

Coated and Uncoated Steel Compatibility in Supercritical CO₂ at 450°-650°C

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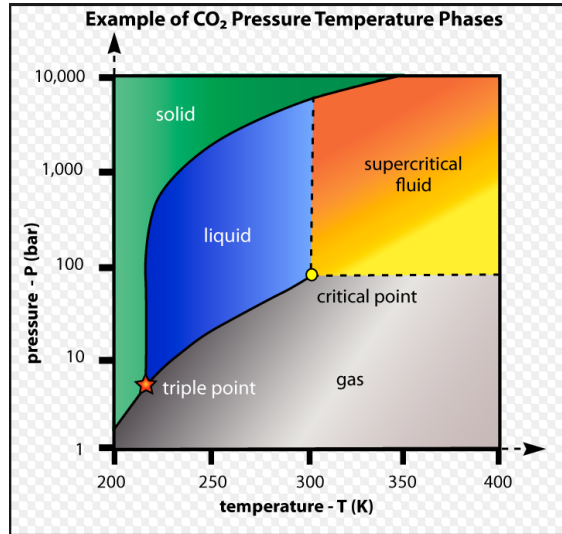
Oak Ridge National Laboratory, Oak Ridge, TN

Acknowledgments

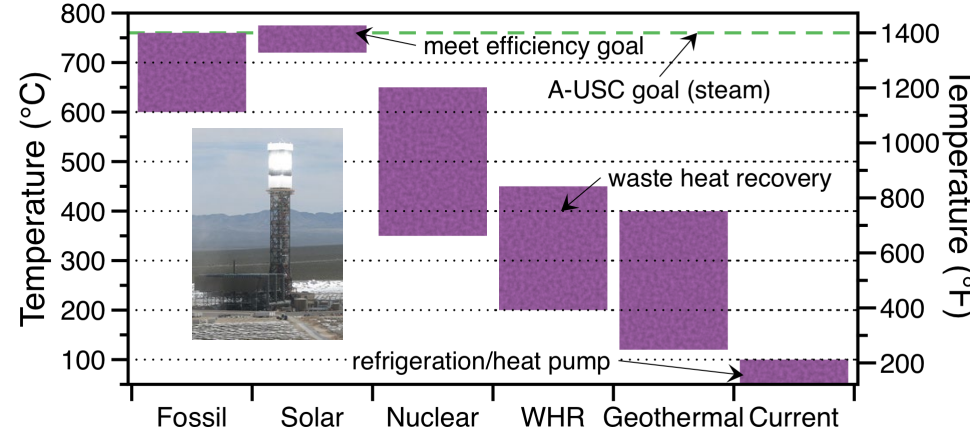
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- B. Johnston, G. Garner — oxidation experiments
- T. Lowe — SEM, image analysis
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 - Tenaris, T91
 - EPRI, VM12
 - Sam Sham, INL (US heat, alloy 709)

Most questions over 10 years were about steels & coatings!

sCO₂ has many unique & attractive aspects

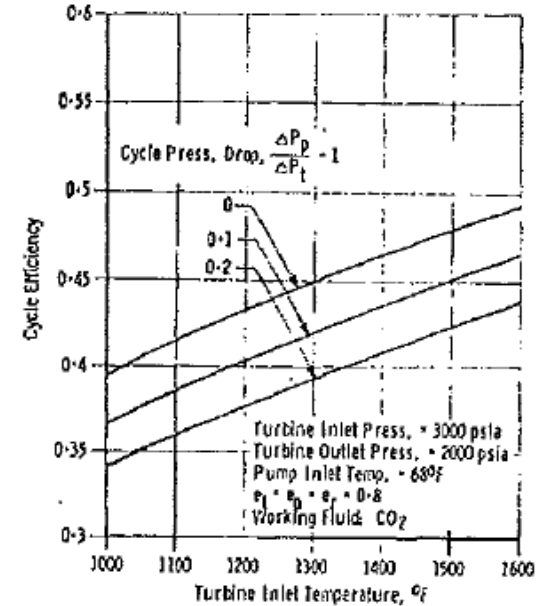


Supercritical CO₂ applications



Interest in >700°C for high efficiency

**Cost is a concern:
Ni-based alloys: OK in sCO₂
Where can steels be used?
Can we coat them?**



Feher, 1965

50% sCO₂ eff @ >720°C

- Low critical point (31°C/7.4 MPa)
- High, liquid-like density
- Flexible, small turbomachinery

Supercritical CO₂ Allam cycle: first clean fossil energy?

NetPower 25MWe test facility (Texas)

Exelon, Toshiba, CB&I, 8Rivers Capital: \$140m

CO₂-free natural gas? CCS project powers grid for first time

By Carlos Anchondo, Edward Klump | 11/17/2021 07:07 AM EST



May 2018: announced first firing
November 2021: 1st grid power

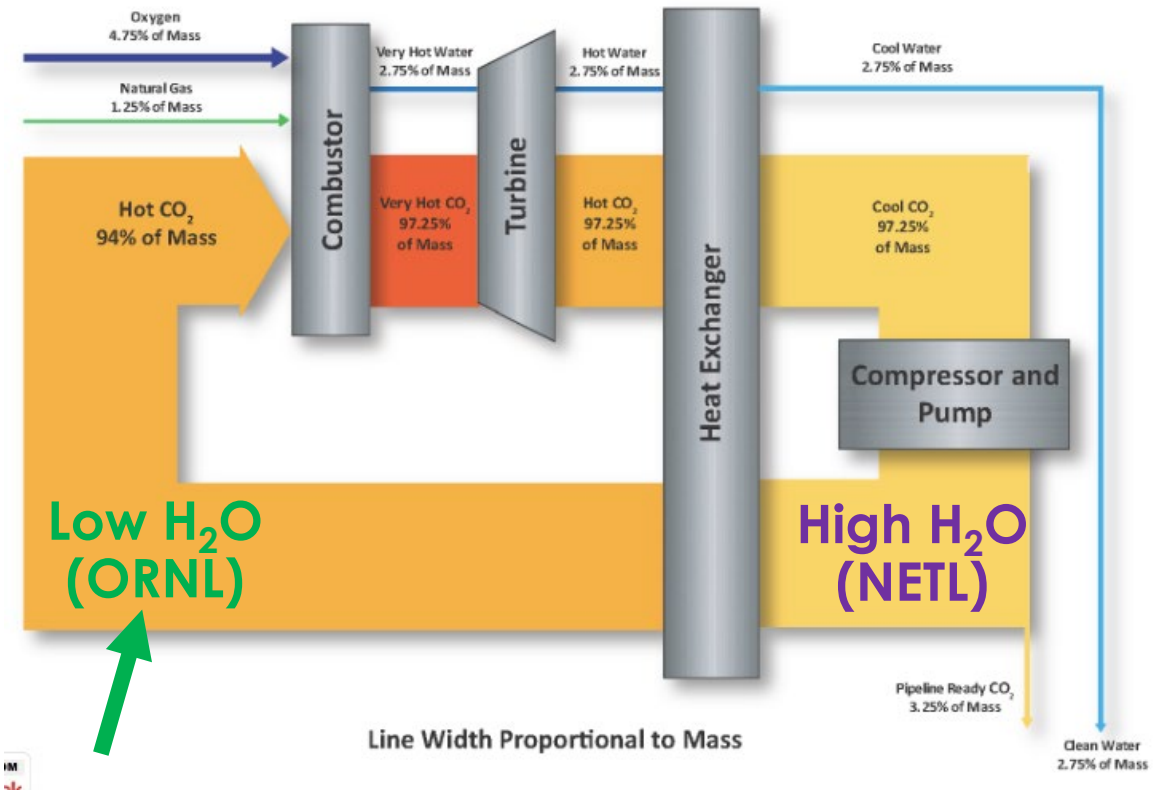
Burning natural gas in sCO₂ creates impurities...what is effect?

Material challenges:

Combustor: **900°C**

Turbine exit: 750°C/300 bar

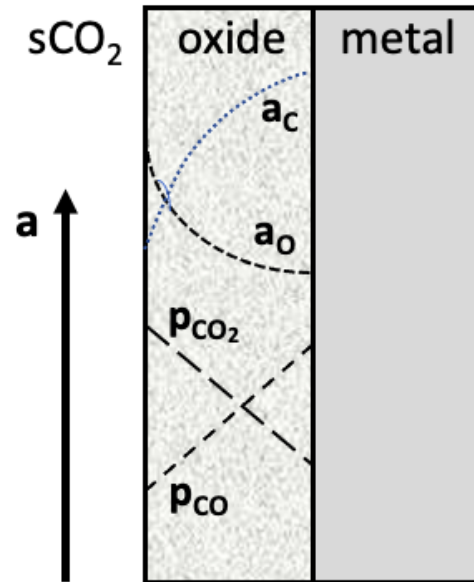
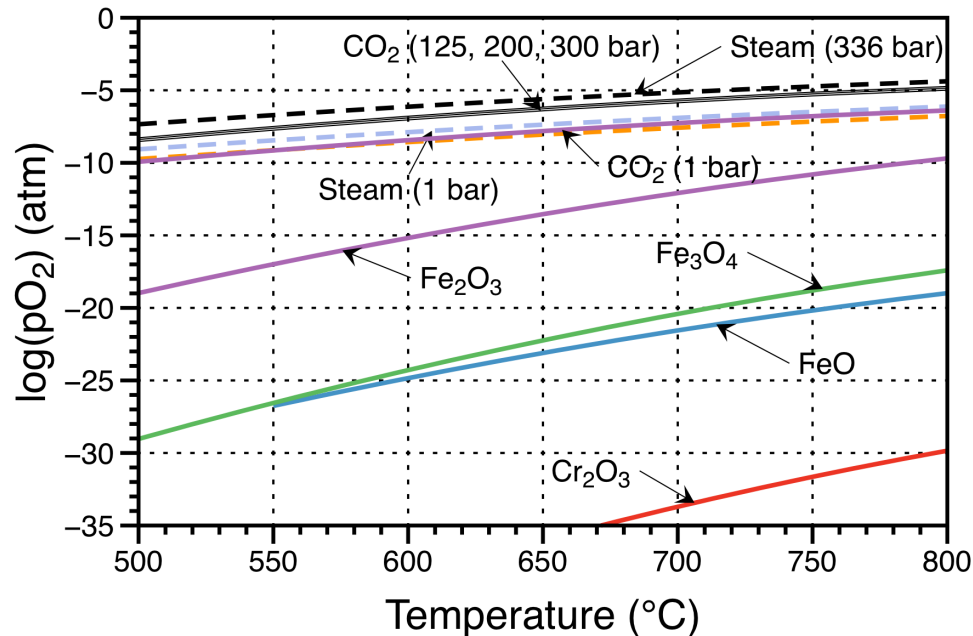
Combustion impurities: O₂, H₂O, SO₂



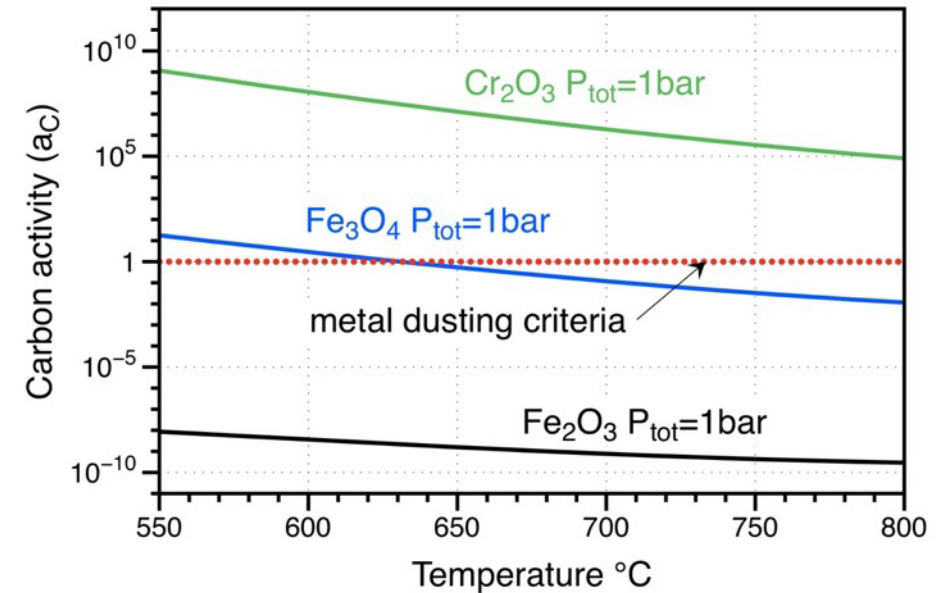
Thermodynamics: Oxygen levels similar in steam/CO₂

Concern about high C activity at m-o interface

Factsage calculations



After Young, et al., 2011



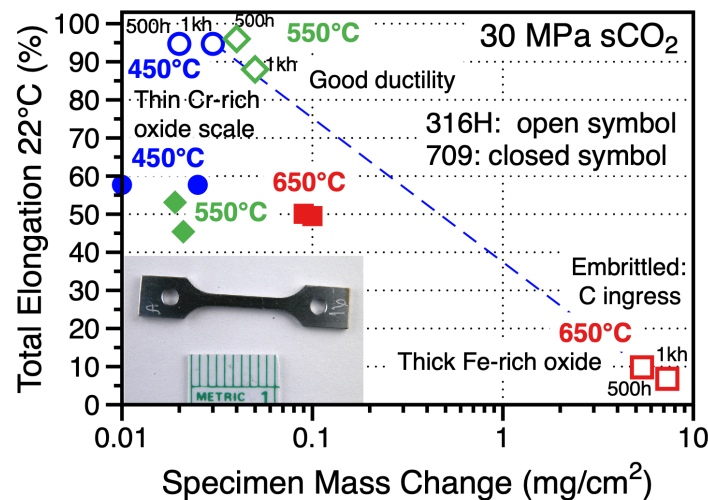
High carbon activity at $P_{total} = 1$ bar (What is $P_{interface}$?)

