

Defining Temperature Limitations for Steels in $s\text{CO}_2$ Applications

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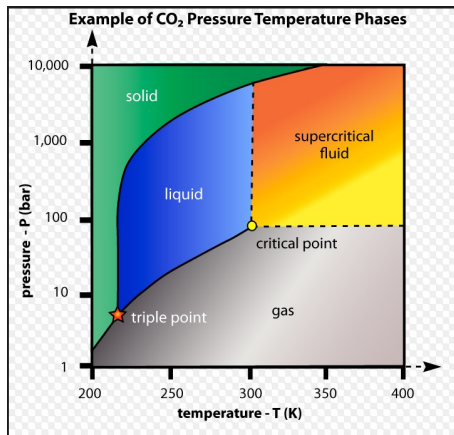
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Acknowledgments

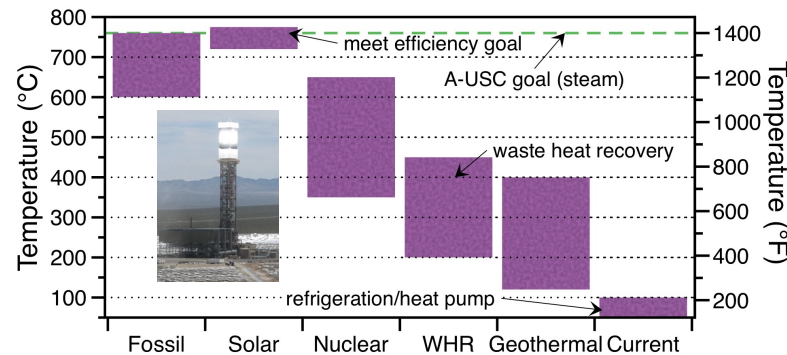
- Funding: DOE Office of Fossil Energy, Crosscutting Research Program
 - S. Nathan, NETL Project monitor
- M. Howell (retired), B. Johnston — oxidation experiments
- T. Lowe — SEM, image analysis
- V. Cox — metallography
- Special thanks for alloys:
 - Tenaris (T91)
 - EPRI (VM12)
 - Sam Sham, INL (709 NE code case material)

sCO₂ cycles are of wide interest: goal is to commercialize!

sCO₂ has many unique & attractive aspects

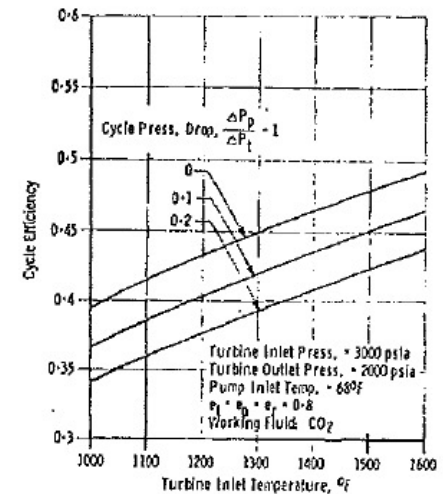


Supercritical CO₂ applications



Interest in >700°C for high efficiency

Cost is a concern:
Ni-based alloys: OK in sCO₂
Where can steels be used?



Feher, 1965

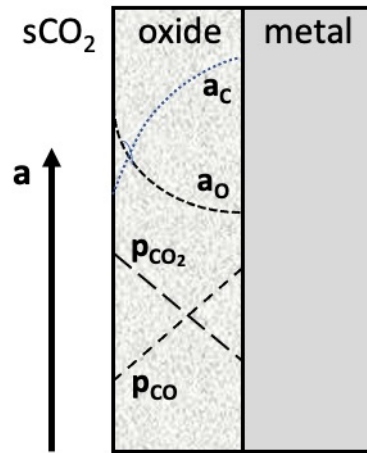
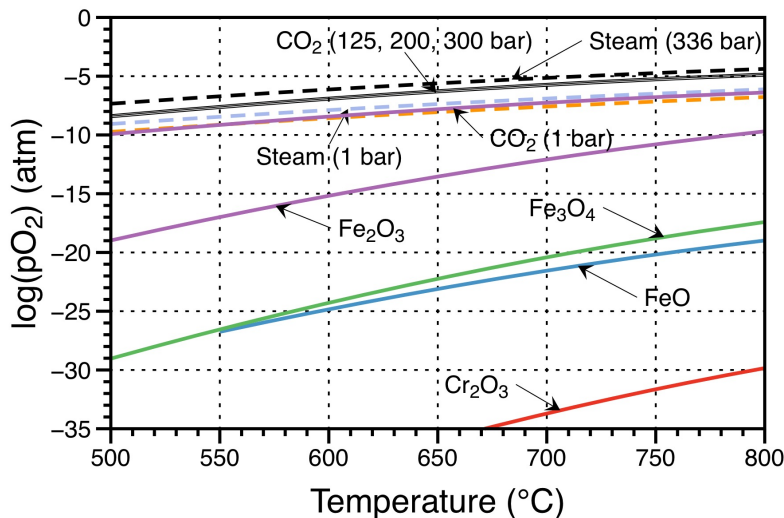
50% sCO₂ eff @ >720°C

- Low critical point (31°C/7.4 MPa)
- High, liquid-like density
- Flexible, small turbomachinery

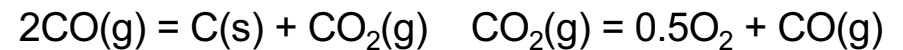
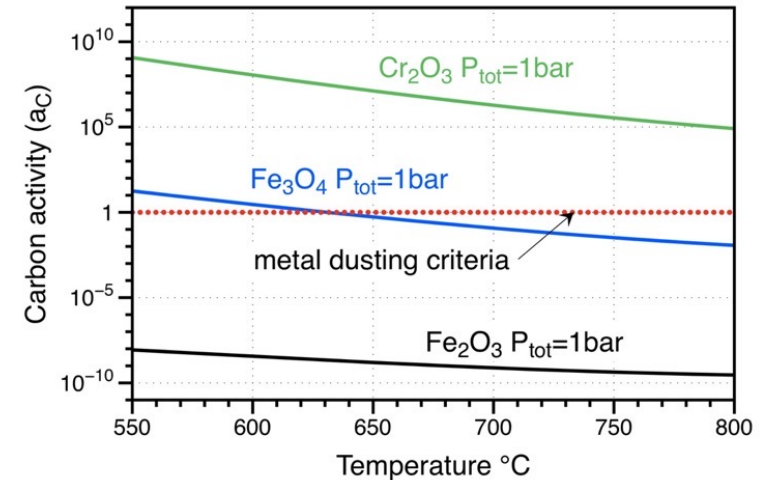
Thermodynamics: Oxygen levels similar in steam/CO₂

Concern about high C activity at m-o interface

Factsage calculations



Young et al., 2011



High carbon activity at P_{total} = 1 bar (What is P_{interface}?)

General conclusion: internal carburization concern for Fe-based alloys

sCO₂ compatibility focused on steels at 450°-650°C in RG CO₂

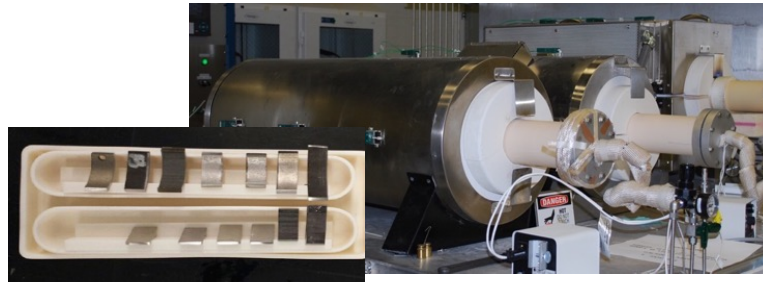
Autoclave: 300 bar sCO₂
500-h cycles



Correct temperature and pressure

4-5 cm² alloy coupons

Tube furnace: 1 bar CO₂
500-h cycles

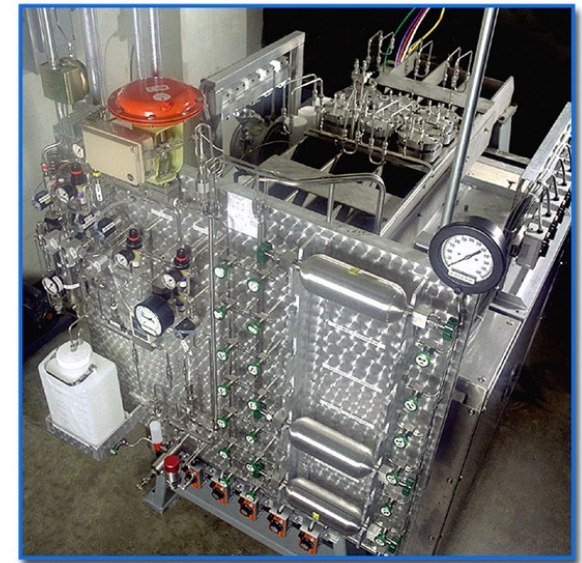


Same cycle frequency as autoclave



Box furnace:
Lab. Air
500-h cycles
(baseline)

