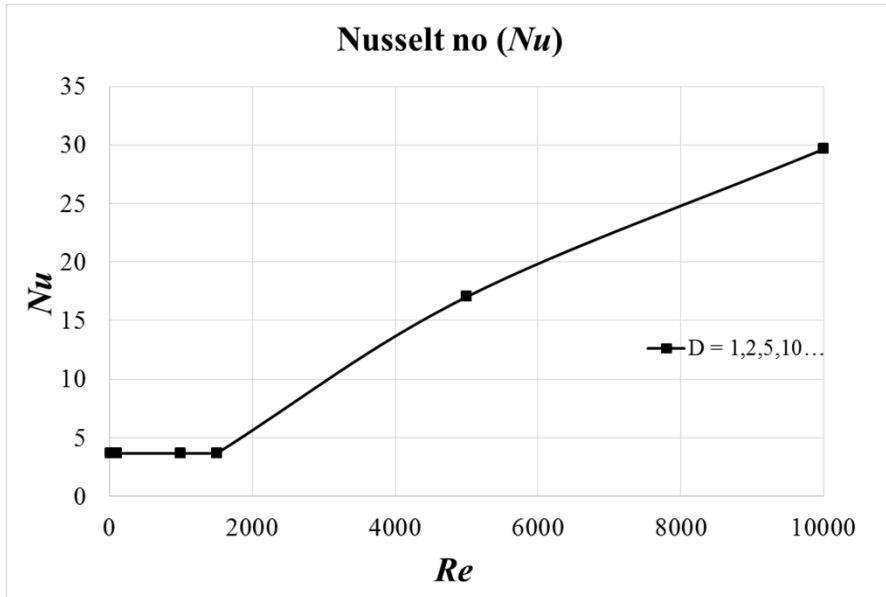
$$f = (0.79 \ln(Re) - 1.64)^{-2} \text{ for } 3000 < Re < 5 \times 10^6$$

$$\Delta P = f \frac{L \rho v^2}{D} \frac{1}{2}$$

➤ In laminar and turbulent region, $f \propto \frac{1}{Re}$

➤ Larger tubes give smaller pressure drop and vice versa.

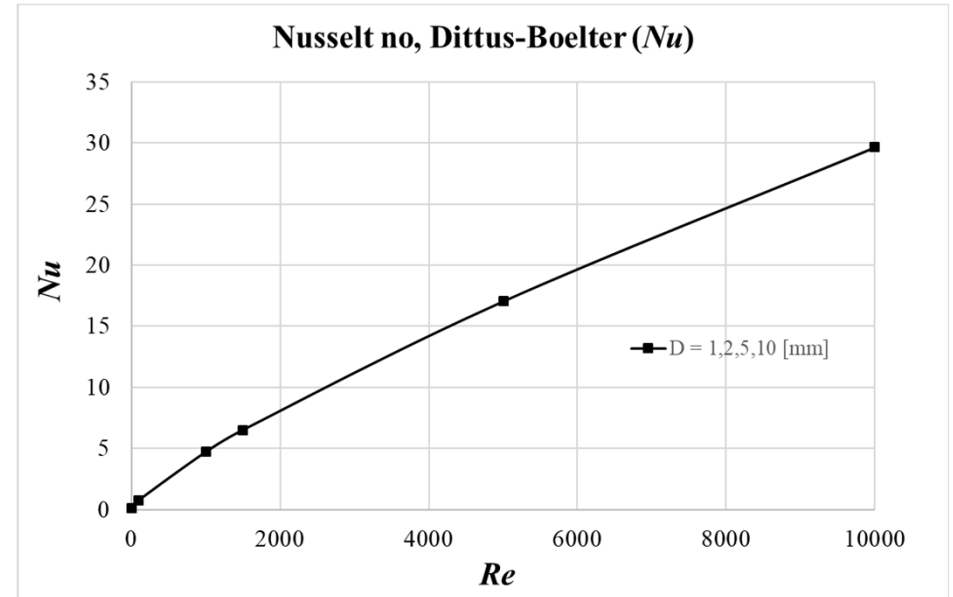
Results (Pipe Flow Analysis)



$Nu = 3.66$ used for the laminar regime (fully developed flow)

$$Nu = 0.023Re^{\frac{4}{5}}Pr^n \left(\frac{\mu_b}{\mu_s}\right)^{0.14} \quad (n = 0.3 \text{ for cooling})$$