General conclusion: internal carburization concern for Fe-based alloys

sCO₂ compatibility: broad range of conditions considered

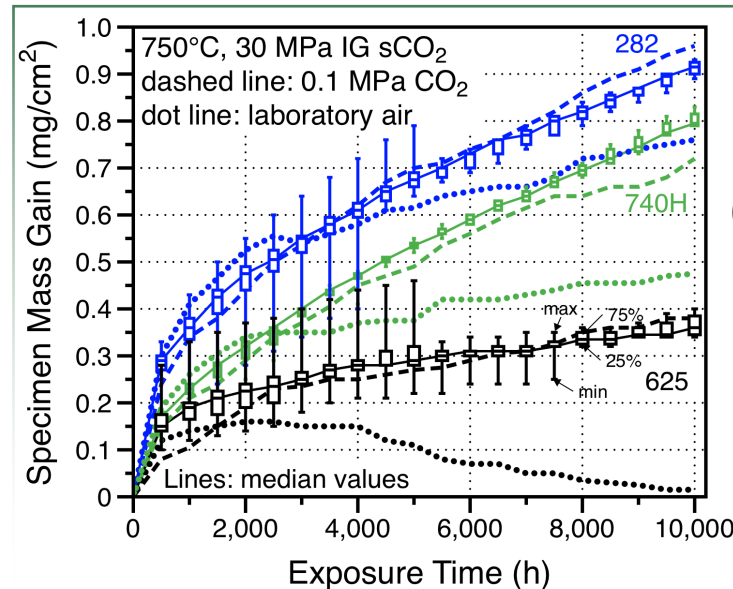
400°-650°C: concern about steel carburization

- Well-known issue from CO₂-cooled reactors
 - Grade 9 steel current issue
- 550°-600°C transition temperature for normal austenitic steels
- Key to low-cost technology**



650°-800°C: Ni-based alloys

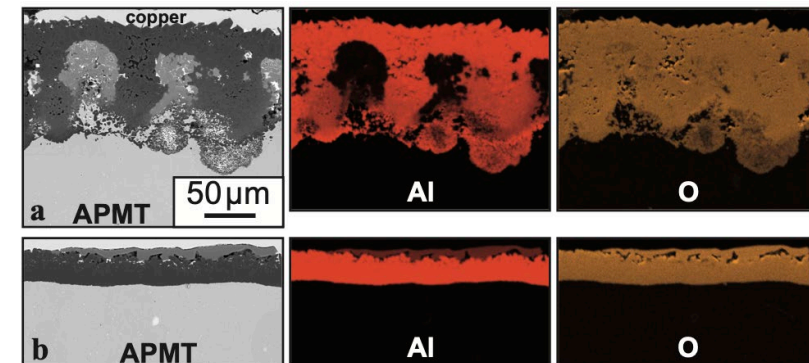
- No issues for Ni-based alloys
 - Low C solubility, protective Cr₂O₃ formation
- Similar rates for air, CO₂ and sCO₂
 - Little or no P effect @ 750°C**



>800°C: challenging for superalloys/cermets/FeCrAl

- Initial results at 0.1 & 2 MPa
 - Subcritical P effect observed**
- Mo/W cermets need coating
- Accelerated attack of Ni-based superalloys
- SiC promising, but not MoSi₂
- FeCrAl attacked at 1200°C
 - Al₂O₃ supposed protective?

0.1 MPa



2 MPa CO₂

ORNL steel project started in August 2019

Test matrix & progress

Temperature	RG sCO ₂	+1%O ₂ +0.1%H ₂ O
450°C (842°F)	2,000 h	1,000 h
550°C (1022°F)	2,000 h	1,000 h
650°C (1202°F)	2,000 h	1,500 h



**Autoclave: 300 bar sCO₂
500-h cycles**



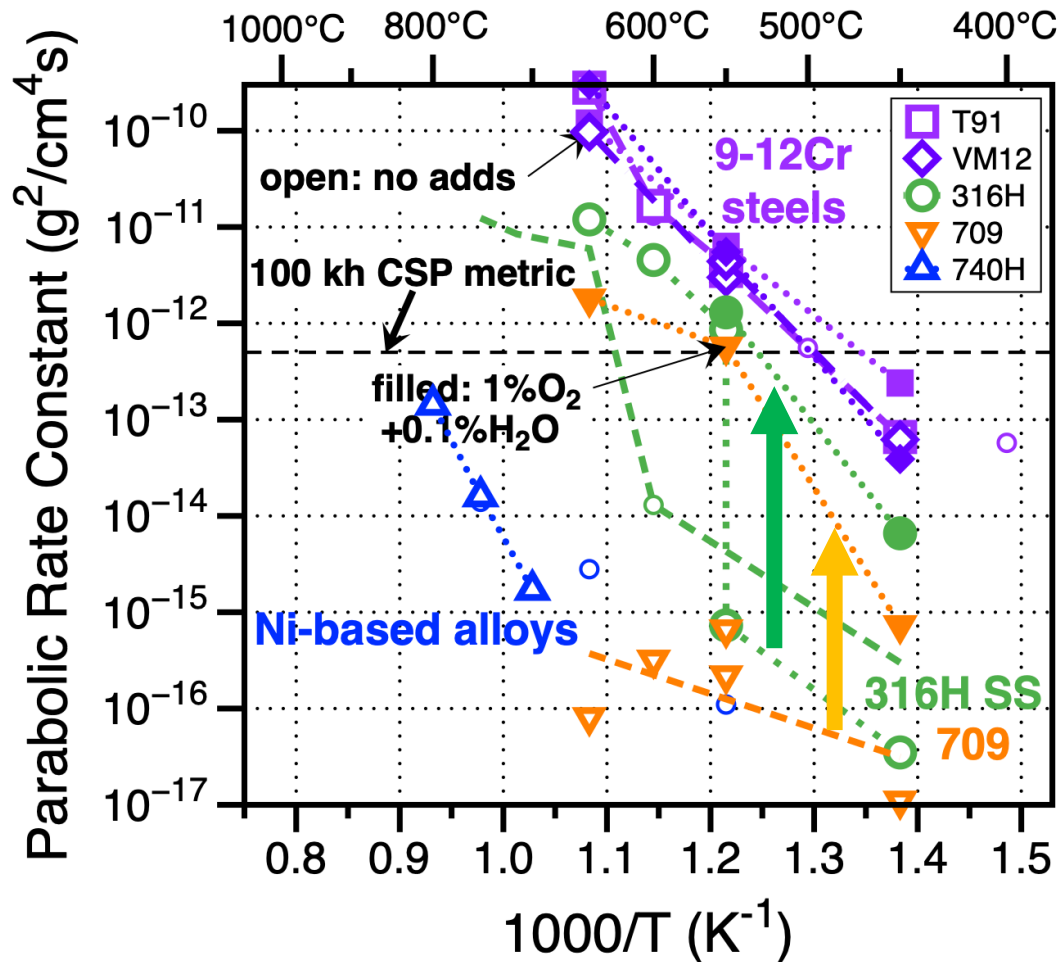
~5 cm² alloy coupons + tensile specimens

- Four primary alloys in test matrix
 - T91: Fe-9Cr-1Mo, creep strength enhanced steel
 - VM12: Fe~12Cr-Co-W
 - 316H: conventional stainless steel
 - 709: advanced austenitic, 20Cr-25Ni+Nb
- 10 specimens of each alloy
- With and without impurities (open vs. closed)

Alloy	UNS	Cr	Ni	Mn	Si	C	N	Other
Gr.91	K90901	8.6	0.3	0.5	0.4	.10	.05	0.9Mo,0.2V
VM12	12CrCoW	11.5	0.4	0.4	0.4	.12	.04	1.6W,1.5Co
316H	S31609	16.3	10.0	0.8	0.5	.04	.04	2.0Mo,0.3Co
NF709	S31025	20.1	25.2	0.9	0.4	.06	.15	1.5Mo,0.2Nb

Baseline of research grade (RG) CO₂: ≤ 5 ppm H₂O and ≤ 5 ppm O₂

Addition of 1%O₂+0.1%H₂O: accelerates SS mass gains!



Bad

Performance metric from CSP

Good

Open symbol: RG sCO₂

Shaded: RG sCO₂ + 1%O₂ + 0.1%H₂O

- no change for Gr91 and VM12

- higher rates for 316H and 709

Based on CSP (solar) metric: all limited to <550°C with impurities

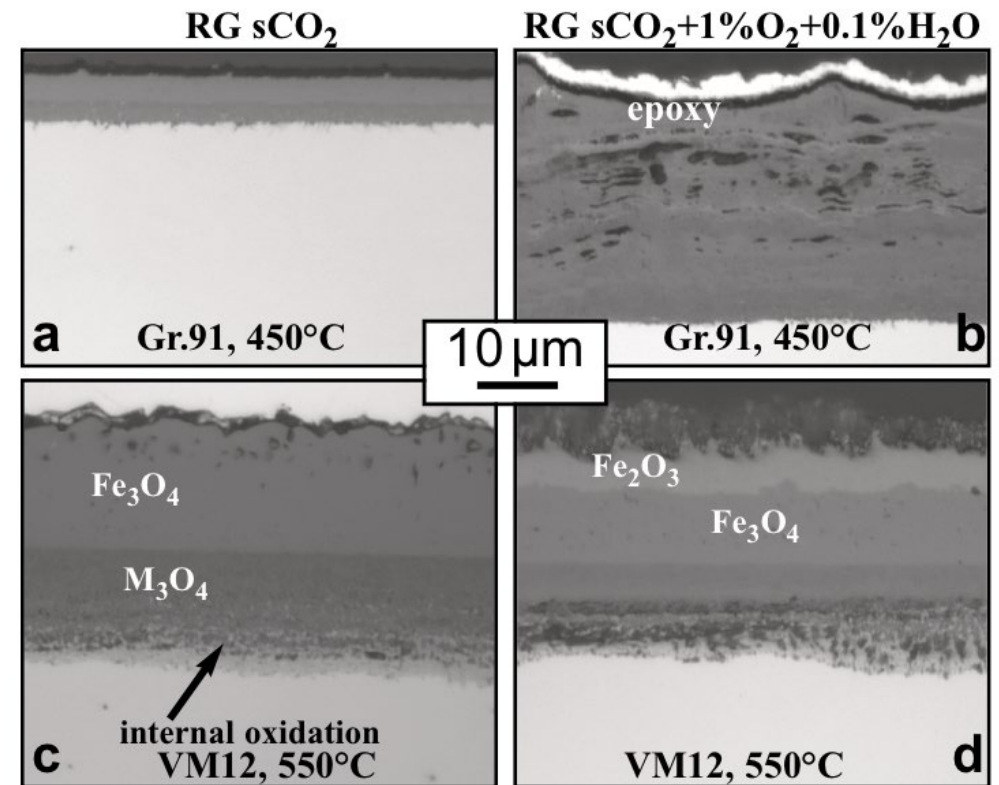
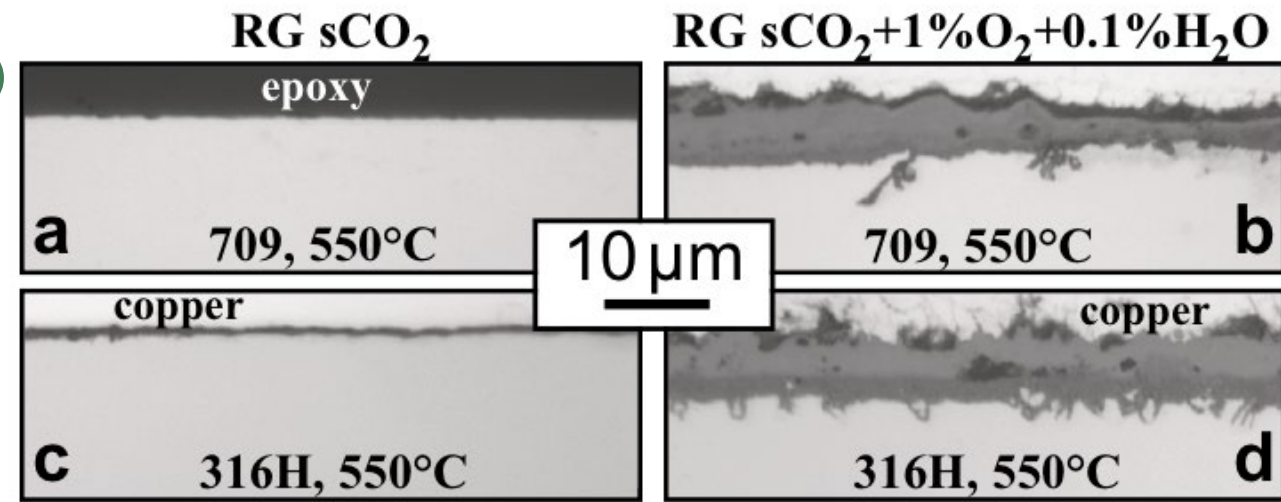
Rates for 709 (UNS S31025): may not reflect steady state at 1000 h

Similar T91 rates in steam. Are we really concerned about rates?