"Keiser" rig:
500-h cycles, 1-43 bar CO₂



Studies at 800°-1200°C

Baseline of research grade (RG) CO₂: ≤ 5 ppm H₂O and ≤ 5 ppm O₂
industrial grade (IG) CO₂: 18±16 ppm H₂O and ≤ 32 ppm O₂

ORNL steel project started in August 2019

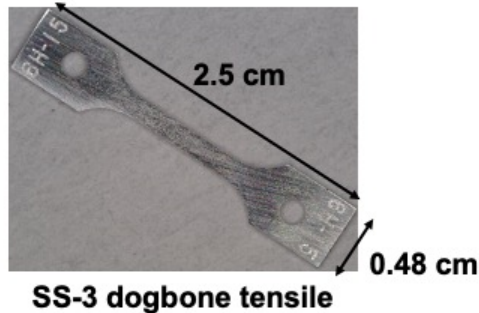
Test matrix & progress

Temperature	RG sCO ₂	+1%O ₂ +0.1%H ₂ O
450°C (842°F)	2,000 h	1000 h
550°C (1022°F)	2,000 h	1000 h
650°C (1202°F)	1000 h	1000 h

Not for today



Autoclave: 300 bar sCO₂
500-h cycles



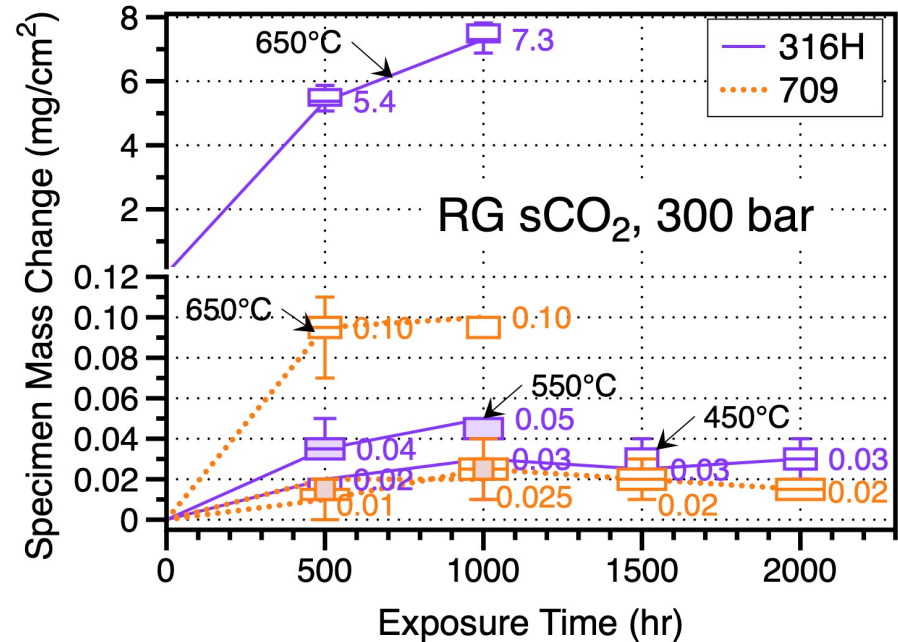
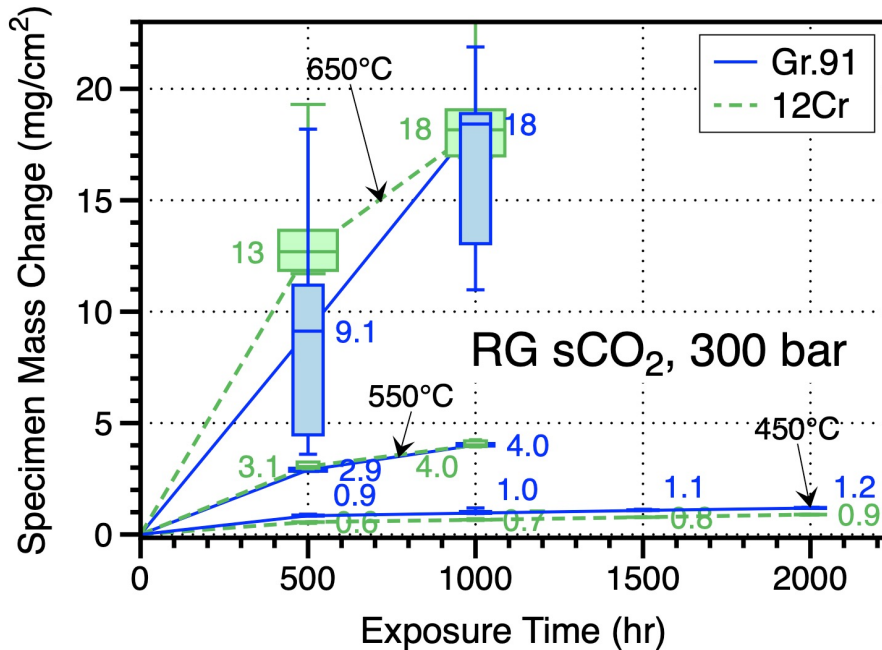
~5 cm² alloy coupons + tensile specimens

- Four primary alloys in test matrix
 - T91 (9Cr-1Mo)
 - VM12 (~11Cr)
 - 316H (conventional stainless steel)
 - NF709 (advanced austenitic, 20Cr-25Ni+Nb)
- 10 specimens of each alloy
- With and without impurities (open vs. closed)

Alloy	UNS	Cr	Ni	Mn	Si	C	N	Other
Gr.91	K90901	8.6	0.3	0.5	0.4	.10	.05	0.9Mo,0.2V
VM12	12CrCoW	11.5	0.4	0.4	0.4	.12	.04	1.6W,1.5Co
316H	S31609	16.3	10.0	0.8	0.5	.04	.04	2.0Mo,0.3Co
NF709	S31025	20.1	25.2	0.9	0.4	.06	.15	1.5Mo,0.2Nb

Baseline of research grade (RG) CO₂: ≤ 5 ppm H₂O and ≤ 5 ppm O₂

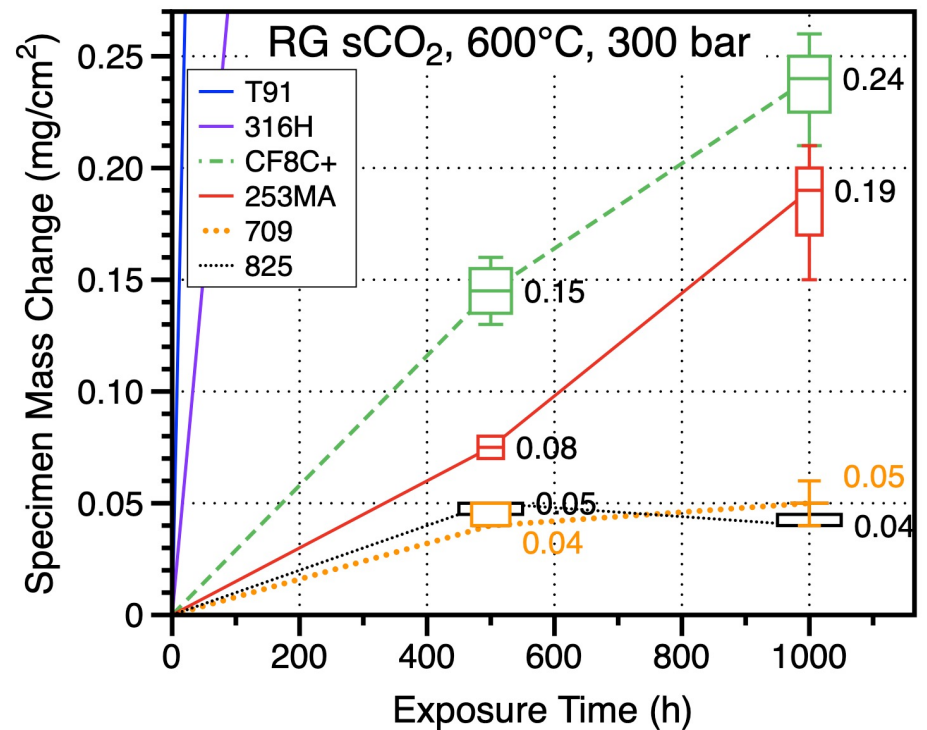
Mass change of 5-6 specimens in RG sCO₂ plotted