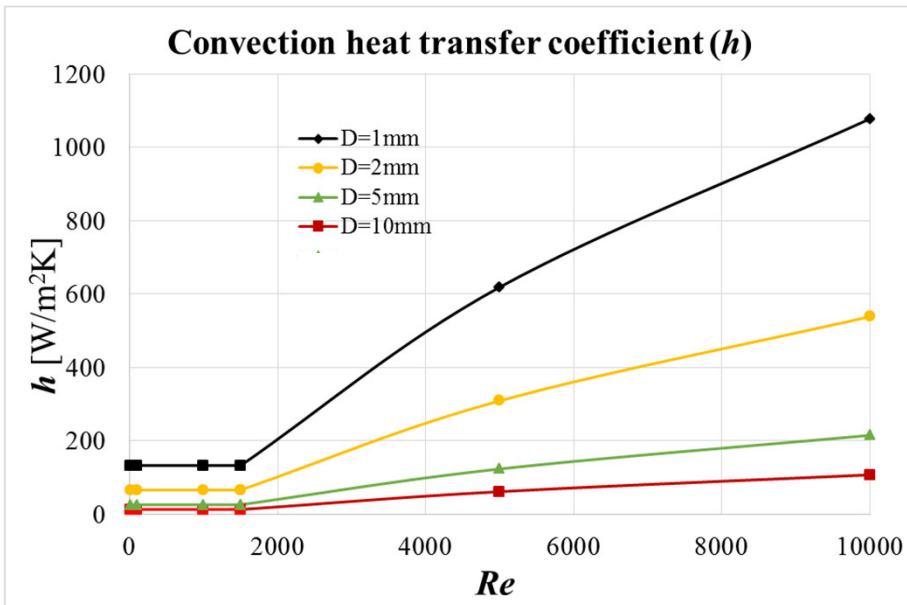
Dittus-Boelter



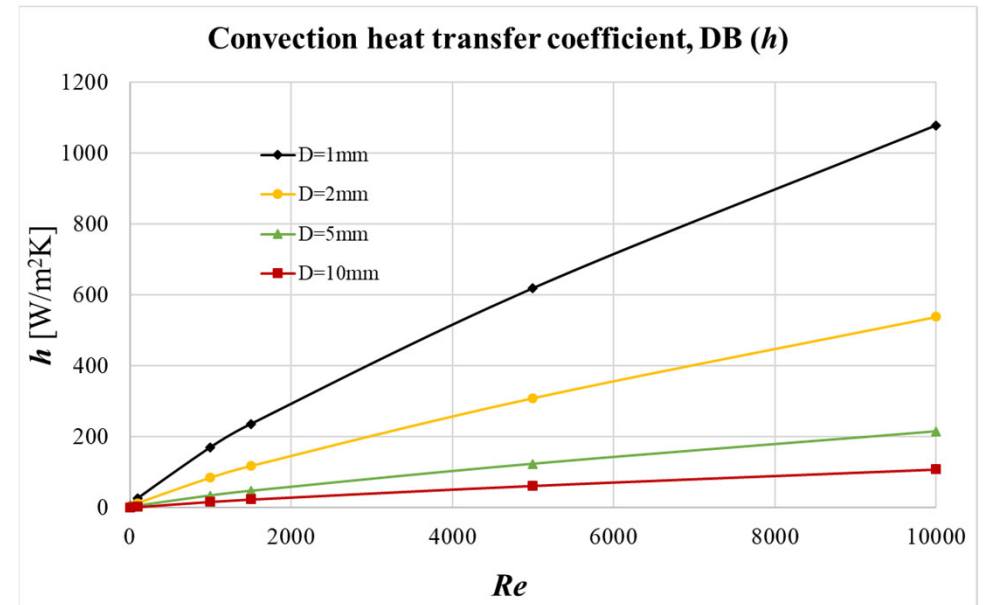
Dittus-Boelter correlation used for laminar and turbulent regime

➤ Dittus-Boelter under-predicts the Nusselt number in the laminar regime.

Results (Pipe Flow Analysis)



$Nu = 3.66$ used for the laminar regime (fully developed flow)

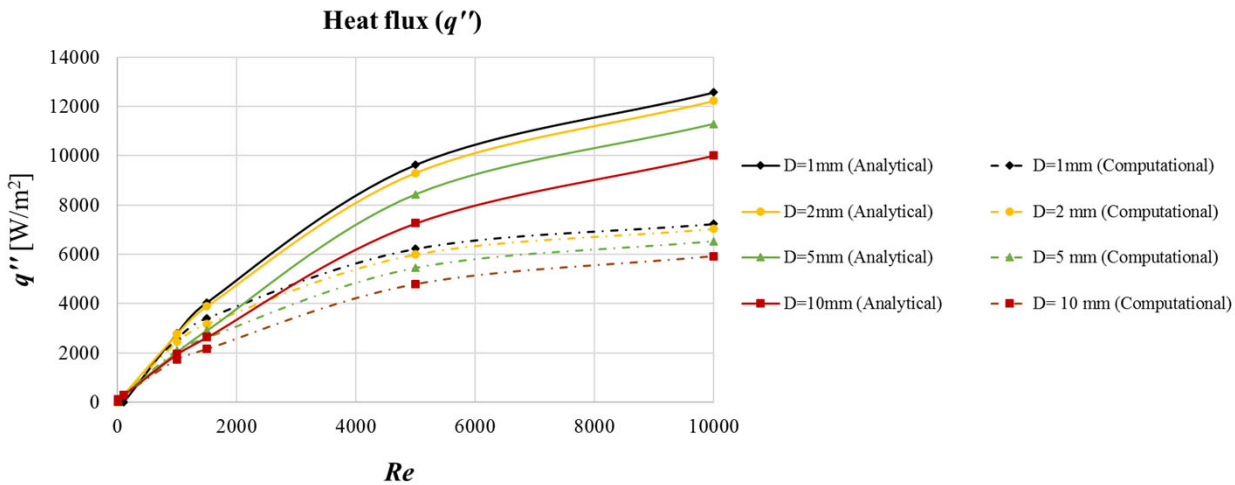


Dittus-Boelter correlation used for laminar and turbulent regime

$$Nu = \frac{hD}{k}; h = \frac{kNu}{D}$$

- The convection heat transfer coefficient, $h \propto Re$ (turbulent region)
- $h \propto \frac{1}{D}$

Results (Pipe Flow Analysis)

