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#### The effect of elevated temperature and pressure on five metallic materials in a quasi-static supercritical-CO<sub>2</sub> environment

sCO<sub>2</sub> Power Cycle Symposium

San Antonio, TX, February 21 – February 24, 2022

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# R&D Activities on sCO<sub>2</sub> Power Cycles at CE-O

**CE-O's R&D on sCO<sub>2</sub> cycles:** initiated in 2006 as a cross-cutting advanced power conversion technology for application to zero- or lowemission nuclear, fossil, and renewable energy sources.

#### CanmetENERGY's G2 Technology:

- Crown patented sCO<sub>2</sub> power cycle technology
- A unique indirectly oxy-fired sCO<sub>2</sub> power cycle (producing electricity, water/steam, and pressurized pipeline-ready  $CO_2$ )
- R&D at pilot scale for de-risking the technology and its main components

In-kind technical support to "STEP 10 MW<sub>e</sub> sCO<sub>2</sub> Pilot Plant Test Facility":

- The project is led by Gas Technology Institute (GTI) and its partners (SwRI, GE-R, DOE-NETL)
- Goal: design, construct, commission, and operate a 10 MW<sub>e</sub> sCO<sub>2</sub> pilot plant test facility in Texas, USA
- CE-O's partners: Carleton University, NRC, CMAT



G2 Technology process flow

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# CE-O/Carleton University sCO<sub>2</sub> Corrosion Test Rig



Used to generate new data and knowledge on alloys corrosion in sCO<sub>2</sub> environment

sCO<sub>2</sub> Corrosion Test Rig

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# Round Robin Test Campaign

- Led by EPRI; involving 8 research institutions
- CE-O participated through Carleton University, with support from NRC for characterization
- 5 Alloys tested:
  - Inconel 740H, Inconel 625, Haynes HR-120, Stainless Steel 316L, and GR 91 steel
  - Alloy samples exposed to  $sCO_2 \otimes 550^{\circ}C$  and  $700^{\circ}C$ , 200 bar for 1500 hrs (in 500-hr intervals)
- Mass gain measured at each 500-hr interval
- Samples characterized using SEM, EDX, and XRD after each 500-hr interval



Alloy samples loaded into specimen boat

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# Visual Appearance (550°C)

- After 500 hours samples maintained polished appearance
  - 625 displayed slight heat-induced discoloration on both faces
- 1000 hours onward all samples darkened due to thickening of the oxide layer



#### Appearance of alloys during the 1500 hours

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### Mass Gain and Results Comparison (550°C)

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600

800

Time (h)

1000 1200 1400 1600

·----

400

200

0.00

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0



# Surface SEM Analysis (550°C)

- All alloys developed a thin layer of oxide after 500 hour exposure
- Oversized oxide clusters identified on the surface of GR 91 and 316L steels
  - Clusters grew and combined throughout the \_\_\_\_ exposure
- Some oversized oxide clusters develop on the surface of 625 and HR-120 at the later exposures



#### Surface SEM of the samples during the 1500 hours

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# EDX and XRD Analysis (550°C)

- Scans of 625 and HR-120 exhibit a Cr-based primary oxide layer  $(Cr_2O_3)$ 
  - XRD scans of 625 and HR-120 indicate low oxidation
  - 625 developed Mn-Ni-Nb spinels
  - HR-120 showed presence of Mn-Fe-Cr spinel structures
- Scans of 316L and GR91 steels indicate formation of Fe-rich primary oxide ( $Fe_3O_4$ )
  - GR91 showed significant presence of  $Fe_3O_4$ which is indicative of poor oxidation resistance



XRD scans of alloys 625 and HR-120 after 1500 hours of exposure



XRD scans of steels 316L and GR91 after 1500 hours of exposure

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### Cross-sectional SEM Analysis (550°C)





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# Visual Appearance (700°C)

- After 500 hour exposure all samples darkened, indicating oxidation
- By the end of 1500 hours:
  - 625 appeared least discoloured
  - 316L appeared most discoloured
  - All coupons exhibited abrasion caused by flow pattern through the specimen boat



#### Appearance of alloys during the 1500 hours

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### Mass Gain and Results Comparison (700°C)







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# Surface SEM Analysis (700°C)

- Substantial layer of oxide was detected on the surface of each alloy
- 316L: uniform base layer with rows of oversized oxide
- HR-120 and 625: uniform layer of oxide after 500 and 1000 hours onward respectively
- 740H: thin base layer with small grains forming after 1000 hours



#### Surface SEM of the samples during the 1500 hours

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# EDX and XRD Analysis (700°C)

- EDX scans of 740H, 625 and HR-120 exhibit a Cr-based primary oxide layer  $(Cr_2O_3)$ 
  - 740H displayed an increased content of Nb within the greater oxide regions and AI content near the base of oxide layer
  - 625 exhibited an Nb rich bottom oxide layer
  - Scans for HR-120 indicated that the Cr-rich base oxides also possess a high content of Mn spinels
- 316L surface oxide consisted of Cr prominent base layer with a Fe-based oversized growths
  - Oxide layer is combination of  $Cr_2O_3$  and  $Fe_3O_4$



XRD scans of alloys 740H and 625 after 1500 hours of exposure



XRD scans of alloys HR-120 and 316L after 1500 hours of exposure

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### Cross-sectional SEM Analysis (700°C)



# Cross-sectional SEM Analysis (700°C)

- 740H displayed a uniform oxide layer composed primarily of  $Cr_2O_3$ 
  - Internal Al<sub>2</sub>O<sub>3</sub> oxide layer was detected which forms a barrier to prevent oxygen diffusion
  - Many voids along grain boundaries were found under the oxide layer
- 625 exhibited the thinnest oxide layer composed mainly of  $Cr_2O_3$
- 316L displayed Cr<sub>2</sub>O<sub>3</sub> base layer and regions of Fe<sub>3</sub>O<sub>4</sub> excessive growth penetrating into substrate through Cr oxide layer
  - This is indicative of low life expectancy of the metal
- HR-120 displayed a discontinuous Cr oxide base layer and a more uniform upper layer
  - Significant oxide penetration was detected and a thick Cr-depletion zone



#### Cross-sectional SEM of the four alloys during the 1500 hours

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## Oxides Thickness Comparison (550°C vs 700°C)



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#### Remarks

- Inconel 625 exhibited the best performance at both temperatures
  - 625 had among lowest mass gains and developed the thinnest and most stable oxide layer
- Longer testing needs to be done to determine maximum possible oxide thickness for these materials
  - Long term exposure testing (4,800 hours) at 700°C/200 bar has been completed for 740H, 625, HR-120 & 316L
- Further testing needs to be performed using  $sCO_2$  in the presence of impurities, in particular O<sub>2</sub> and H<sub>2</sub>O
  - New sCO<sub>2</sub> corrosion test facility with impurities will be commissioned by the end of this year at CE-O

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