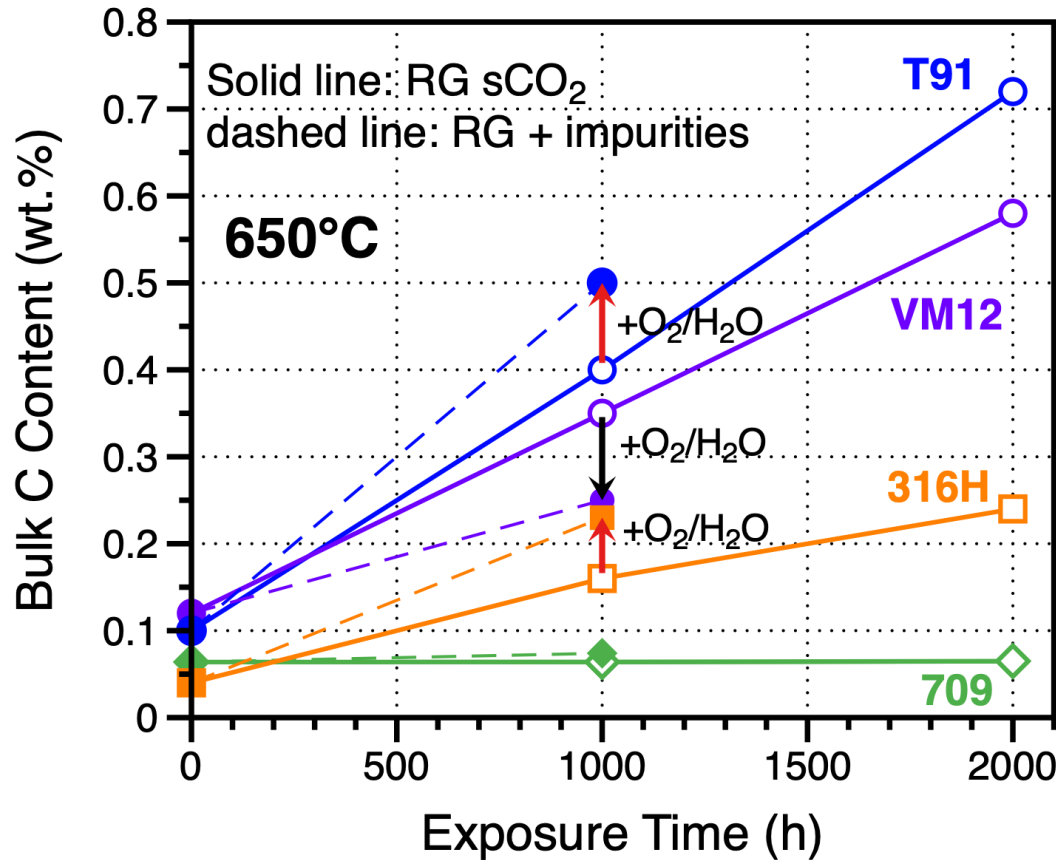
SS: faster attack with O_2/H_2O

FeCr: Fe_2O_3 formation

- Stainless steel (316/709)
 - RG sCO_2 : thin Cr-rich scale
 - Good C barrier
 - Impurities: duplex scale
 - Common with steels
 - Now is C ingress possible?
- 9-12Cr steels
 - 450°C: increased T91 attack
 - 550°-650°C
 - Clear duplex scale in both cases
 - With 1% O_2 form Fe_2O_3 layer
 - VM12: no benefit of higher Cr content after 1,000 h



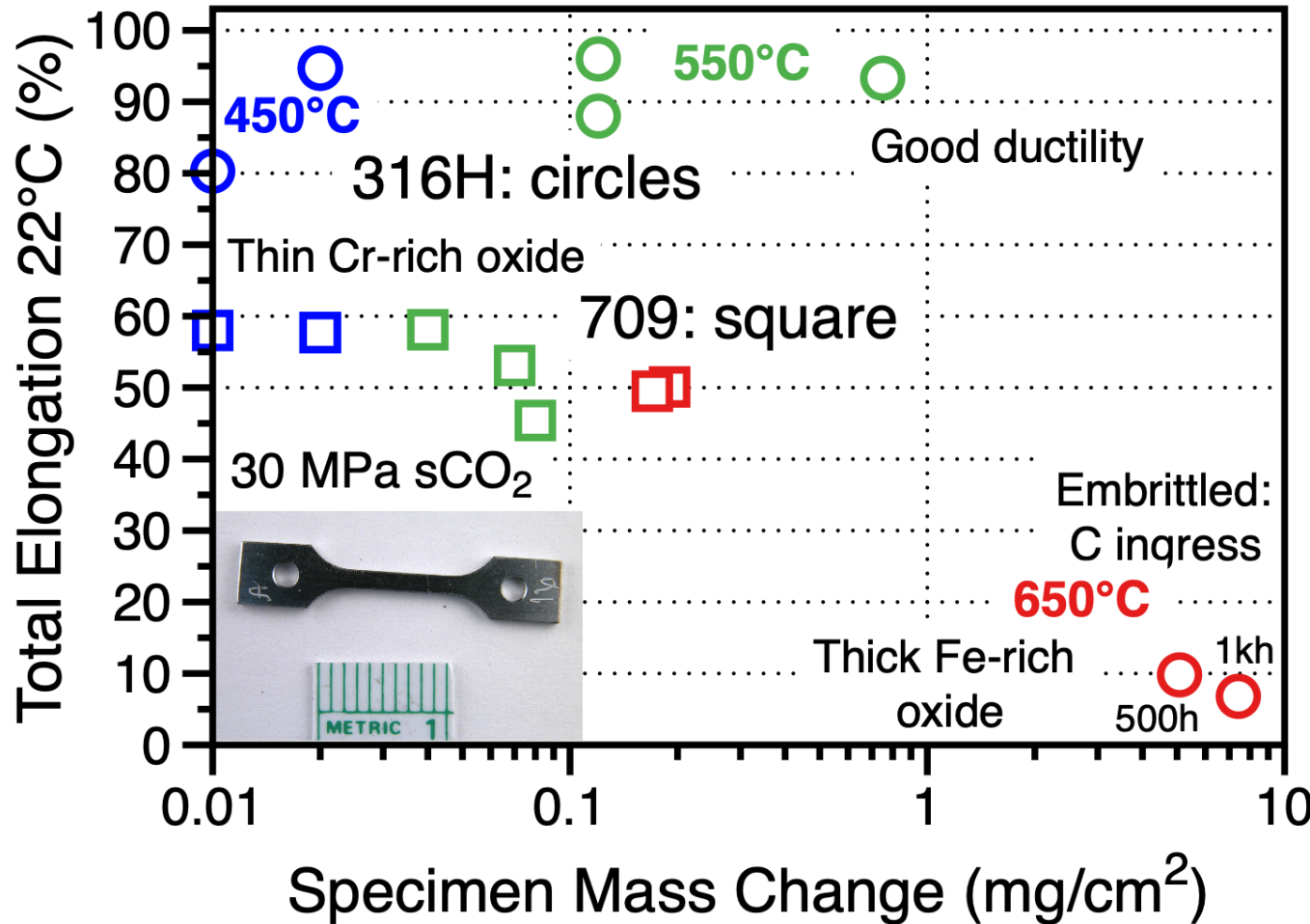
Bulk C measurements: Fe-rich oxides allow C ingress



- Measurements by combustion analysis, increasing with time
- Focus on 650°C results, less ingress at 600° and 550°C
- sCO₂ impurities tend to increase C ingress (but not all cases)

RG sCO₂: combination of mass change & 25°C ductility to illustrate issue with 316H at 650°C (709 not affected)

High ductility & low mass gain



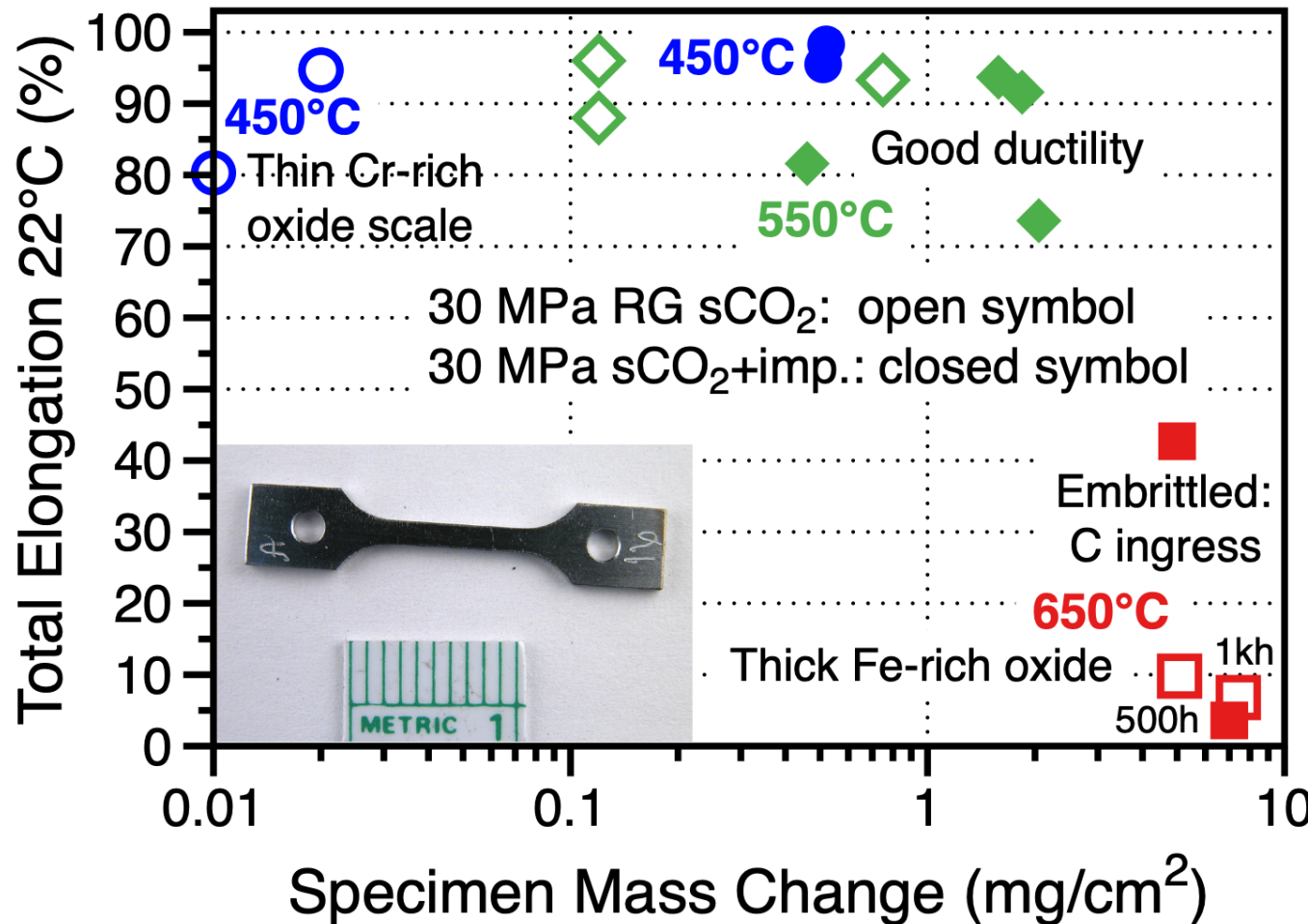
500h & 1000 h measurements for each stainless steel