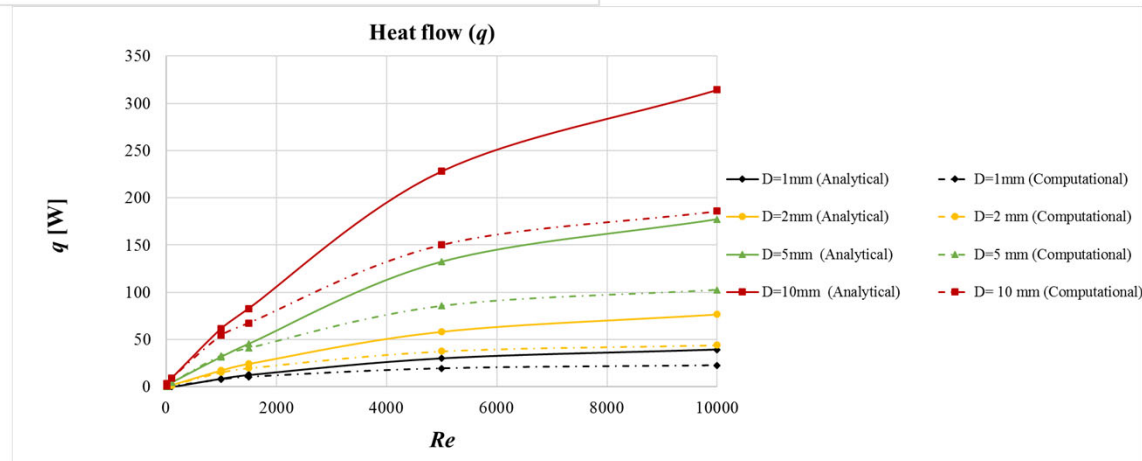


$$q = \bar{U} A_s \Delta T_{lm} = \dot{m} c_p \Delta T$$

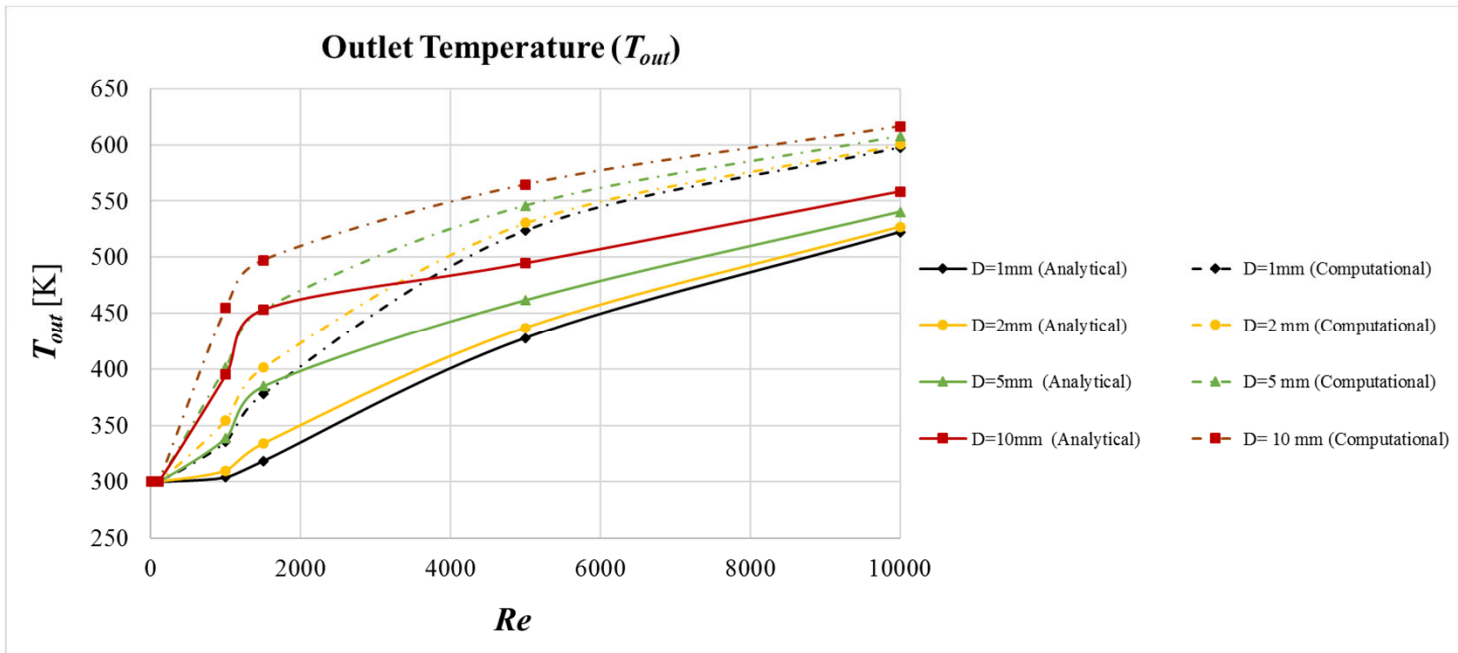
$$q'' = q / A_s$$

The heat flow, $q \propto D$
The heat flux, $q'' \propto \frac{1}{D}$

Analytical estimates exceed numerical estimates of q and q'' by a factor of ~ 2



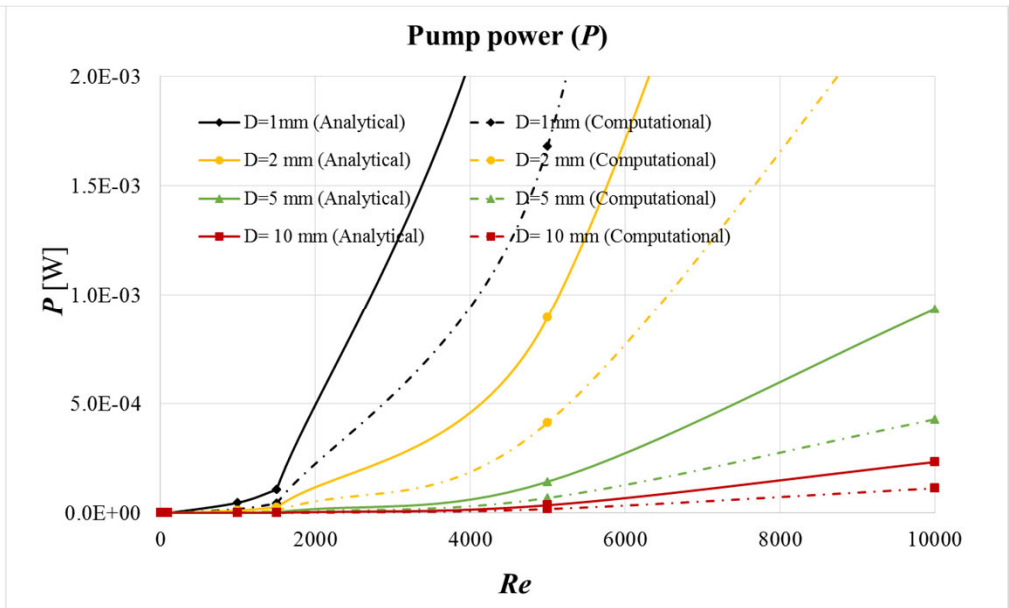
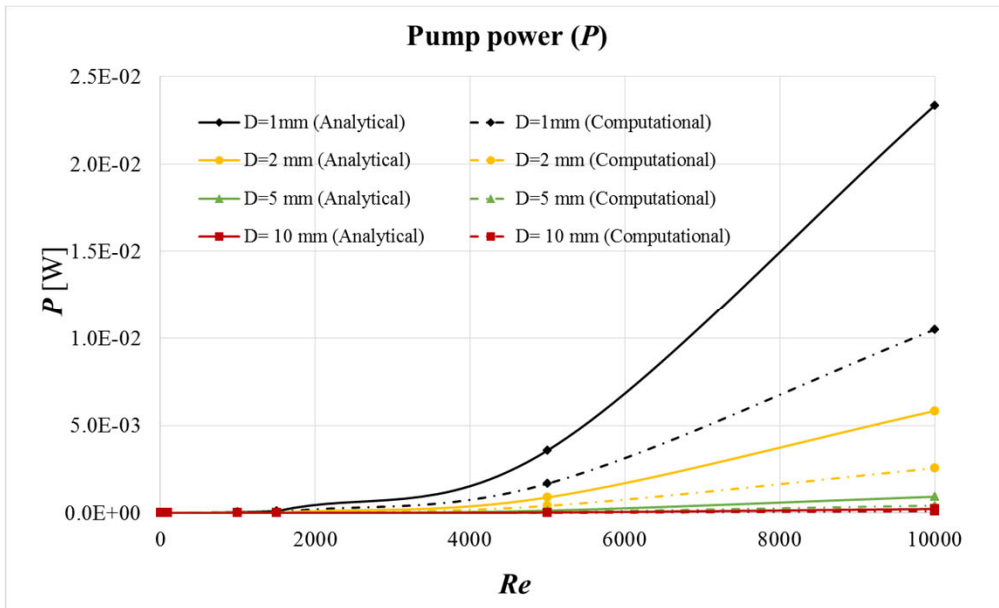
Results (Pipe Flow Analysis)



$$\frac{T_{\infty} - T_{outlet}}{T_{\infty} - T_{inlet}} = \exp\left(\frac{\bar{U}A_s}{\dot{m}c_p}\right)$$