- One specimen of each alloy removed at 500 h for metallography
 - High mass gains for 9-12%Cr steels in all cases
 - Low mass gains for FCC steels except 316H at 650°C
- Ran 450°C experiment to 2,000 h for improved assessment

sCO₂ 600°C: 2021 explored some additional candidates

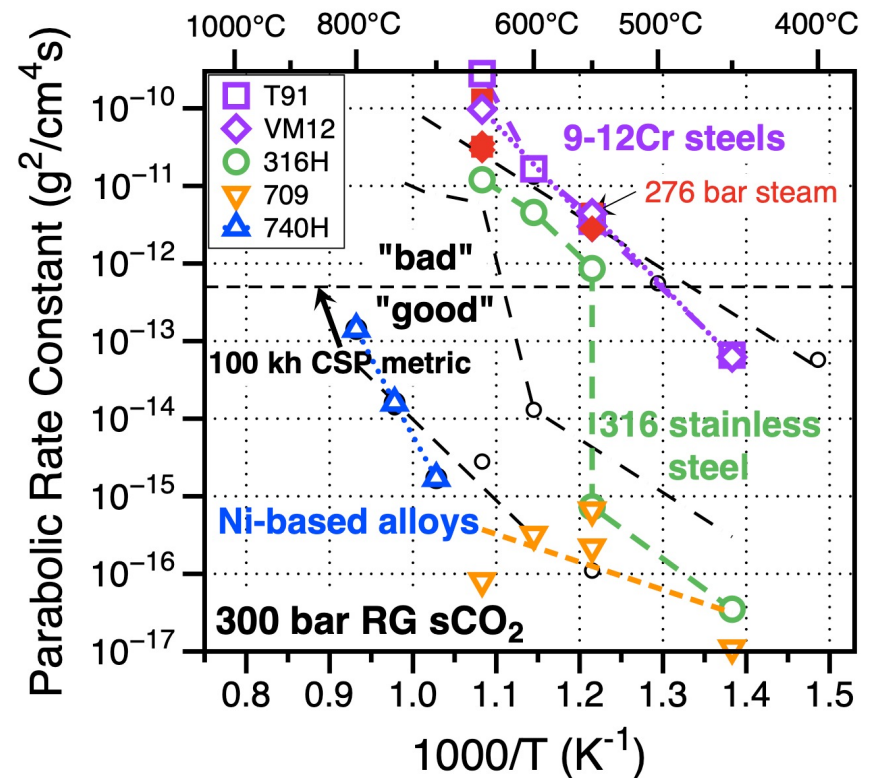
- 550°-600°C critical temperature
 - 30 MPa, RG CO₂
 - 4-6 specimens of each alloy
- Ni-based alloy 825 (Ni-31Fe-23Cr)
 - Small mass change (as expected)
 - Similar to advanced austenitic 709 (Fe-20Cr-25Ni)
- 253MA: higher Cr, Si + Ce/La
 - accelerating mass gain = FeO_x nodules...
- CF8C-Plus (cast 347): high Mn
 - Higher strength version of CF8C
 - Also accelerating mass gain



- **825**: 40Ni-31Fe-23Cr-3Mo-1.7Cu-1Ti-0.5Mn-0.3Si
- **CF8C-Plus**: Fe-12.5Ni-19.5Cr-4Mn-0.7Nb-0.7Si-0.5Cu-0.3Mo-0.25N (ASTM spec.)
- **253MA**: 65Fe-11Ni-21Cr-1.5Si-0.7Mn-0.3Mo-0.15N-0.03Ce-0.01La

Measured rates in sCO₂ consistent with the literature

- Metric developed for Solar CSP
 - Slow rate = OK for 100kh life
- Ni-based alloys all “good”
 - Lifetime model: $\leq 800^{\circ}\text{C} = 100\text{kh}$
- Steel limitations
 - Ferritic-martensitic alloys $< 500^{\circ}\text{C}$
 - Austenitic alloys $< 600^{\circ}\text{C}$
 - Obvious jump in kinetics
 - Advanced austenitics, better
 - Value in 20-25%Cr, 20-25%Ni
 - Low values hard to measure



9-12Cr steels have similar rates in 276 bar steam

Scales on 9-12Cr steels after 1,000 h exposures at each temperature

- Fe-rich oxide in each case
 - Classic two-layer structure
 - Fe_3O_4 with low $p\text{O}_2$ in $s\text{CO}_2$
 - Internal oxidation zone
- Little benefit of higher Cr content in VM12 under these conditions

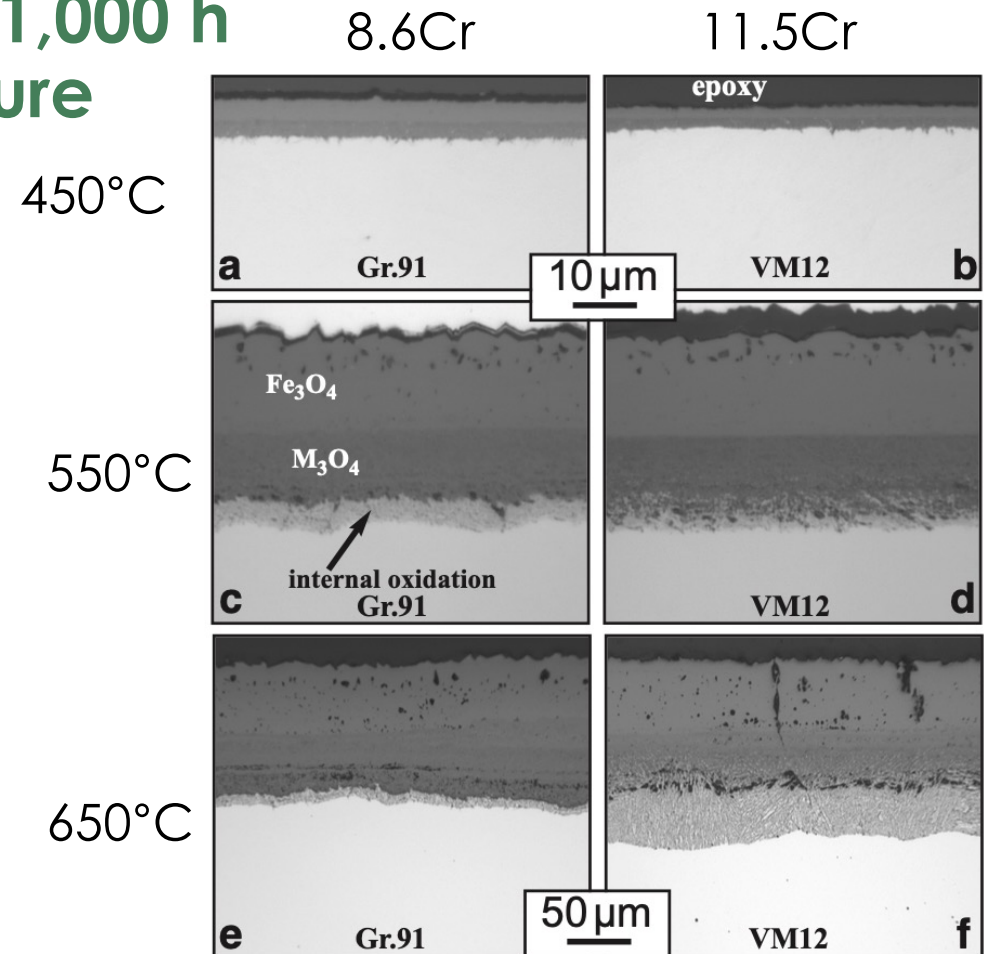


Figure 3. Light microscopy of polished cross-sections of (a,c,e) Gr.91 and (b,d,f) VM12 after 1000 h in RG $s\text{CO}_2$ at (a,b) 450°C, (c,d) 550°C and (e,f) 650°C.