Low ductility & high mass gain

316H: not much change in 25°C ductility with impurities

More mass gain, similar embrittlement at 650°C

High ductility & low mass gain

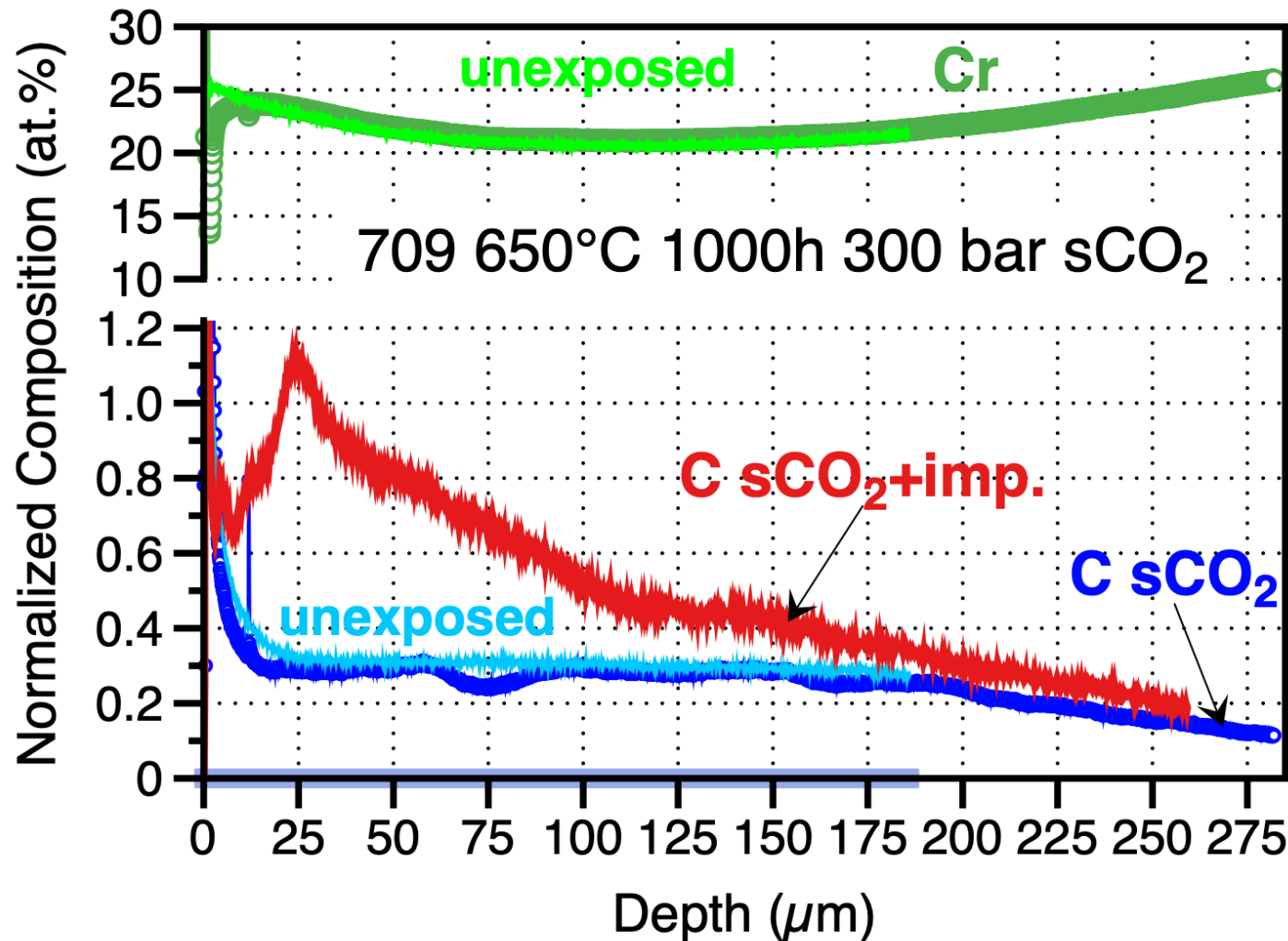


500h & 1000 h measurements for 316H

Low ductility & high mass gain

GDOES of 709 specimens after 1000 h

GDOES: Glow discharge optical emission spectroscopy

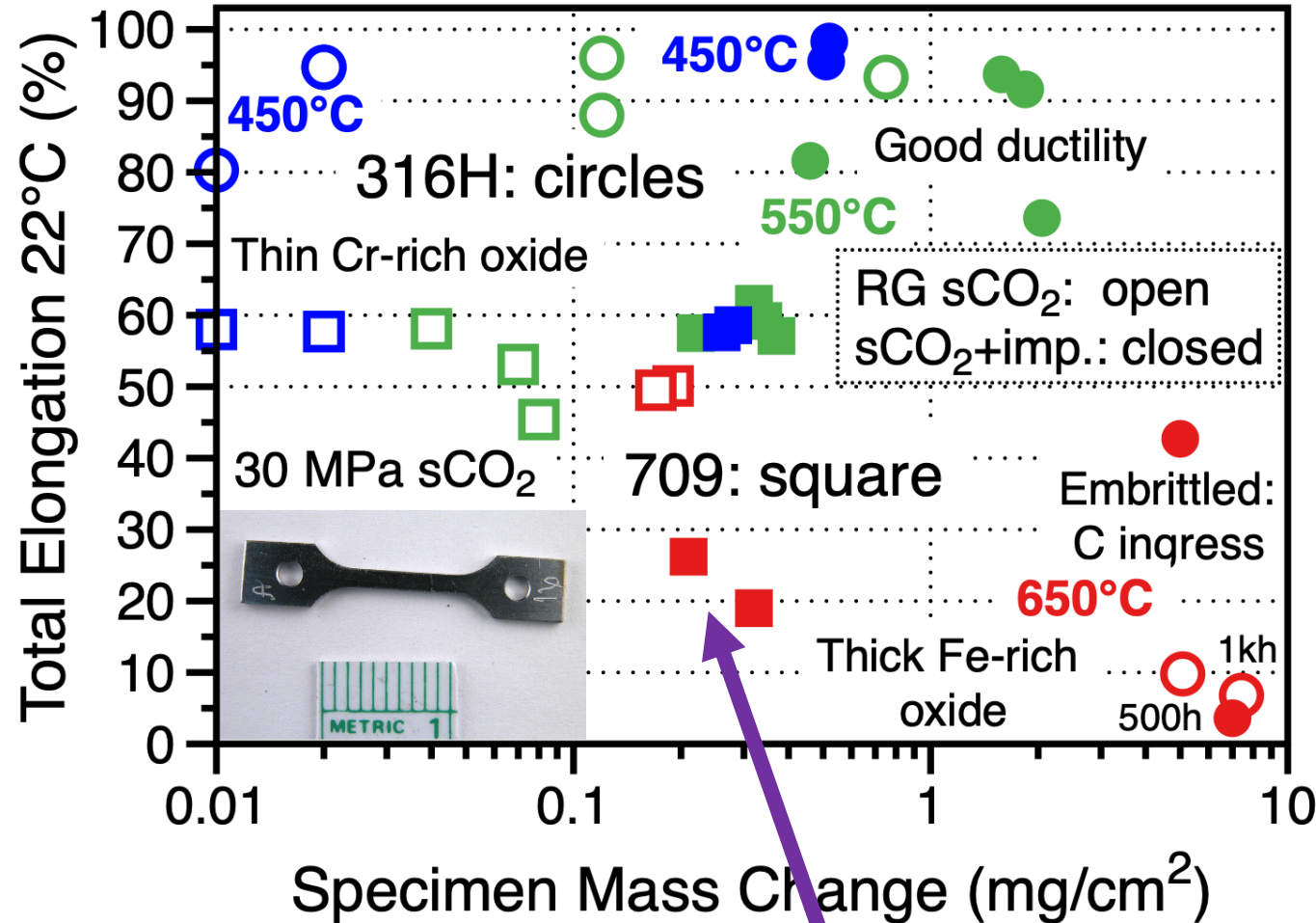


650°C: more C ingress with impurities

GDOES: signal normalized based on starting measured composition

709 showed drop in 25°C ductility at 650°C with the addition of O₂/H₂O impurities

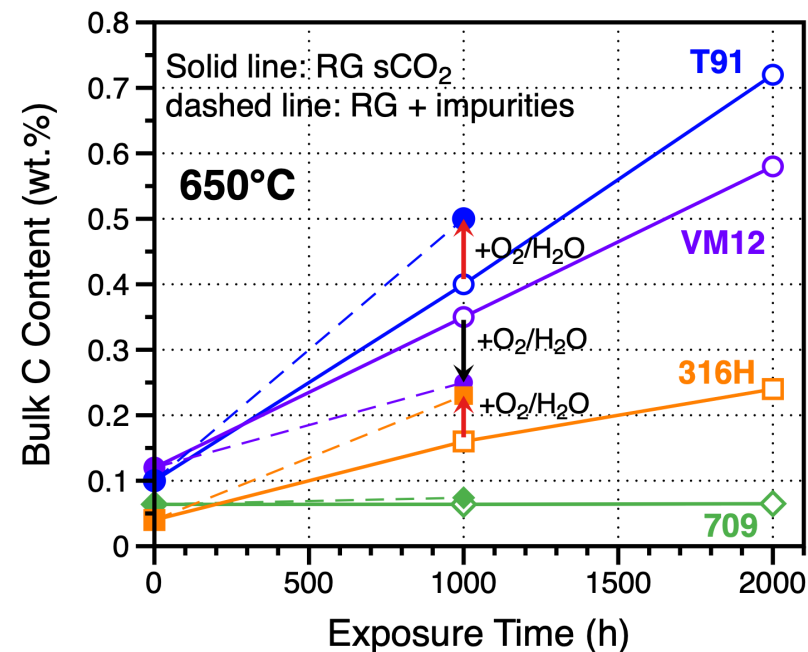
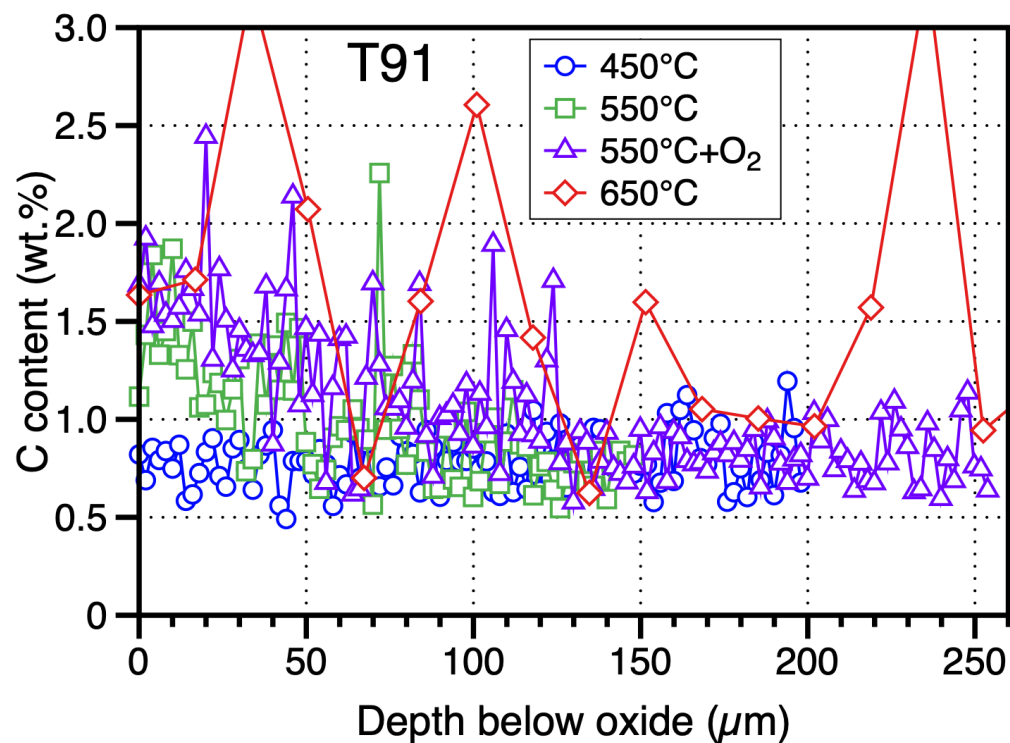
High ductility & low mass gain