$Re \uparrow = T_{out} \uparrow$
 $D \uparrow = T_{out} \uparrow$

Results (Pipe Flow Analysis)



Pump power, $P = Q\Delta P = A \cdot v \cdot \Delta P$

➤ $P \propto$ volume flow rate Q and pressure drop ΔP

➤ $P \propto \frac{1}{D}$

$$\Delta P = f \frac{L \rho v^2}{D} \frac{1}{2}$$

A = cross sectional area | Q = volume flow rate | ΔP = pressure drop | v = velocity

Analytical estimates exceed numerical estimates of power, P by a factor of ~ 2

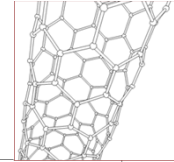
Conclusion

- The thermophysical properties of sCO₂ at P=8.12MPa and T=450-700[K] is consistent with established trends
- Large pressure drop requires greater pumping power. Tradeoffs in cost should be determined between larger and smaller tubes
- Geometric simplification error of 2D assumption resulted in less accurate numerical results

Acknowledgements

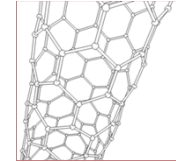
- Texas A&M High Performance Research Computing (HPRC)

Acknowledgements



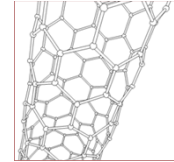
- **NSF (TTP/ SGER Program, SBIR Program):** Dr. Al Ortega, Dr. Pat Phelan, Dr. Sumanta Acharya, Dr. R-H Chen
- **AFRL/AFOSR: ASEE Summer Faculty Fellowship (RZ, RX):** Dr. R. Ponnappan, Dr. K. Yerkes, Dr. A. Roy, Dr. S. Ganguly
- **SPAWAR (Space and Naval Warfare Systems Command Center): ASEE Summer Faculty Fellowship:** Dr. Ryan Lu, Dr. Ajax Ramirez
- **ONR (Thermal Management Program):** Dr. Mark Spector
- **DARPA (MTO, MF³ Center):** Dr. Dennis Polla, Dr. Amit Lal, Dr. Abe Lee, Dr. Tayo Akinawande
- **NASA (URETI), JPL**
- **DOE (Solar Energy Program)**
- **ARPA-E (ARID):** U. Cincinnati, **UCLA**, **UC-Berkeley**, **Boeing**, **Evapco**, **Maulbetsch Consulting**
- **TSGC (Texas Space Grants Consortium)**
- **TEES (Texas A&M Engineering Experimentation Station)**
- **Crissman Institute for Petroleum Research**
- **Energy Institute**
- **Mary Kay O'Connor Process Safety Center**
- **Mechanical Engineering Dept., Texas A&M (New Faculty Start-Up Grant)**
- **QNRf (Qatar National Research Foundation)**
- **Industry Collaborators:**
 - **3M Corp.**
 - **ADA Technologies (ONR STTR Phase I)**
 - **Alstom**
 - **Aspen Thermal Systems (ONR SBIR Phase I)**
 - **Boeing**
 - **EVAPCO**
 - **Maulbetsch Consulting**
 - **ESI Corp.**
 - **General Dynamics (Anteon Corp.): AFRL Seed Grant**
 - **General Electric (GE): Corporate Research & Development, Global Research Center, NY**
 - **Irvine Sensors (AFOSR SBIR Phase II)**
 - **Lynntech Inc. (ARO SBIR Phase II)**
 - **Marathon Oil**
 - **MRV Systems (ONR SBIR Phase I)**
 - **NanoInk Inc.**
 - **Nano-MEMS Research (NSF SBIR Phase I, AFOSR SBIR/STTR Phase I & II)**
 - **Photonics Corp./ Trianja Inc. (Silicon Venture Partners/ BG Group)**

Acknowledgements



- **DOE Solar Energy Program (NREL, Golden, CO):**
 - **Brian Hunter, Allie Aman, Ryan Shining, Brad Ring, Greg Glatzmaier PhD, Craig Turchi PhD**
- **US Navy (SPAWAR, San Diego, CA):**
 - **R. Nguyen, C. Huynh, R. Lu PhD, A. Ramirez PhD**
- **AFRL (WPAFB, Dayton, OH):**
 - **R. Ponnappan PhD (AOARD), K. Yerkes PhD , T. Michalak, A. Flemming, S. Patnaik PhD, C. Obringer, L. Byrd PhD**
 - **A. Roy, PhD, S. Ganguly, PhD, L. Gschwender PhD, Ed Snyder PhD, R. Naik PhD, J. Slocik, L. Brott PhD**
- **NASA (JPL, CA):**
 - **Dr. Anu Kaul, Dr. K.G. Megerian**
- **Villanova University: A. Ortega PhD**
- **U. Maryland (UMD): J. Kim PhD**
- **U. New Haven: S. Sinha PhD**
- **U. Texas (Austin): S. Banerjee PhD**
- **U. Texas (Dallas): R. Baughman PhD**
- **U. Texas (Arlington): D. Shin PhD**
- **U. Cincinnati: R. Manglik PhD, M. Jog PhD**
- **UCLA: V. Dhir PhD**
- **UC-BERKELEY: Van P. Carey PhD**
- **TAMU – Qatar: R. Sadr PhD**