Light microscopy

Scales on stainless steels after 1,000 h exposures at each temperature

- Cr-rich oxides in 5 of 6 cases
 - Thin, protective: difficult to image using light microscopy
- Fe-rich oxide for 316H at 650°C
 - Two-layer structure, but only inner layer was retained
- Strong benefit of increasing the Cr/Ni content in alloy 709
 - You get what you pay for!

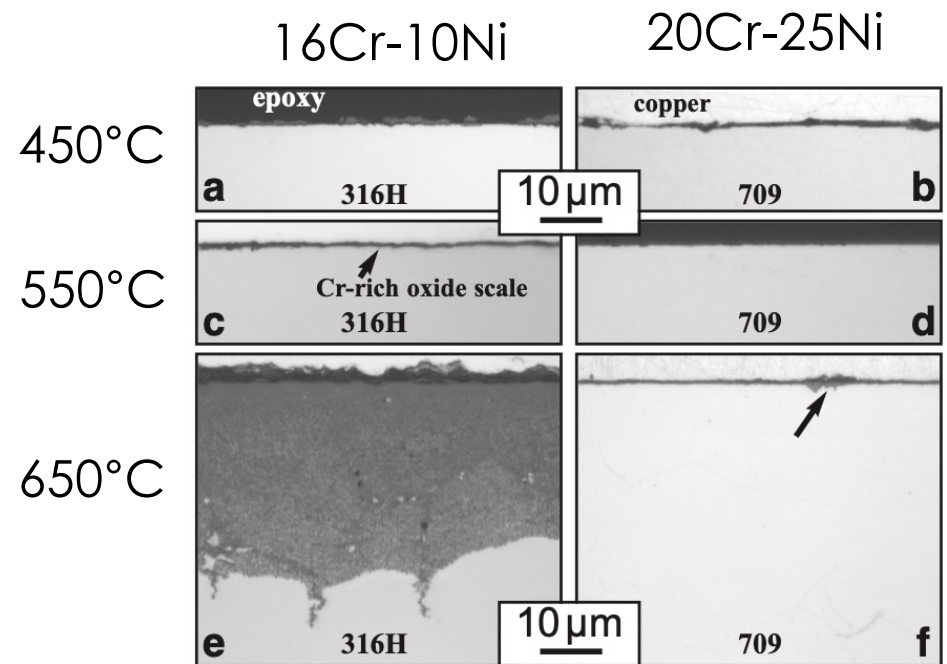


Figure 4. Light microscopy of polished cross-sections of (a,c,e) 316H and (b,d,f) 709 after 1000 h in RG sCO₂ at (a,b) 450°C, (c,d) 550°C and (e,f) 650°C.

Light microscopy

Scales on stainless steels after 1,000 h exposures at each temperature

- Cr-rich oxides in 5 of 6 cases
 - Thin, protective
 - Cr depletion beneath scale
 - Nodules primarily Cr-rich
- Fe-rich oxide for 316H at 650°C
 - Two-layer structure, but only inner layer was retained
- Strong benefit of increasing the Cr/Ni content in alloy 709

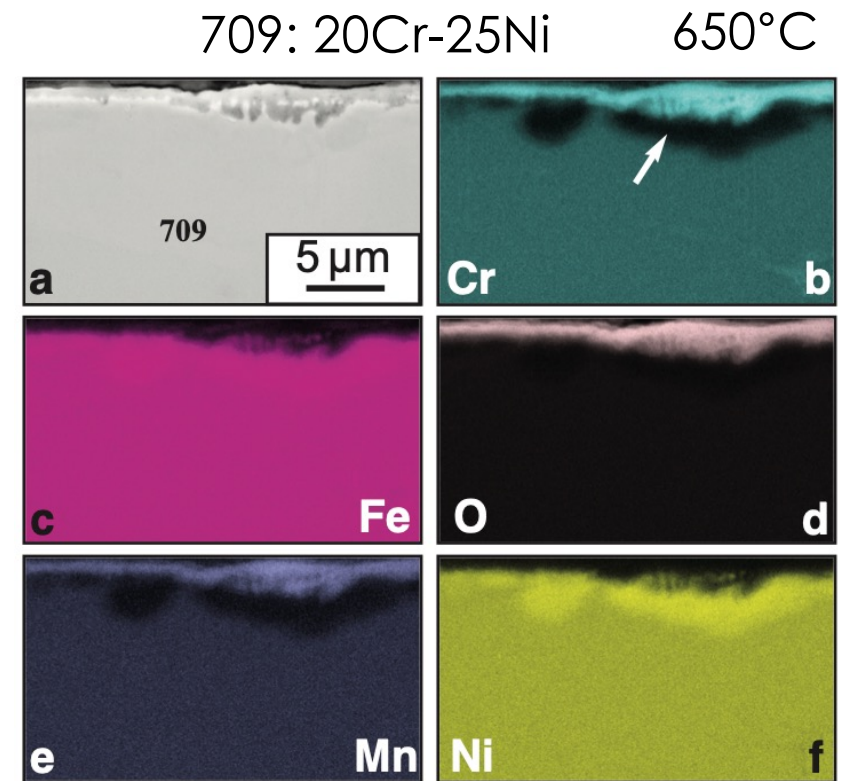


Figure 7. (a) SEM secondary electron image of 709 after 1000 h at 650°C in RG sCO₂ and (b-f) EDX maps of the same region.

Scanning electron microscopy/EDX maps

Scales on stainless steels after 1,000 h exposures at each temperature

- Cr-rich oxides in 5 of 6 cases
 - Thin, protective
- Fe-rich oxide for 316H at 650°C
 - Two-layer structure, but only inner layer was retained
 - Cr-rich oxide at metal-oxide interface with Cr depletion
 - Perhaps Ni metal incorporated
- Strong benefit of increasing the Cr/Ni content in alloy 709

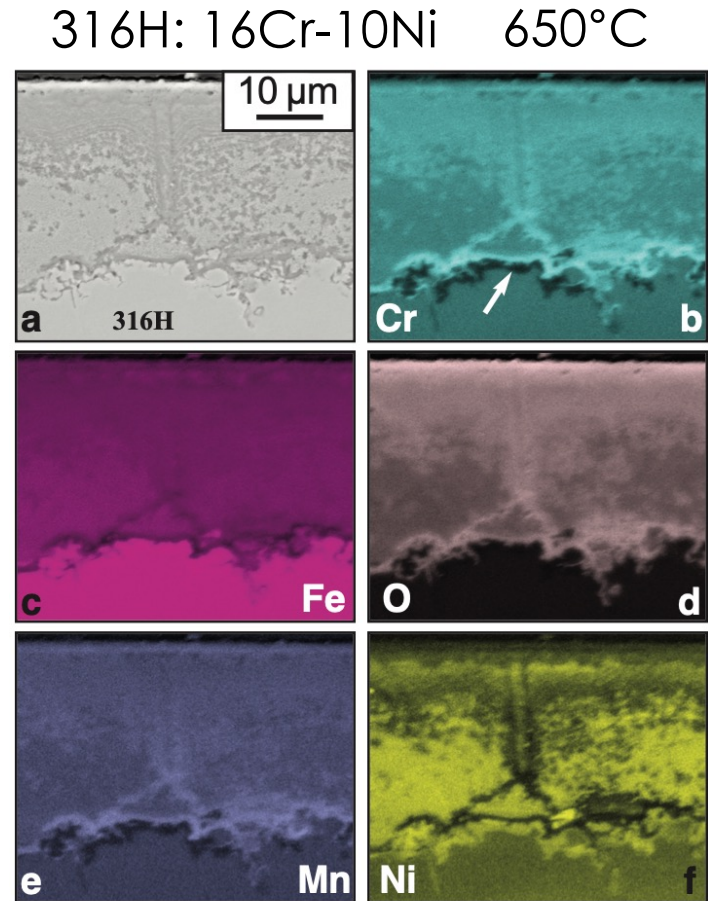
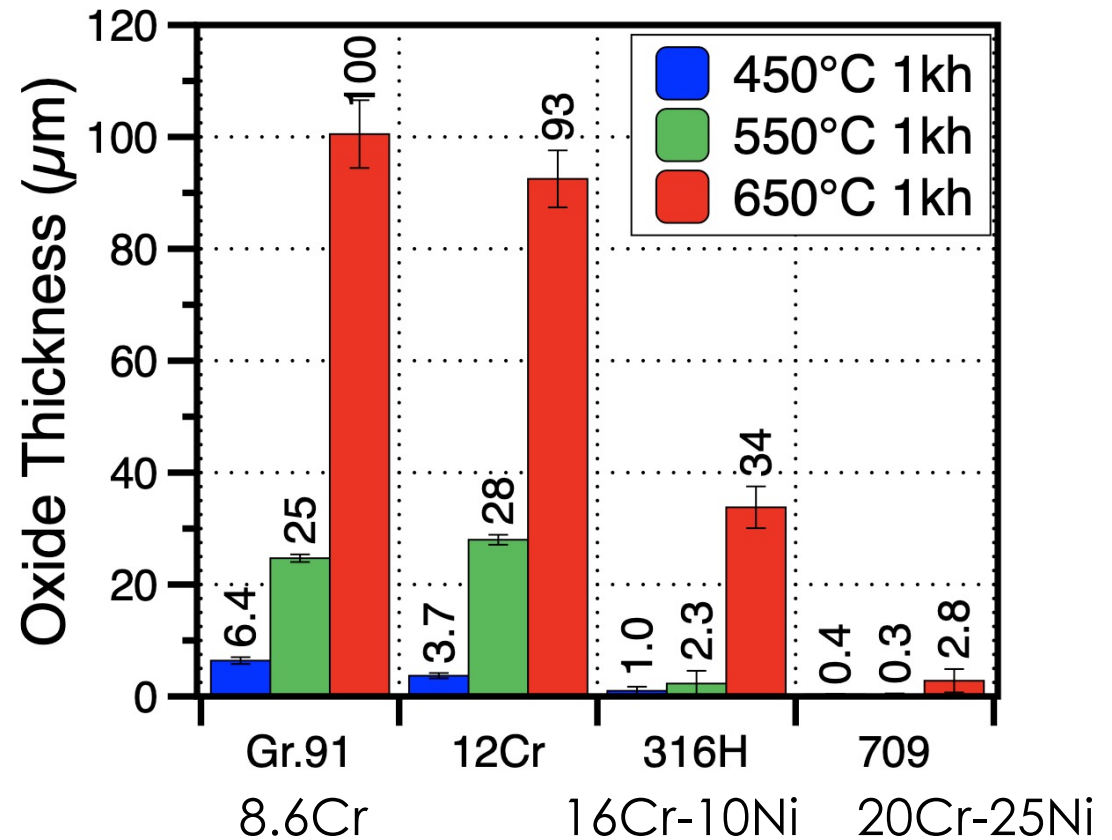