500h & 1000 h measurements for each stainless steel

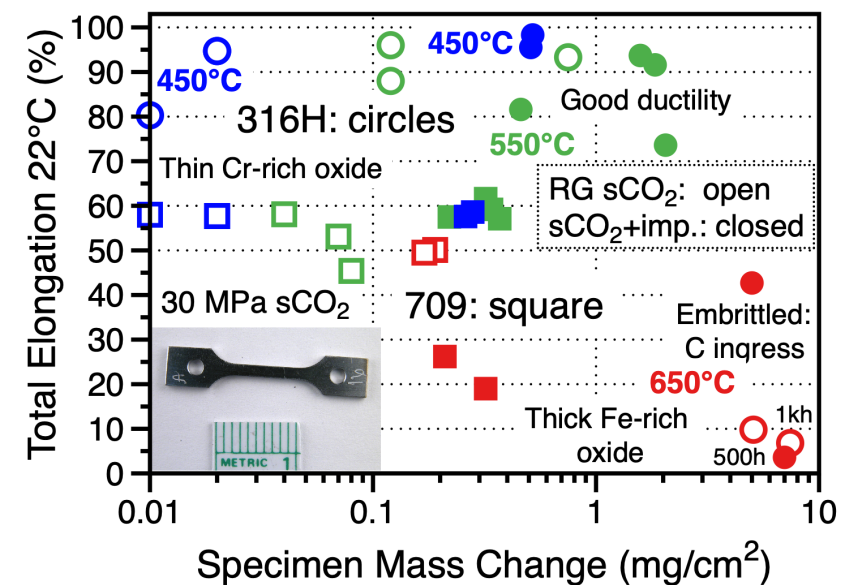
Low ductility & high mass gain

Observed massive internal carburization of steels

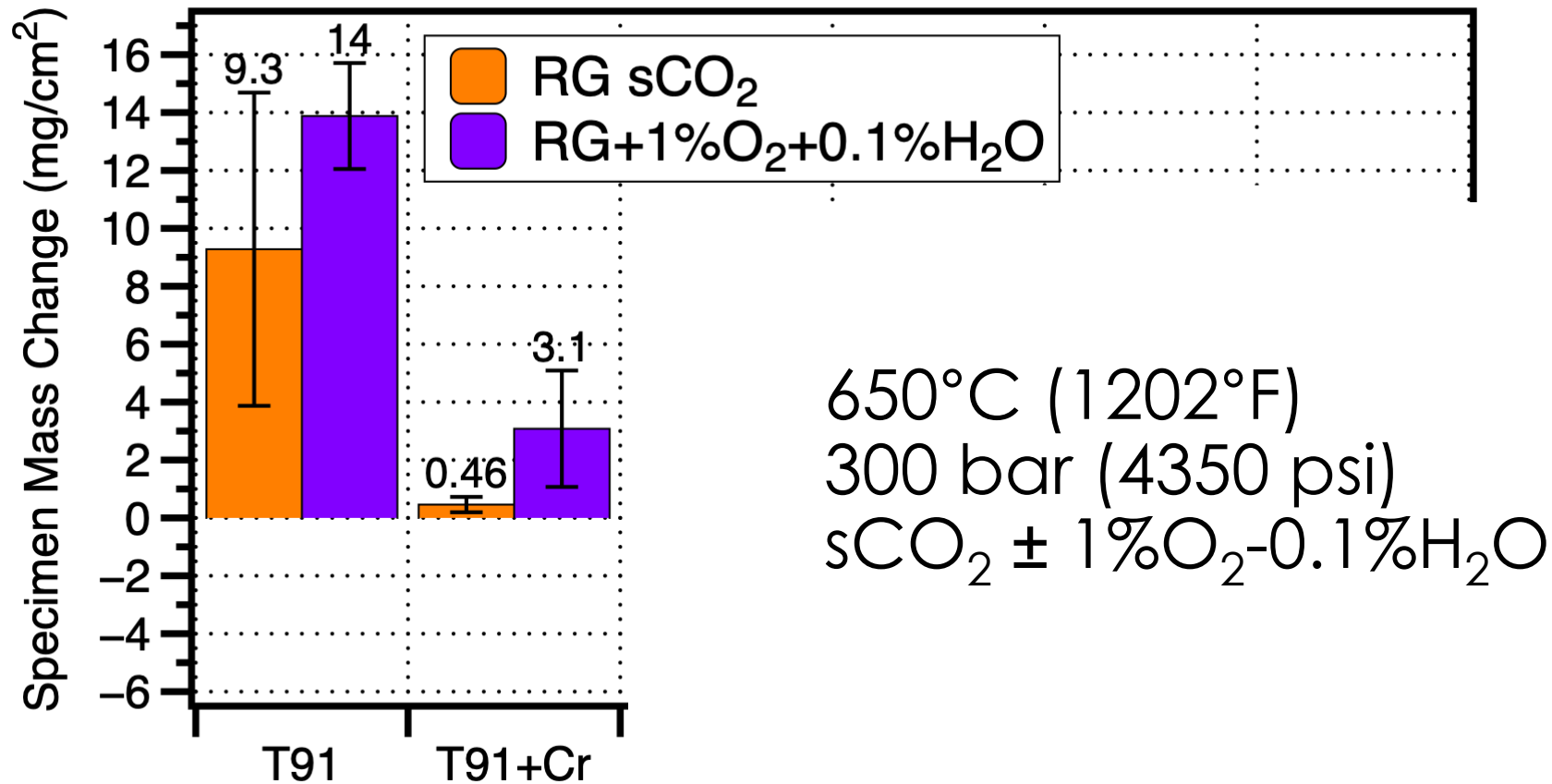


- T91: Fe-9Cr-1Mo
- 316H: Fe-16Cr-10Ni
- Massive C uptake at 650°C for both alloys
 - Reflected in bulk C measurements
 - And loss of room temperature ductility

Can coatings help inhibit C uptake?

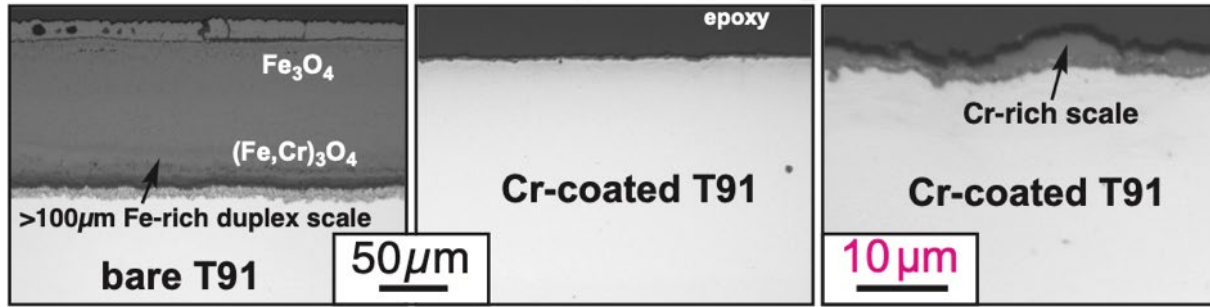


Initial sCO₂ testing at 650°C ± O₂/H₂O with pack Cr coating

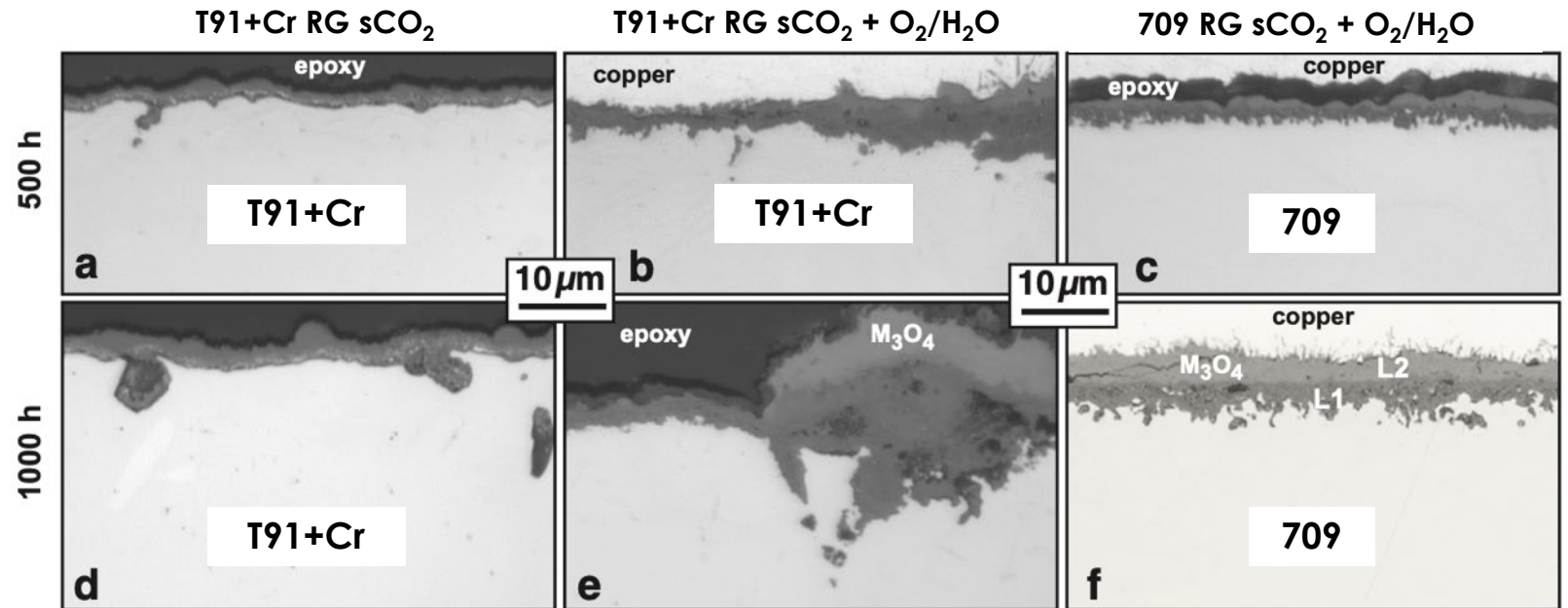


- Pack Cr coating by industrial partner: no details available
- High mass gains for uncoated 9Cr steel: variable due to spallation
- Low mass gain in RG sCO₂ but mass gain increased with impurities

All 650°C, 500 h, sCO₂

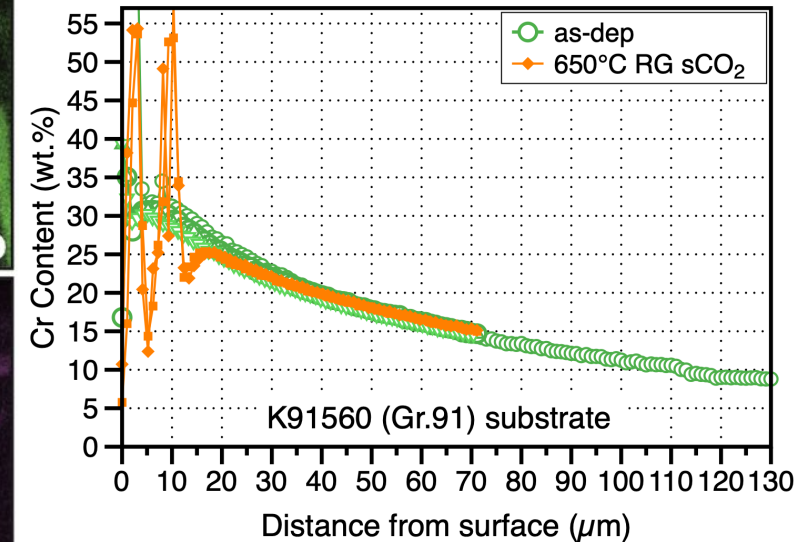
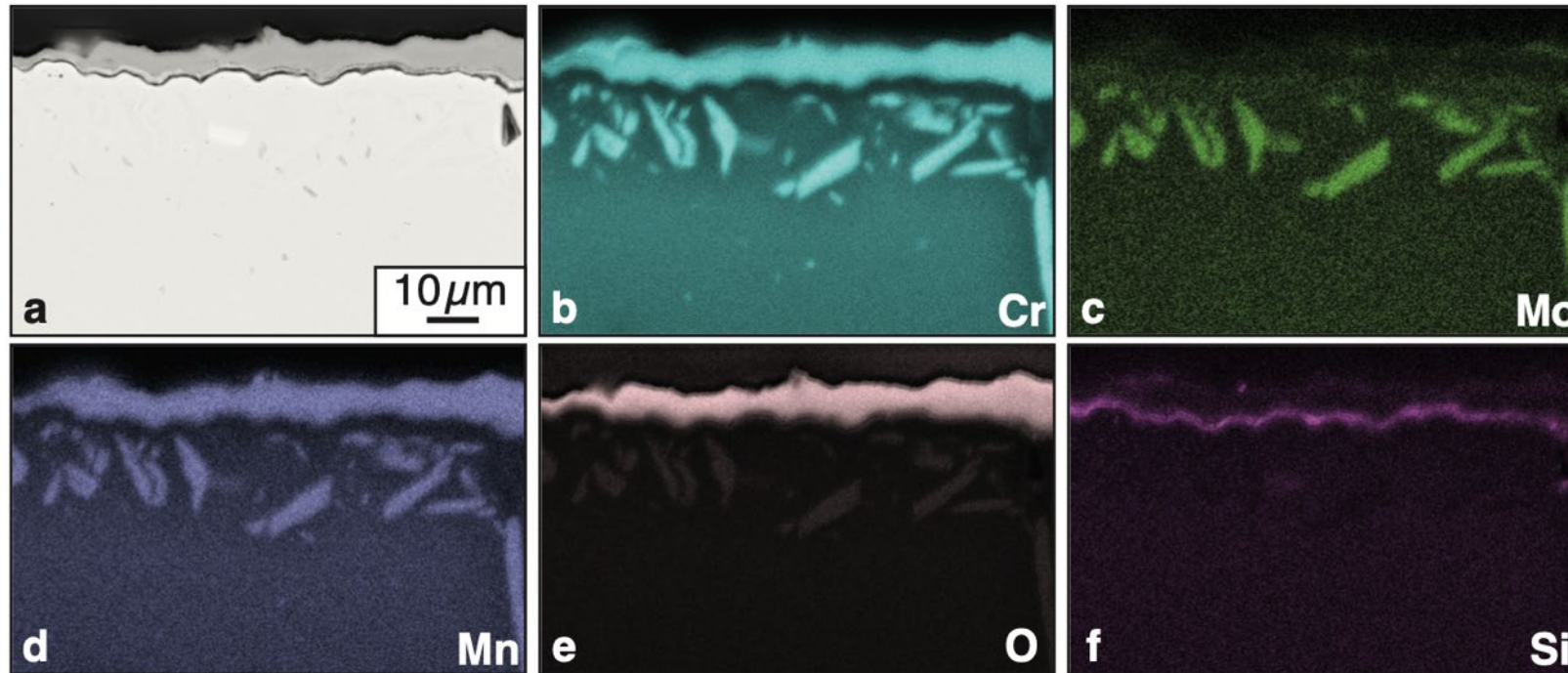