Acknowledgements



- DARPA (MF³ Center): 12 universities, 20 faculty, 8 companies, 2 National Labs.
 - George Whitesides (Harvard)
 - Luke Lee, Liwei Lin (UC Berkeley)
 - Juan Santiago (Stanford)
 - Marc Madou, Bill Tang, Abe Lee, Mark Bachman, Robert Corn, Jim Brody, Elliott Hui, M. Khine (UCI)
 - Steve Werely (Purdue)
 - Hugh Fan (University of Florida)
 - Jeff Wang (Johns Hopkins)
 - Don Devoe (U. of Maryland, College Park)
 - Ian Papautsky (U. Cincinnati)
 - Tianghong Cui (U. Minnesota)
 - David Beebe (U. Wisconsin)

Industry Partners (DARPA-MF³ Center)


- Douglas Scientific
- Pioneer Hi-Bred International, Inc.
- IDEX Health & Science
- SHRINK Nanotechnologies
- Microfluidic Innovations
- ESI Group
- Symbient Product Development
- Sierra Proto Express
- Lawrence Livermore National Labs.
- NASA Ames Research Center
- Beckman Coulter, Inc.
- BIOCUM



Past Industry Partners (DARPA-MF³ Center)


- Monsanto Company
- Invitrogen
- Applied Biosystems (Life Technologies)
- Irvine Sensors Corporation

Contact Information


 US010220410B2

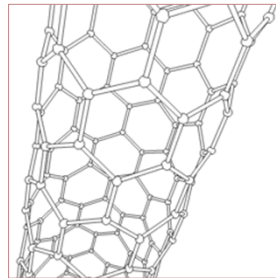
(12) **United States Patent**
Banerjee et al.
 (10) Patent No.: **US 10,220,410 B2**
 (45) Date of Patent: **Mar. 5, 2019**


(54) SYSTEMS AND METHODS FOR IN-SITU FORMATION OF NANOPARTICLES AND NANOFIBS
 (52) U.S. CL. CPC *B05D 5/02* (2013.01); *A61K 9/1273* (2013.01); *A61K 9/5115* (2013.01);


 US010215730B2

(12) **United States Patent**
Vann et al.
 (10) Patent No.: **US 10,215,730 B2**
 (45) Date of Patent: **Feb. 26, 2019**

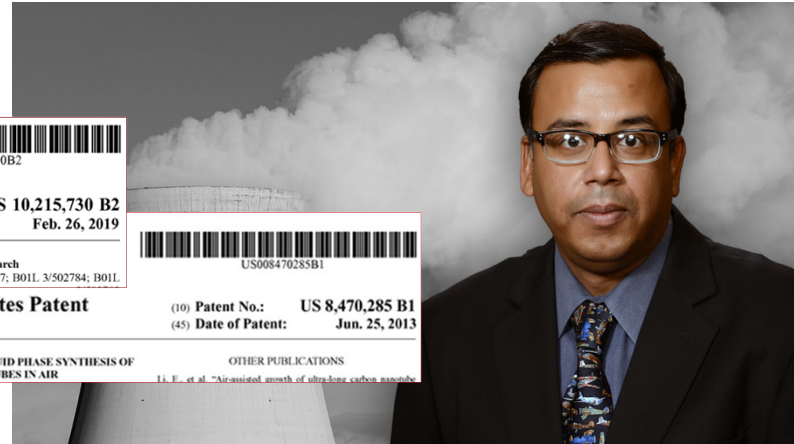
(54) ELECTROWETTING DISPENSING DEVICES AND RELATED METHODS
 (58) Field of Classification Search CPC B01L 2400/0427; B01L 3/502784; B01L



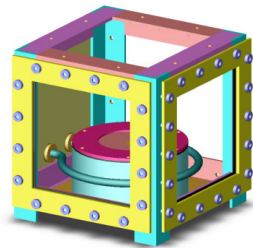
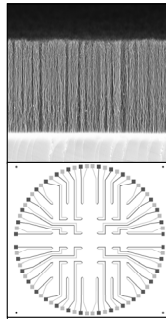

 US008470285B1

(12) **United States Patent**
Lu et al.
 (10) Patent No.: **US 8,470,285 B1**
 (45) Date of Patent: **Jun. 25, 2013**

(54) METHOD OF LIQUID PHASE SYNTHESIS OF CARBON NANOTUBES IN AIR
 OTHER PUBLICATIONS
 Li, F., et al. "Air-assisted growth of ultralong carbon nanotubes"