Figure 6. (a) SEM secondary electron image of the inner scale formed on 316H after 1000 h at 650°C in sCO₂ and (b-f) EDX maps of the same region.

Scanning electron microscopy/EDX maps

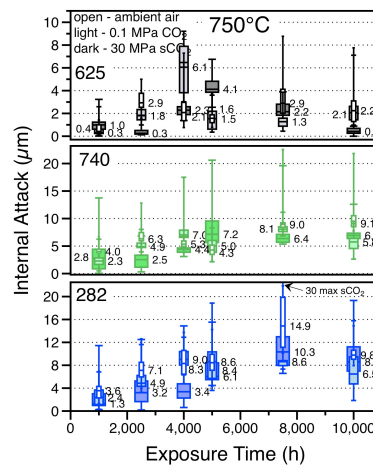
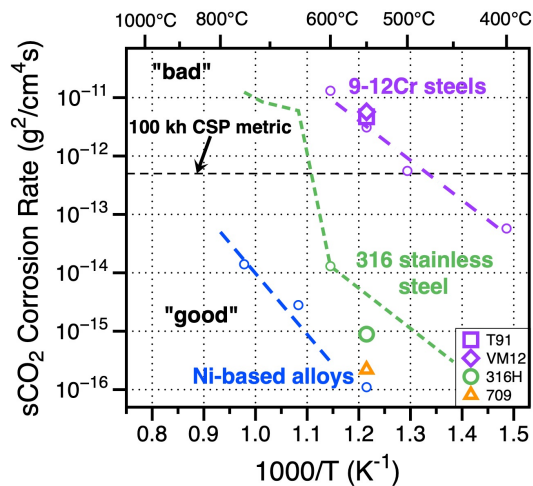
Oxide thickness after 1,000 h exposures at each temperature

- Consistent with mass change data
- Large increase for 316H at 650°C
 - ~50% of oxide spalled!

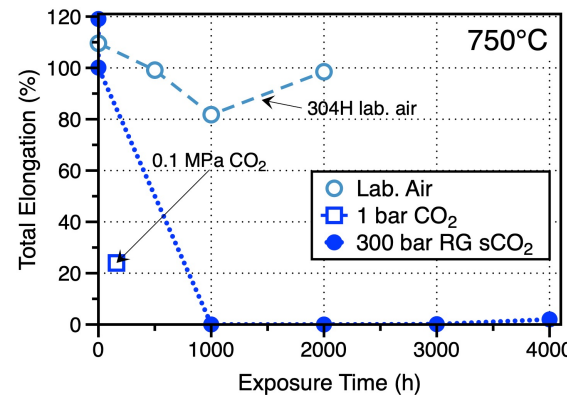


Do we need new metrics for steel sCO₂ compatibility?

Current metrics:
Mass change/rate constants/oxide thickness



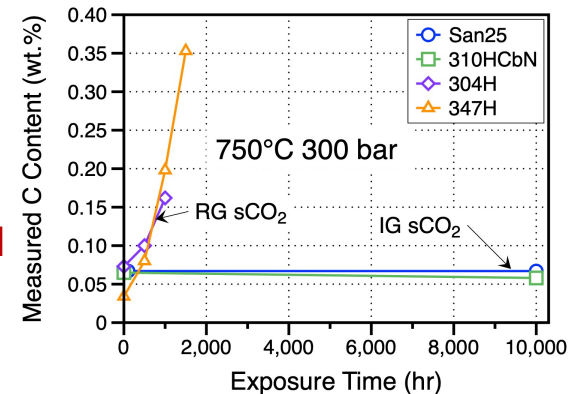
New metrics:
C uptake & retained ductility



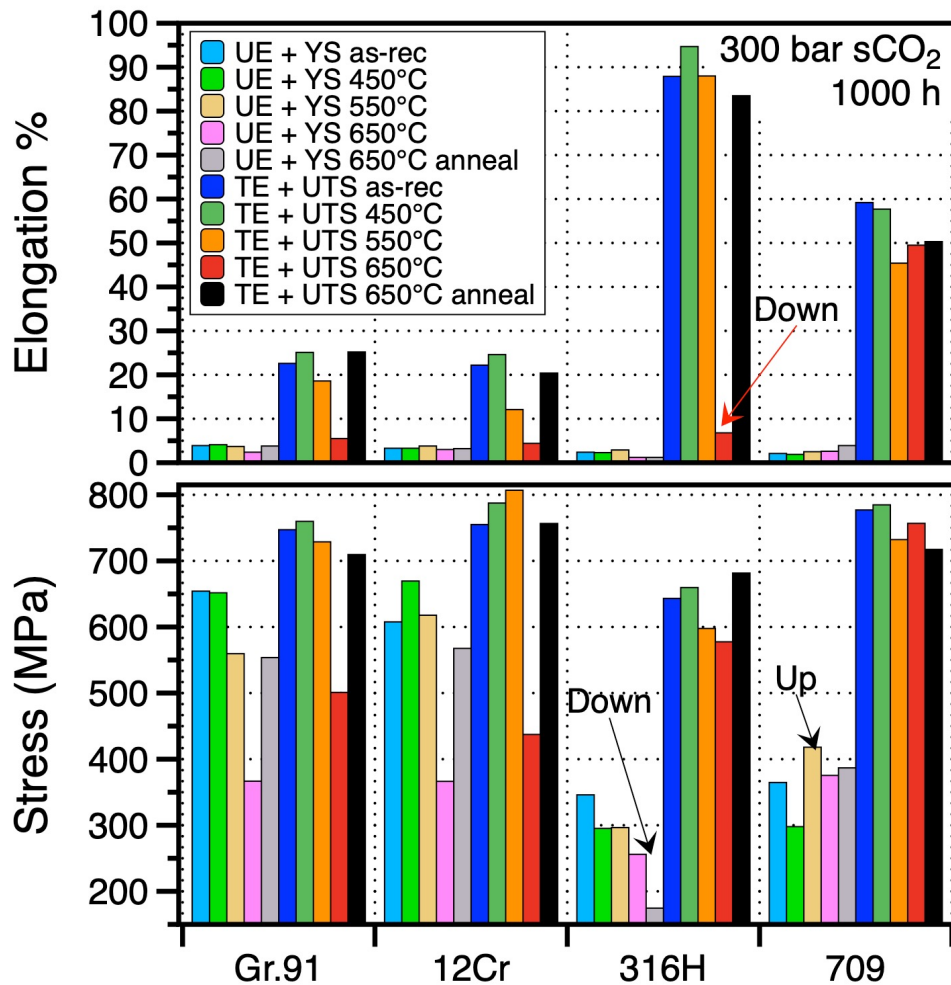
304H stainless:
embrittled in CO₂
not in air at 750°C

How useful is this information in materials selection?