Cr coating on T91 more effective in RG sCO₂



- Thin scale formed in RG sCO₂ : not thick Fe-rich oxide
- Oxide nodules formed with O₂/H₂O additions

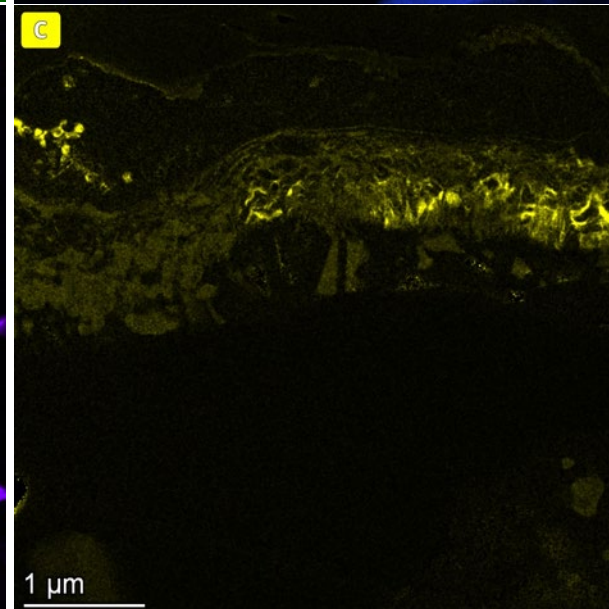
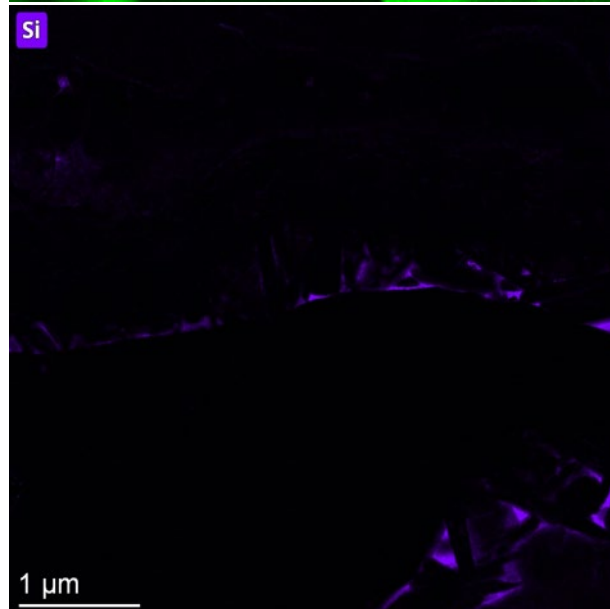
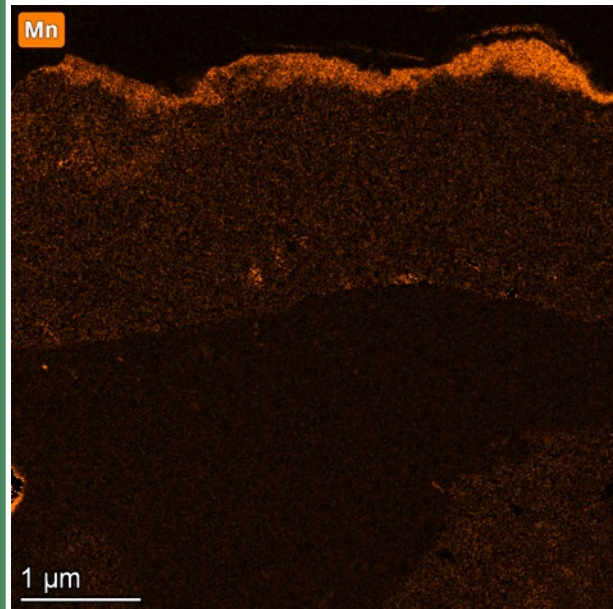
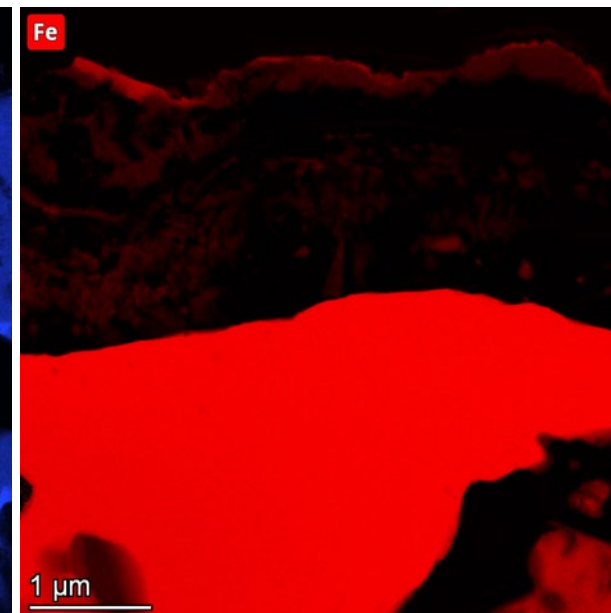
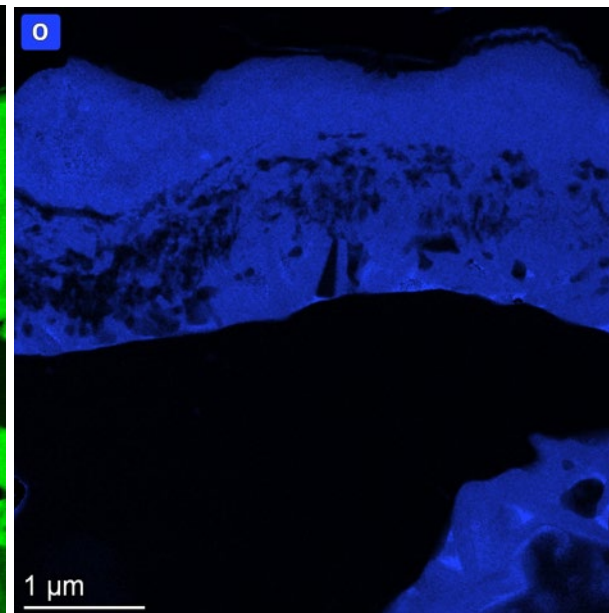
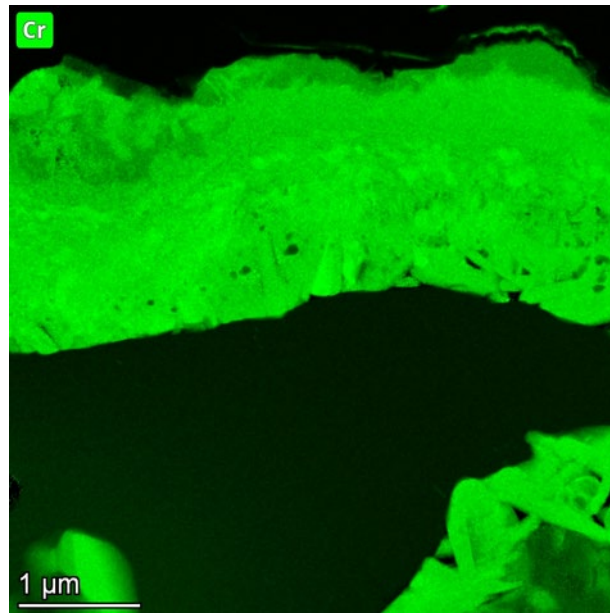
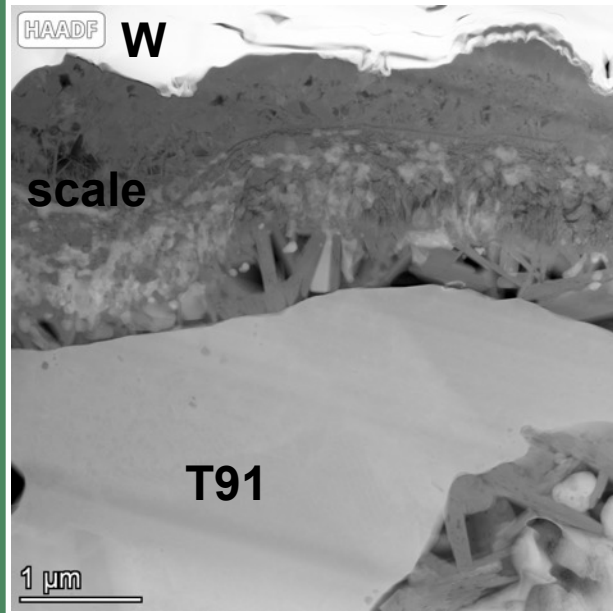
Precipitates in coating after 1,000 h at 650°C in RG sCO₂



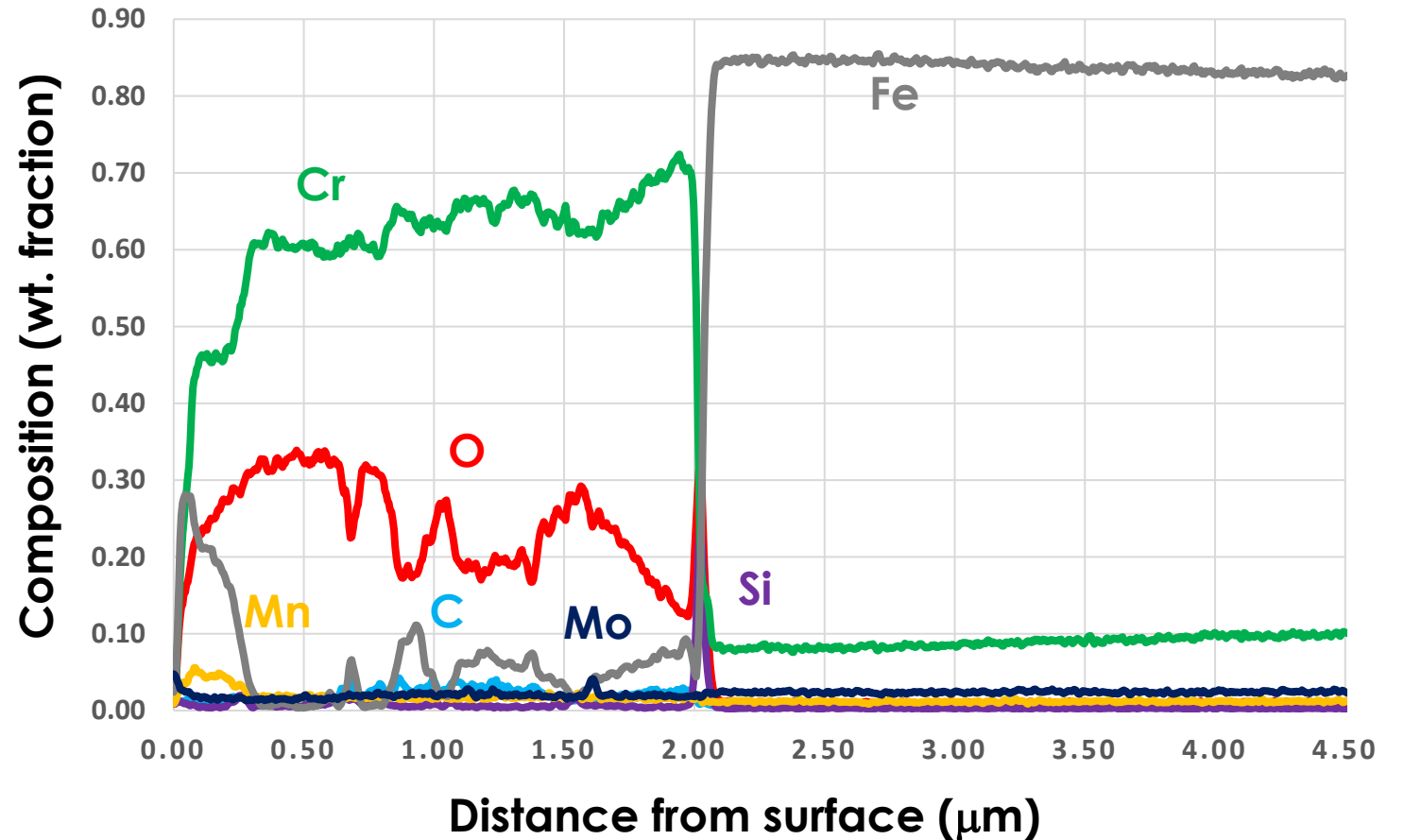
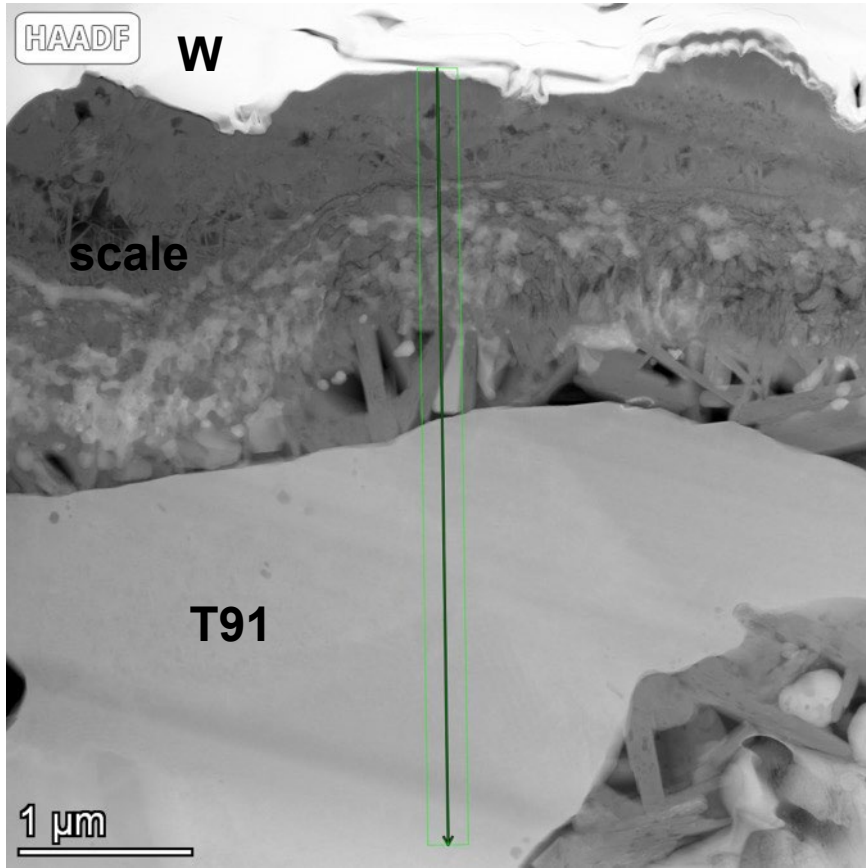
SEM/EDS Cr profile in T91