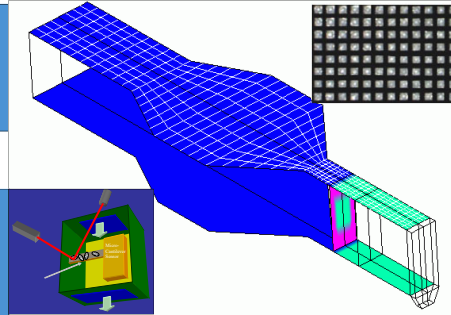


MS 3123 TAMU,
 College Station, TX.
 Ph: (979) 845-4500;
 Fax: (979) 845-3081



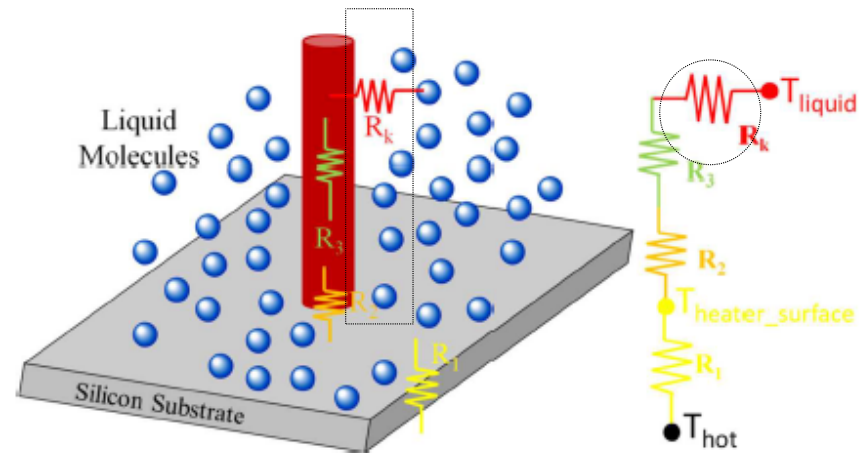
**Thermal-Fluids
 Technologies**

**Emerging
 Technologies**



Interfacial Thermal Resistance

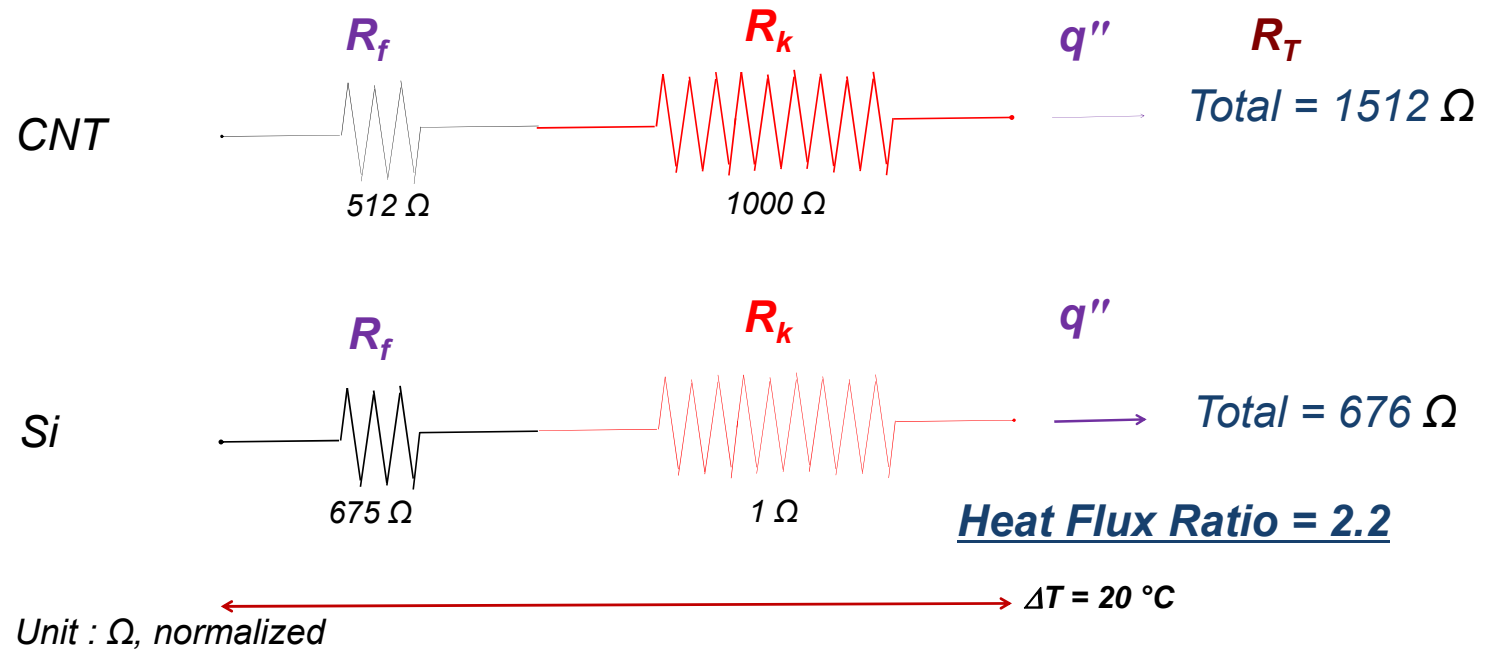
Interfacial thermal resistance :between nanofin and liquids¹⁾



1) N. Singh, & D. Banerjee, *Nanofins: Science and applications*, Springer, 2013

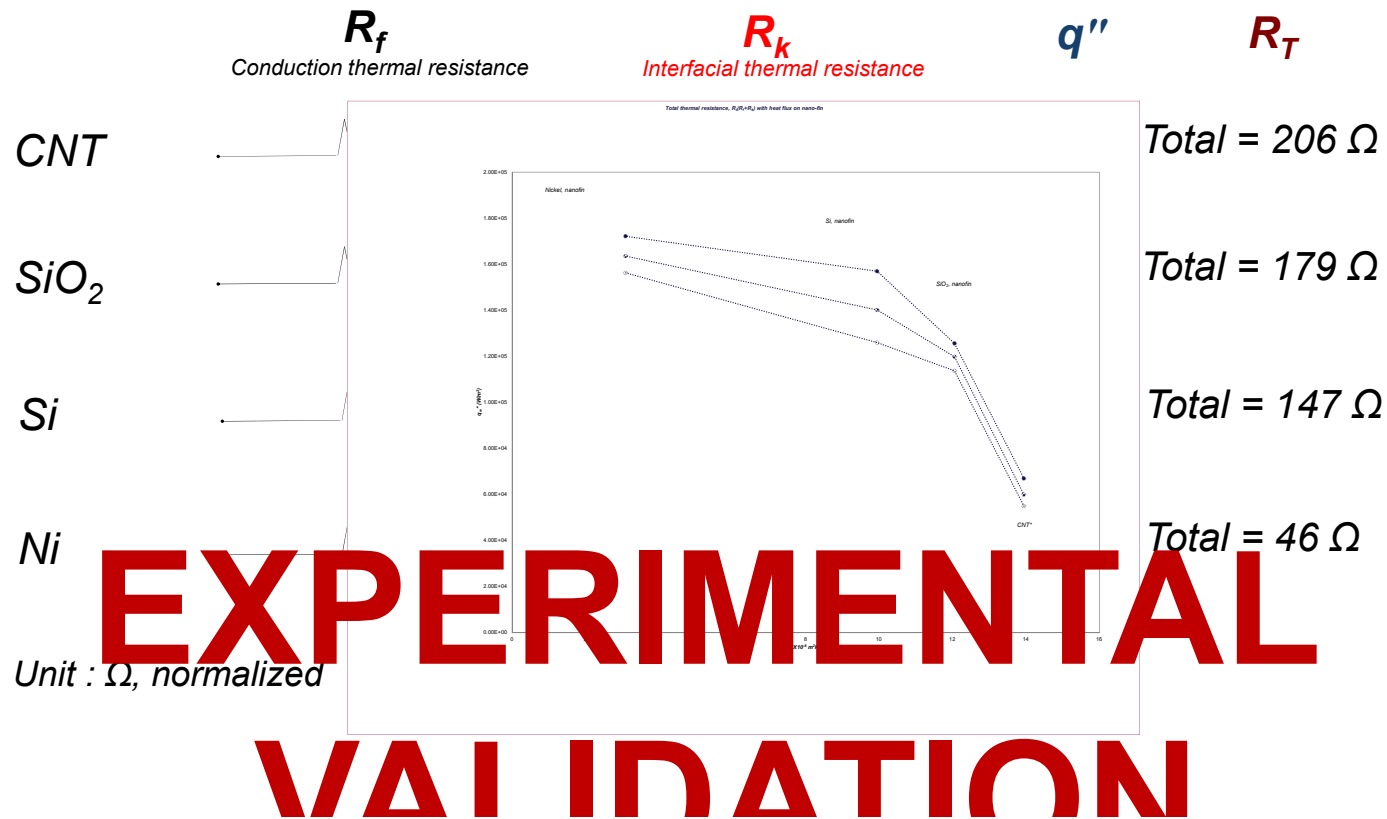
nanoFin Effect (nFE)

Interfacial thermal resistance (R_f) is critical for heat transfer between solid and liquid.



nanoFin Effect (nFE)

Interfacial thermal resistance (R_i) is critical for heat transfer between solid and liquid.