750°C:
High C uptake for 304H/347H
Not for advanced austenitics



Room temperature tensile properties: 650°C/316H impact



316H/650°C:

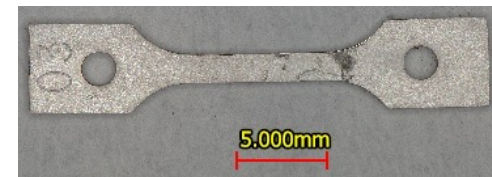
Drop in ductility

Increase in YS

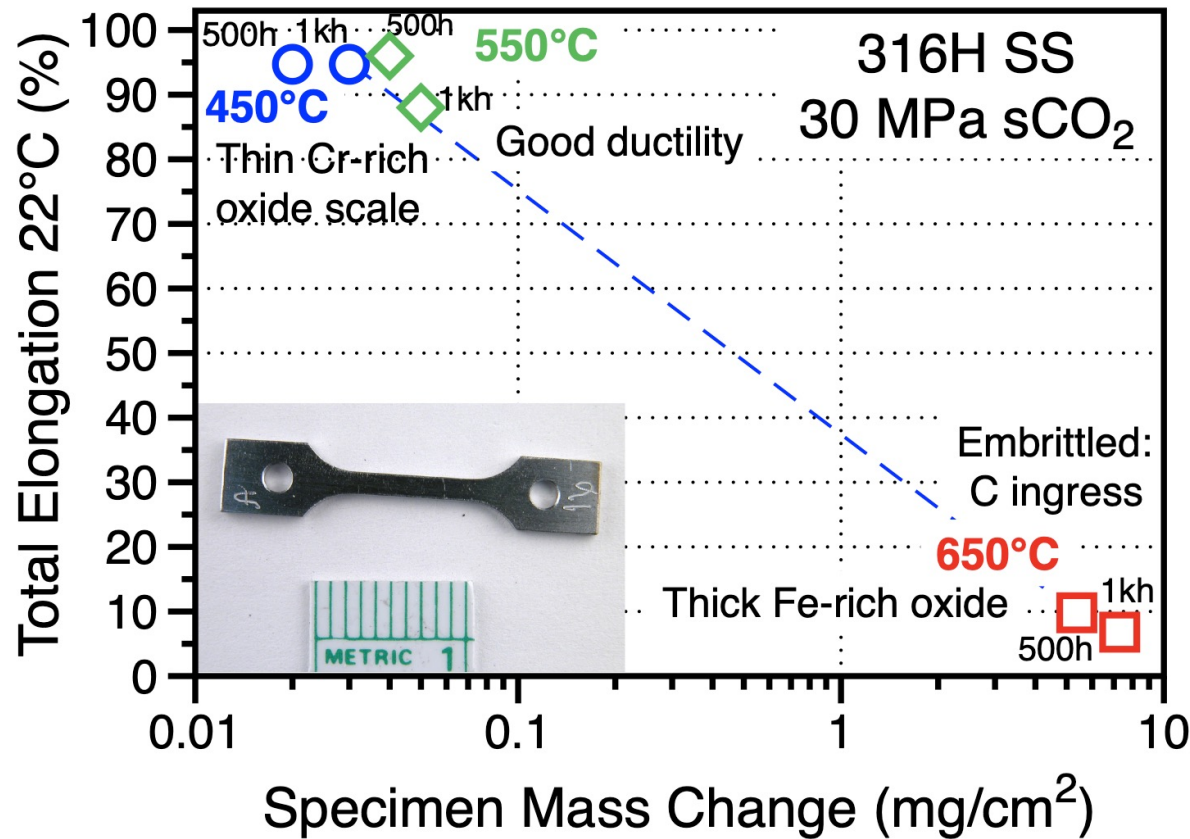
Consistent with C ingress

FS/650°C: also degraded

650°C anneal: minor loss in ductility



Room temperature tensile properties: 650°C/316H impact

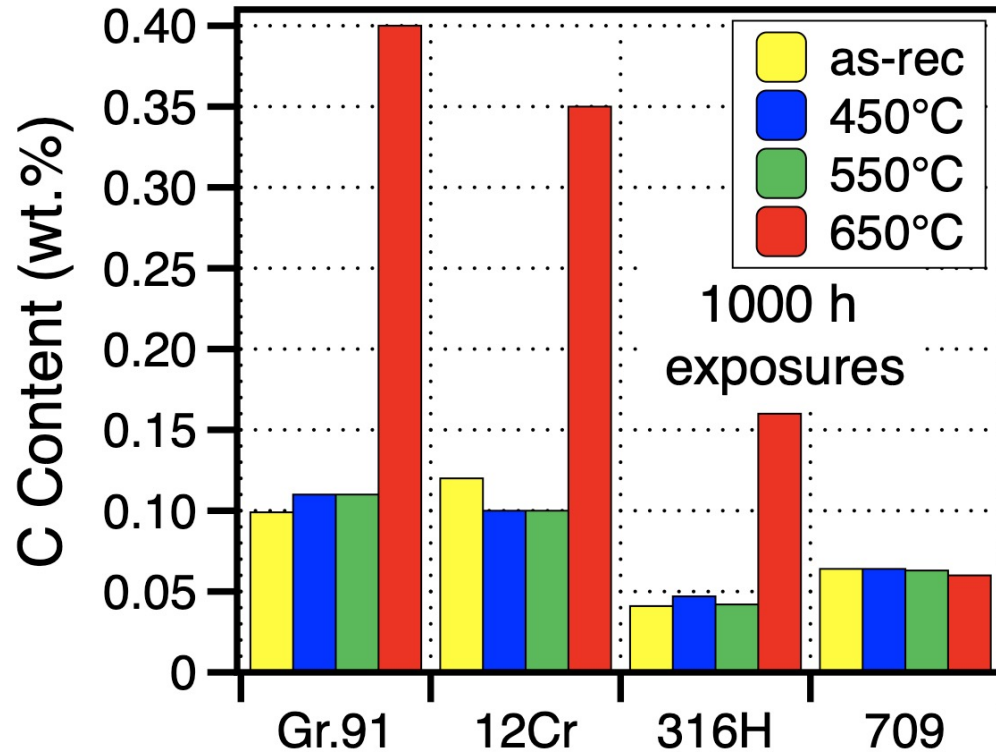


Low mass gain =
Cr-rich oxide
= high ductility
(prevented C ingress?)