- Coating ~110 μm thick
- Confirmed carbide precipitates by EPMA

STEM/EDS of T91+Cr 1kh/650°C sCO₂: C in Cr₂O₃

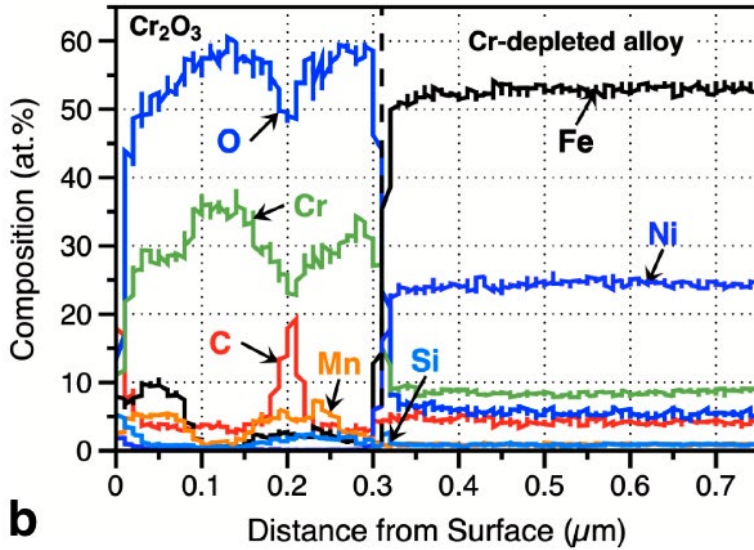
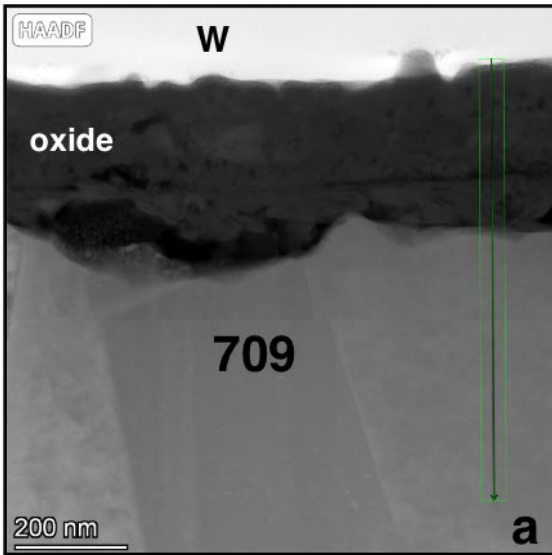


STEM/EDS of T91+Cr 1kh/650°C sCO₂: C and Mo in Cr₂O₃

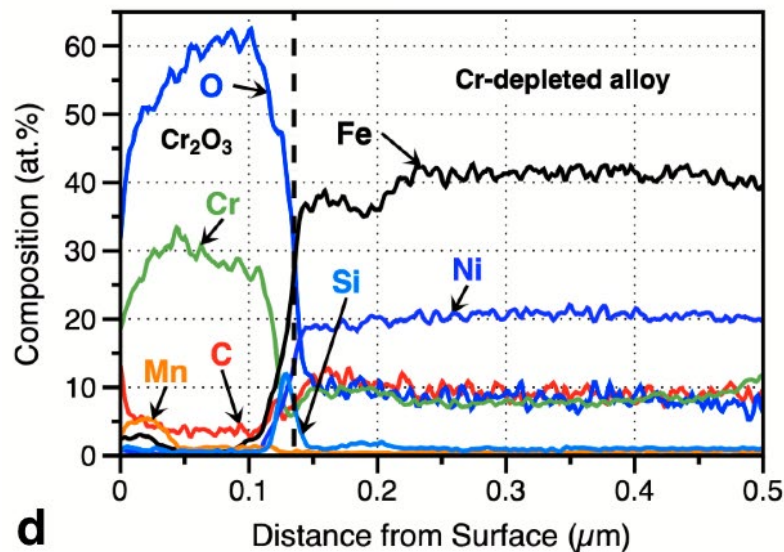
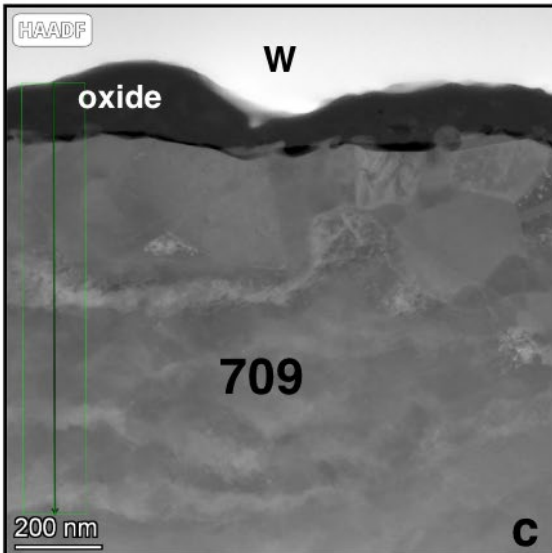


- Unusual area with oxidized precipitate under scale
 - <10%Cr left in coating at interface
- C and Mo within inner scale
- Fe and Mn in outer scale

709 500h: thicker complex scale at 300 bar, not 1 bar CO₂



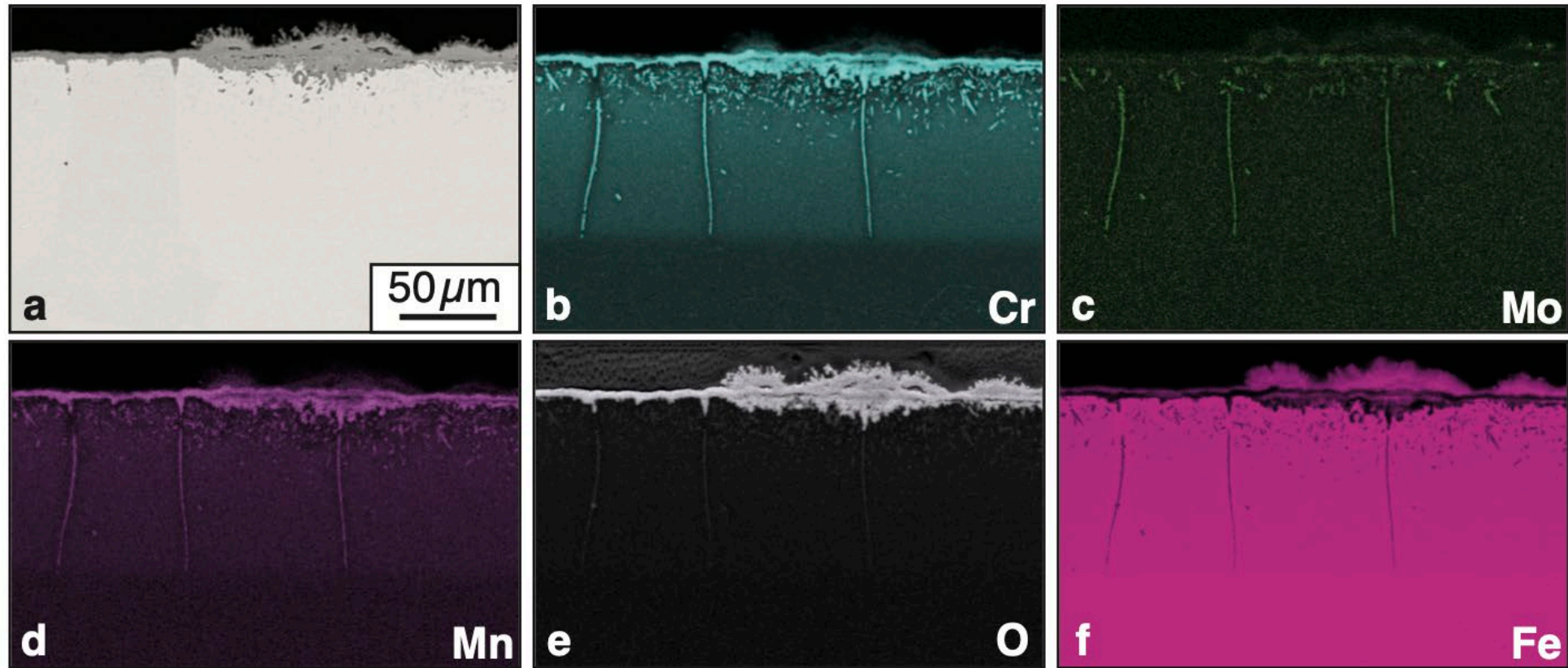
300 bar sCO₂



1 bar CO₂

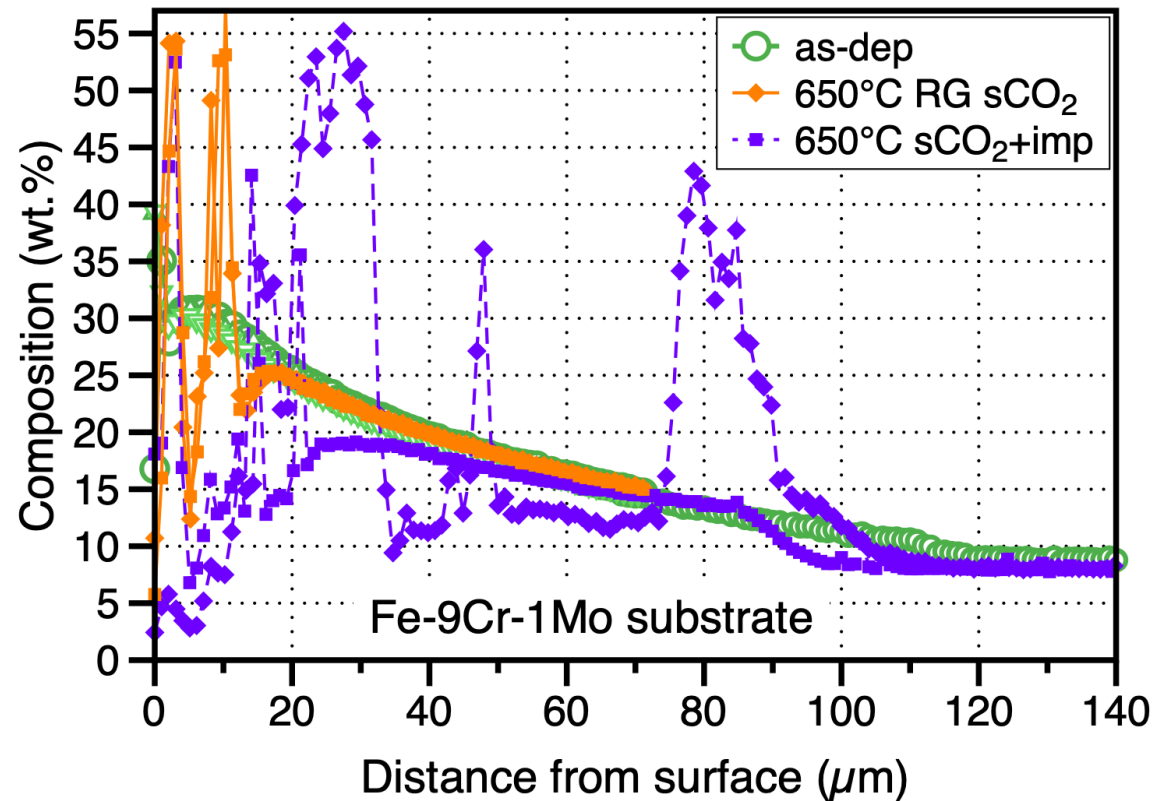
Heating to 650°C in sCO₂ may be an issue