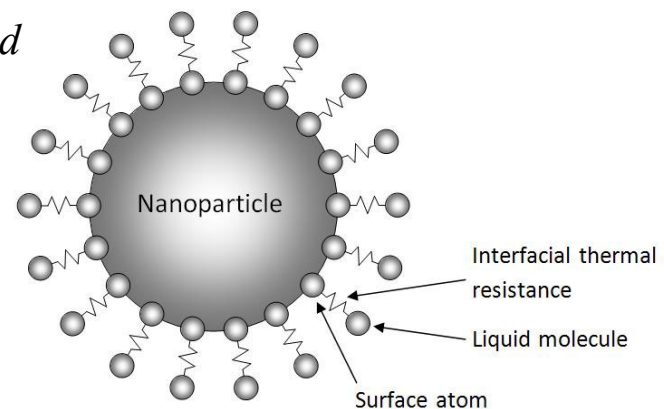
Mechanisms

- Three Modes:
 - Mode I: Higher specific heat of nano-particles
 - Mode II: Particle-Fluid Interfacial Interactions
 - Mode III: “Semi-Solid” Layering of the Liquid Phase on the nano-particle

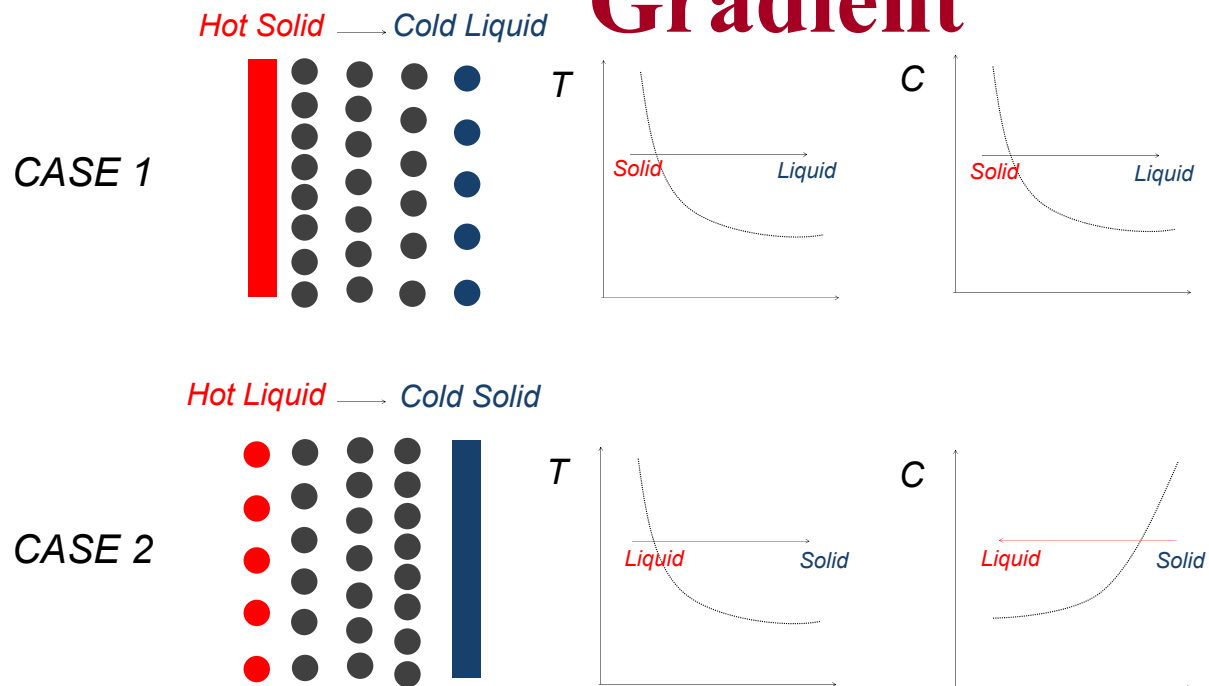
~~$$C_{p,TOTAL} = x \cdot C_{p,particle} + (1-x) \cdot C_{p,fluid}$$~~

$$C_{p,TOTAL} = x \cdot C_{p,particle} + y \cdot C_{p,compressed-phase} + (1-x-y) \cdot C_{p,fluid}$$

- Shin and Banerjee, 2012, ASME. J. Heat Tr. (in press)
- Shin and Banerjee, 2011, Int. J. Heat & Mass Tr.
- Shin and Banerjee, 2011, ASME. J. Heat Tr.
- Shin and Banerjee, 2010, Int. J. of Str. Chg. In Solids
- Shin, D., 2011, PHD Thesis, Texas A&M
- Jung, S., 2012, PHD Thesis, Texas A&M
- Jo, B., 2012, PHD Thesis Texas A&M



Temperature & Concentration Gradient



Temperature and concentration gradient has same direction at CASE1, but it has opposite direction at CASE2, hence this drive the heat flux bias .