High mass gain =
Fe-rich oxide
= low ductility

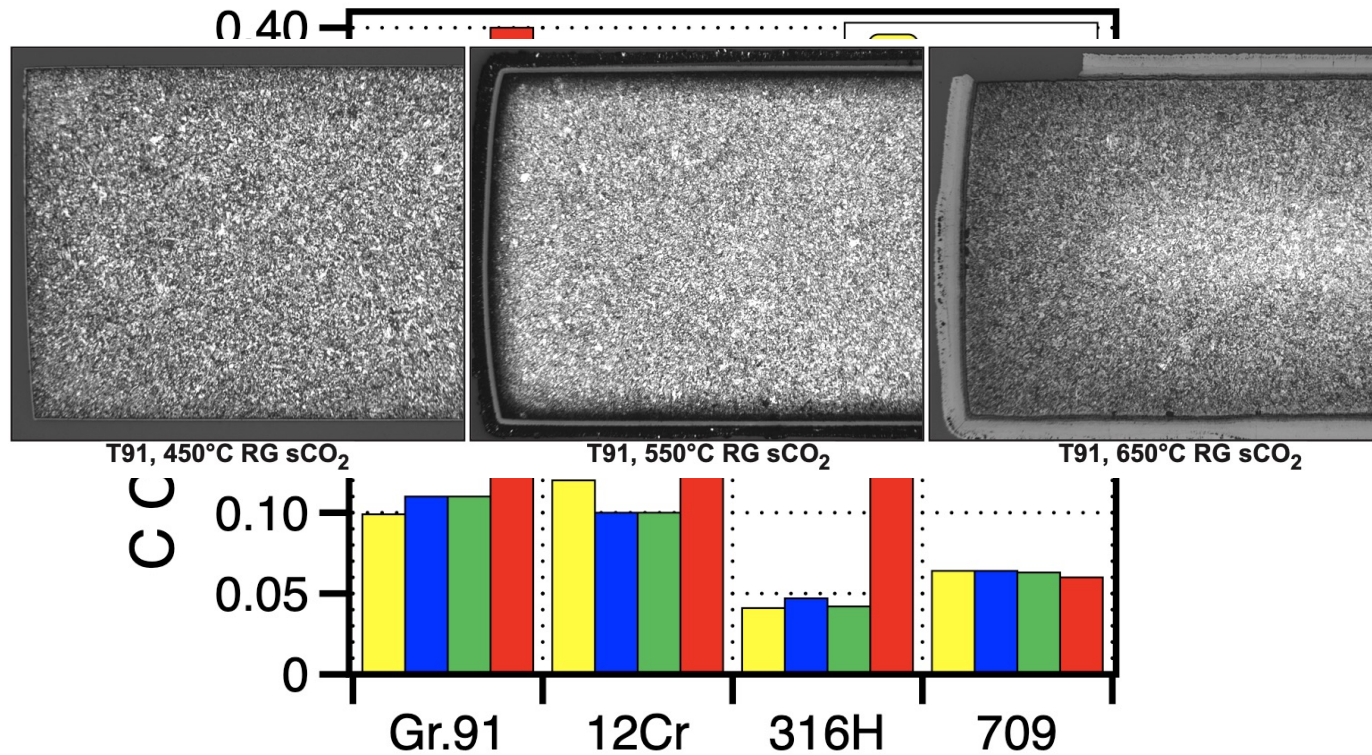
Due to C ingress?

Bulk C content: major C increase at 650°C for 3 steels



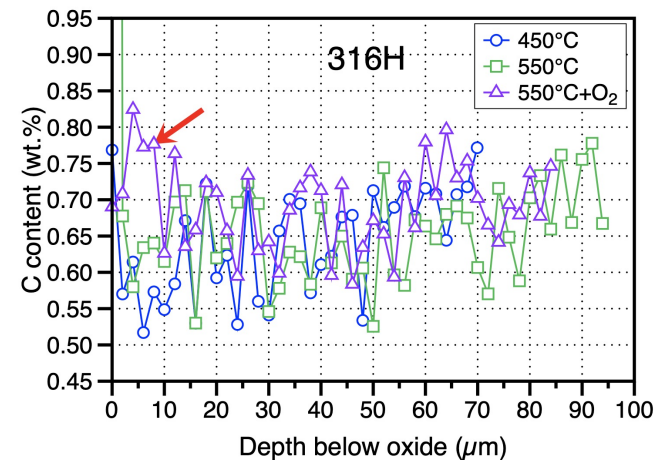
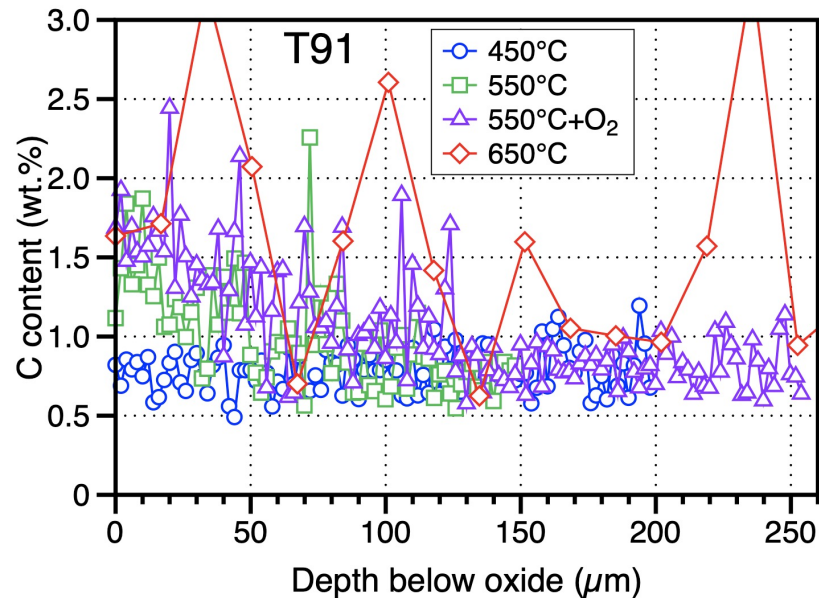
650°C: thick Fe-rich oxides allowed C ingress
550°C: longer exposures needed

Bulk C content: major C increase at 650°C for 3 steels

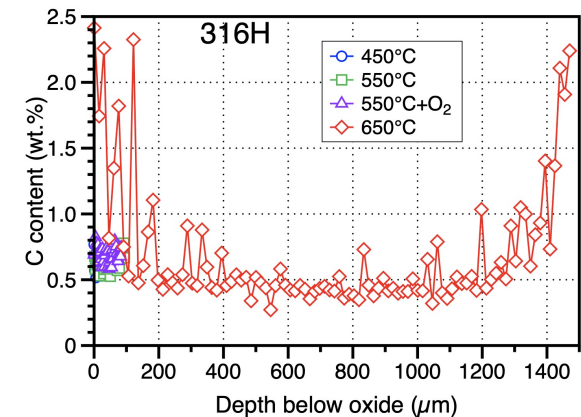


650°C: thick Fe-rich oxides allowed C ingress
550°C: longer exposures needed

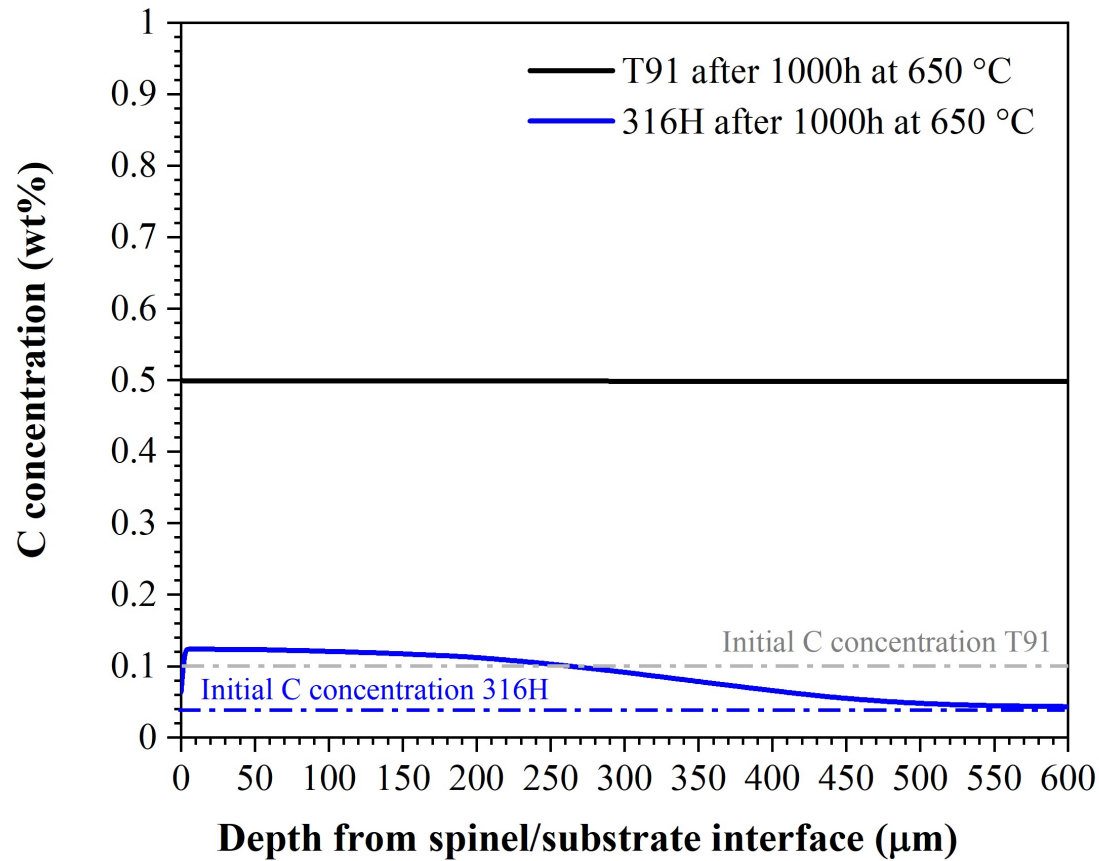
EPMA: measurements of C ingress for modeling