Thermal Diode

1) Heat flux by temperature and concentration gradient

$$q'' = \sum h_i J_i - K \nabla T = \bar{h} J_x - K \nabla T \quad \text{and} \quad J_x = -\rho D \nabla C - \rho D_T C_o (1 - C_o) \nabla T \quad \longrightarrow \quad q'' = -\rho D \bar{h} \nabla C - [\rho \bar{h} D_T C_o (1 - C_o) + K] \nabla T$$

2) Define dimensionless parameter

$$q_o'' = K \frac{(T_s - T_\infty)}{L_c} \quad Nu = \frac{h L_c}{K} = \frac{(q'' / \Delta T) L_c}{K} = \frac{q''}{q_o''} \quad q'' = q_o'' Nu \quad \theta = \frac{T - T_\infty}{T_s - T_\infty} \quad C^* = \frac{C - C_\infty}{C_s - C_\infty} \quad \nabla C = \nabla C^* \frac{(C_s - C_\infty)}{L_c} \quad \nabla T = \nabla \theta \frac{(T_s - T_\infty)}{L_c}$$

3) Dimensionless analysis

$$K \frac{(T_s - T_\infty)}{L_c} Nu = -\rho \bar{h} D \frac{(C_s - C_\infty)}{L_c} \nabla C^* - [\rho \bar{h} D_T C_o (1 - C_o) + K] \frac{(T_s - T_\infty)}{L_c} \nabla \theta \quad \longrightarrow \quad Nu = -\frac{\rho \bar{h} D (C_s - C_\infty)}{K (T_s - T_\infty)} \nabla C^* - \left[\frac{\rho \bar{h} D_T}{K} C_o (1 - C_o) + 1 \right] \nabla \theta$$

4) Nu. Number for diode effect

CASE1(Nu._f): **Hot solid** → **Cold liquid**

$$Nu_f = -\frac{\rho \bar{h} D (C_s - C_\infty)}{K (T_s - T_\infty)} \nabla C^* - \left[\frac{\rho \bar{h} D_T}{K} C_o (1 - C_o) + 1 \right] \nabla \theta$$

CASE2(Nu._b): **Hot liquid** → **Cold solid**

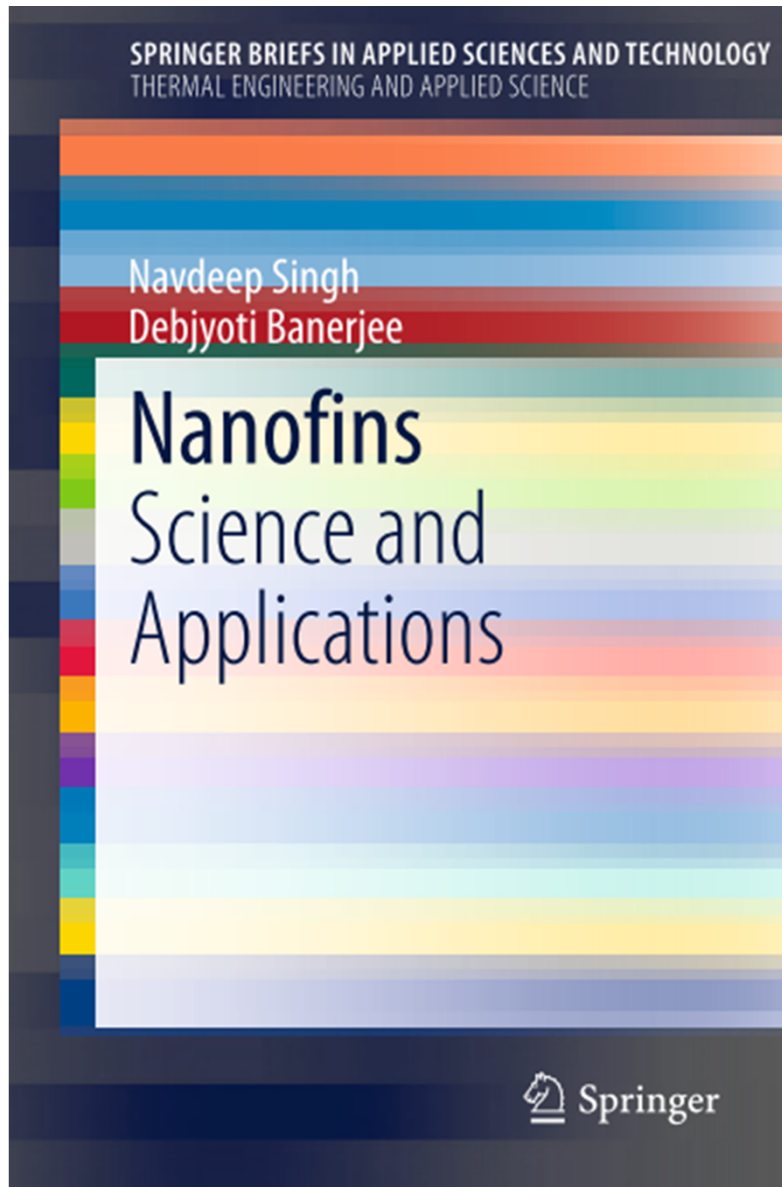
$$Nu_b = \frac{\rho \bar{h} D (C_s - C_\infty)}{K (T_s - T_\infty)} \nabla C^* - \left[\frac{\rho \bar{h} D_T}{K} C_o (1 - C_o) + 1 \right] \nabla \theta$$

$$Nu_D = Nu_f - Nu_b$$

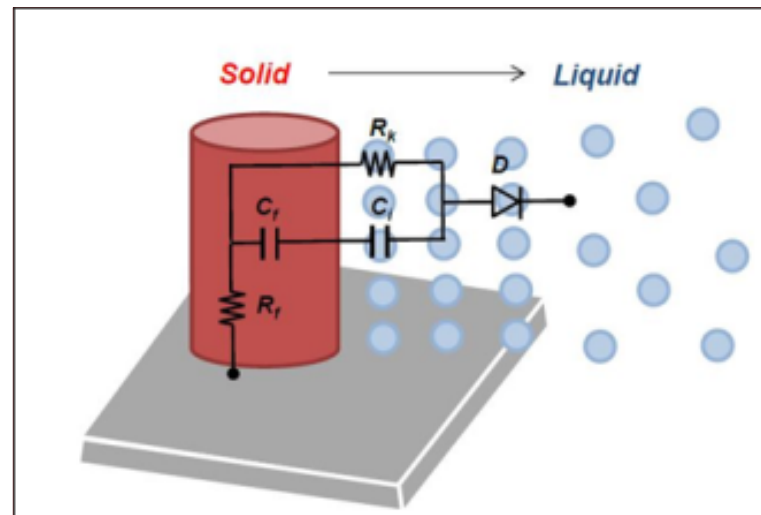
$$Nu_D = Nu_f - Nu_b = -2 \frac{\rho \bar{h} D (C_s - C_\infty)}{K (T_s - T_\infty)} \nabla C^*$$

→ **Same direction**

→ **Opposite direction**



“*nanoFin* Effect”



Corrosion reduced by 50~100%

Work in collaboration with Dr. Diana Berman:
Lee, J. *, Kuchibhotla, A. *, Banerjee, D., and Berman, D.,
“Silica Nanoparticles as copper corrosion inhibitors”,
Materials Research Express, May, 2019.

<https://doi.org/10.1088/2053-1591/ab2270>