- Confirmed massive C uptake at 650°C for T91/316H
- Very little uptake at 450°C for both alloys
- Gradient observed at 550°C
- Collecting more EPMA data to feed modeling task

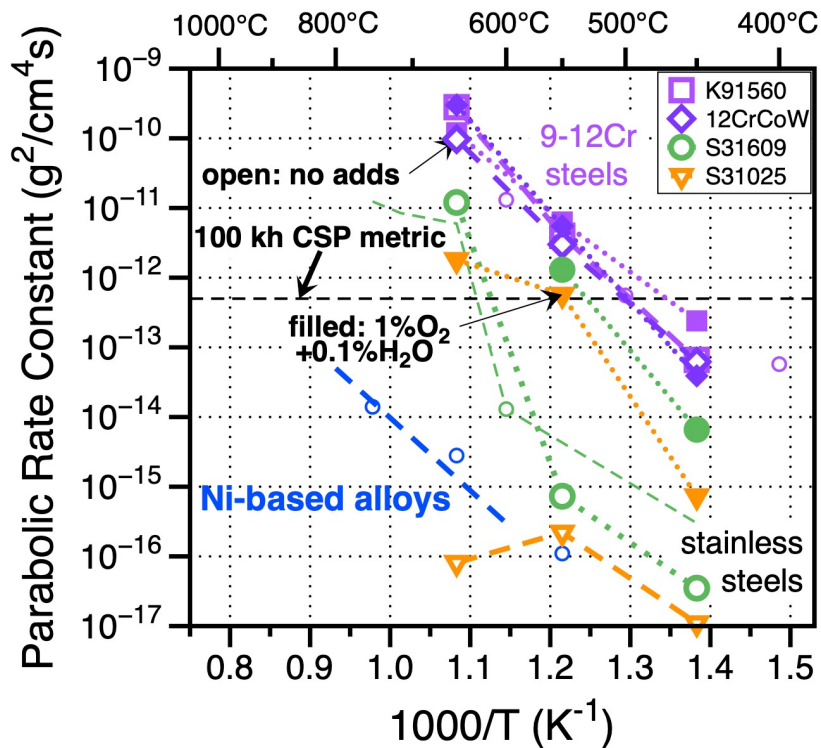


Initial steel modeling: Calculated average C profiles



Goal: predict 100,000 kh C ingress as a function of temperature for T91 and 316H

NACE 2022: Addition of 1%O₂+0.1%H₂O: accelerates all !



Open symbol: RG sCO₂
Filled: RG sCO₂ + 1%O₂ + 0.1%H₂O
 All 300 bar

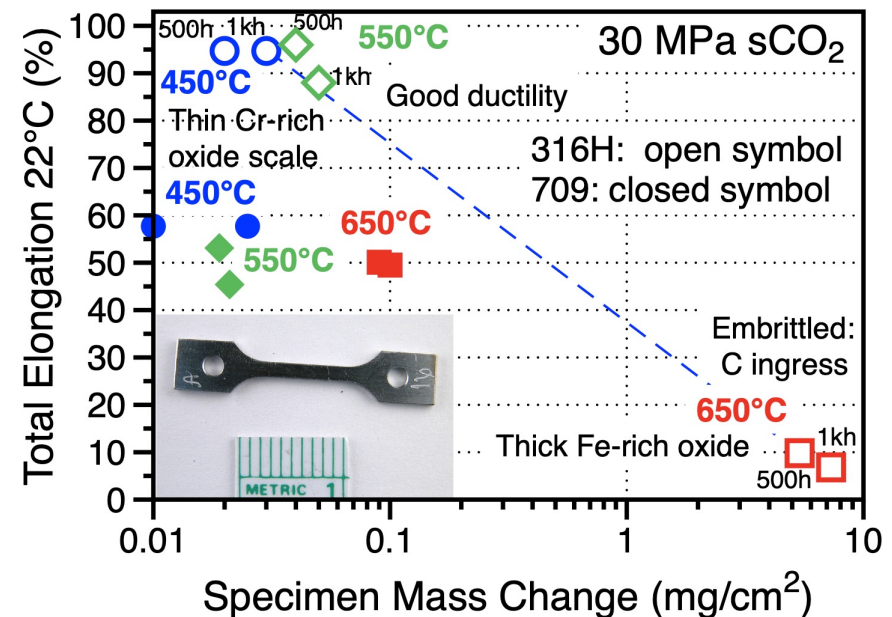
2019: Ni-based alloys 2-5X rate increase
 with impurities at 750°C

Based on CSP (solar) metric: all limited to <550°C with impurities
 9-12Cr steels: high rates with or without impurities

Rates for 709 (20Cr/25Ni): may not reflect steady state at 1000 h

Summary: $s\text{CO}_2$ is a challenging environment

- At 650°-800°C, Ni-based alloys appear compatible
- Steels have problem forming protective scales:
 - 9-12%Cr all conditions 450°-650°C
 - Fe-rich oxide formation observed in $s\text{CO}_2$
 - These steels form Cr-rich scale in air
 - 316H at 650°C in RG $s\text{CO}_2$
 - Carbon ingress + embrittlement
 - 709 formed Cr-rich oxide in all cases
 - Longer times at 650°C?
 - 310HCbN/alloy 25: no C ingress at 750°C
 - With O_2 impurities: all have problems!



Thank you!

All 650°C, 500 h, sCO₂

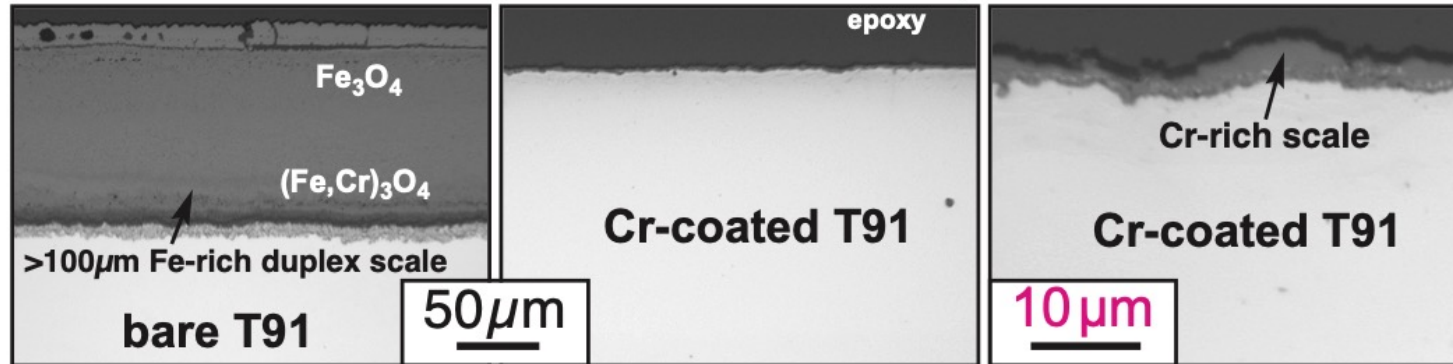


Table 1
Alloy chemical compositions (weight %)(1)

Alloy	UNS#	Fe	Cr	Ni	Mo	Mn	Si	C	S(ppm)	Other
Gr.91	K90901	88.8	8.6	0.3	0.9	0.46	0.35	0.10	6	0.2V,0.06Nb,0.045N
12CrCoW		83.3	11.5	0.4	0.4	0.38	0.42	0.12	3	1.6W,1.5Co,0.2V,0.04N
316H	S31609	69.5	16.3	10.0	2.0	0.84	0.46	0.041	6	0.3Cu,0.3Co,0.04N
709	S31025	51.3	20.1	25.2	1.5	0.89	0.41	0.064	3	0.2Nb,0.06Cu,0.15N

(1) measured by inductively coupled plasma analysis and combustion analysis
< indicates below the detectability limit of <0.01%