T91+Cr, sCO₂ w/O₂+H₂O: forming Fe-rich oxide nodules



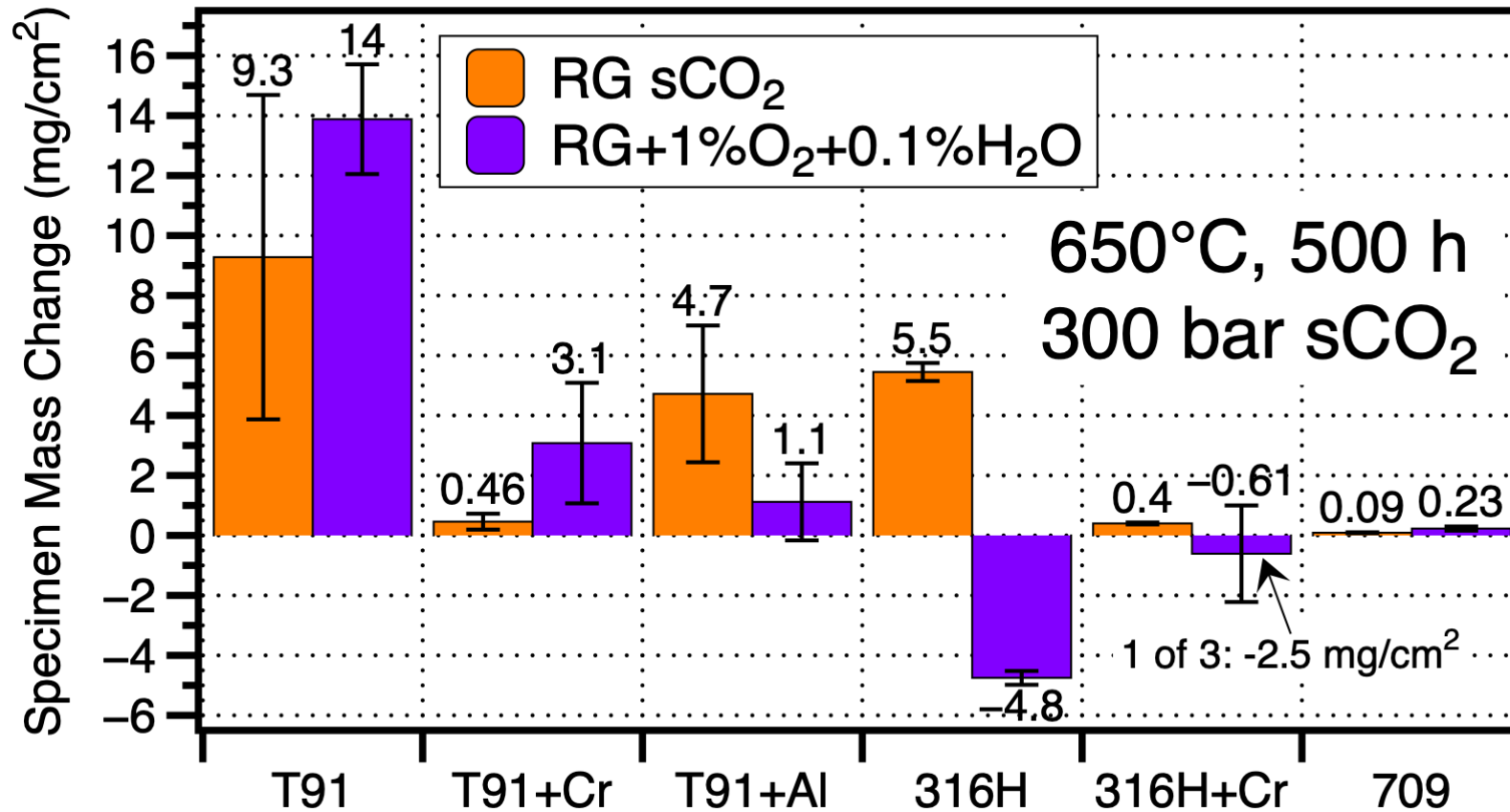
- Unable to form protective Cr-rich oxide with 1%O₂+0.1%H₂O
- Cr-rich precipitates also formed beneath scale after 1000 h

sCO₂ with impurities: significant Cr consumption



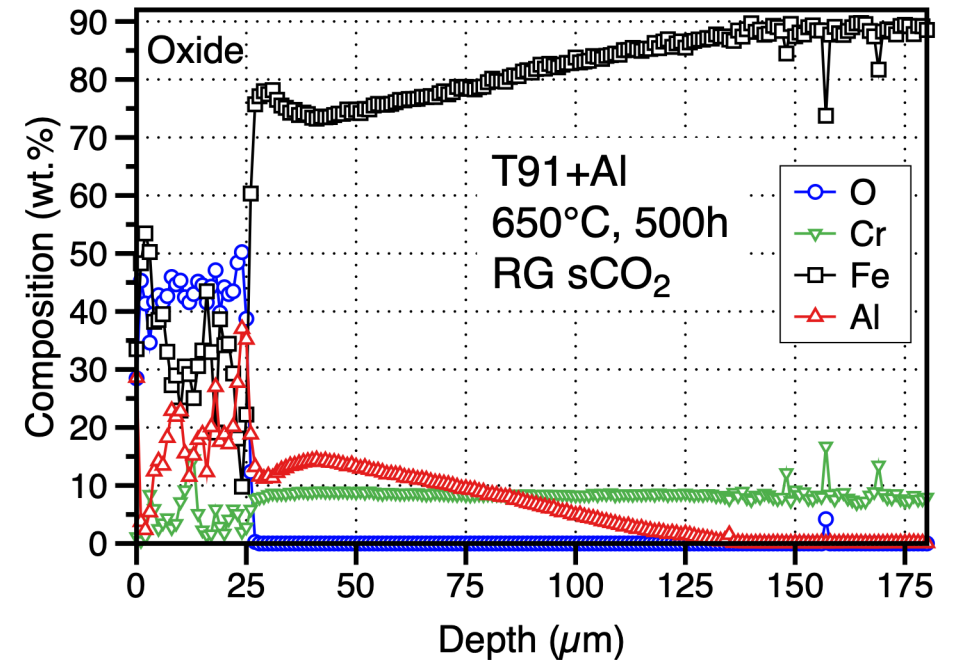
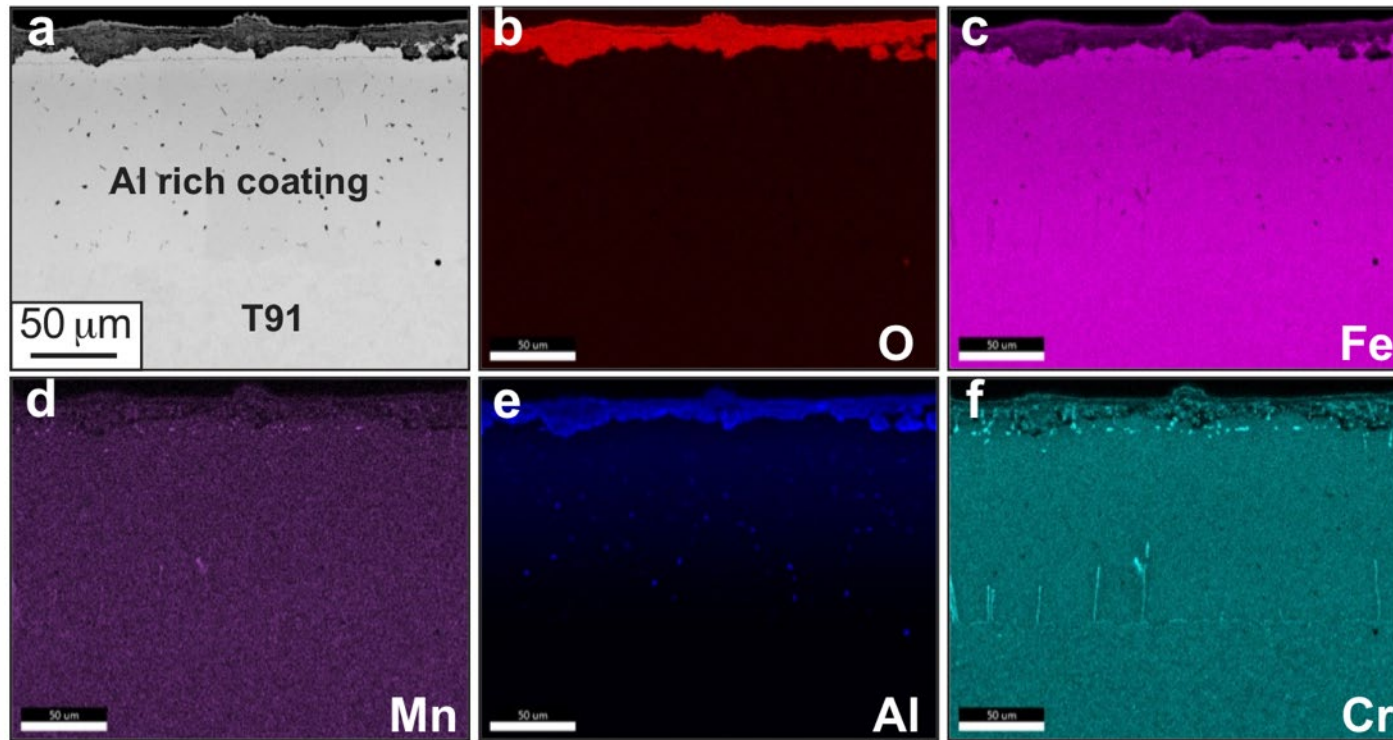
- 1000 h/650°C: higher Cr consumption with impurities
- 650°C: temperature too high for ~110 μm thick Cr pack coating

Second batch of pack coatings tested at 650°C



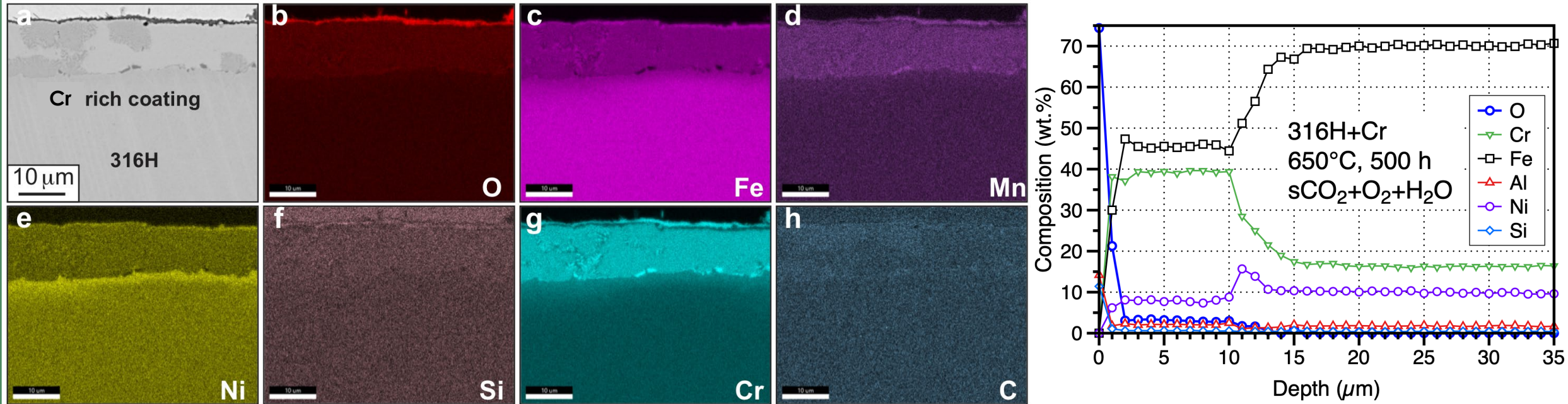
- Al at TTU: 30min at 1050°C, 20wt.%Cr-10%Al+2%NH₄Cl+78%Al₂O₃
- Pack aluminizing T91 did not work as well as Cr in RG sCO₂
- Pack chromizing 316H effective in RG sCO₂ at 650°C

T91: Pack aluminide formed mixed Al-Fe oxide at 650°C



- Targeted low $\sim 20\text{wt}\%$ Al coating with good CTE mismatch with T91
- High mass gain due to mixed Fe-Al oxide formation
 - 650°C is low temperature for selective Al oxidation
 - High Al coating may perform better

Chromized 316H: thin Cr-rich scale formed + precipitates

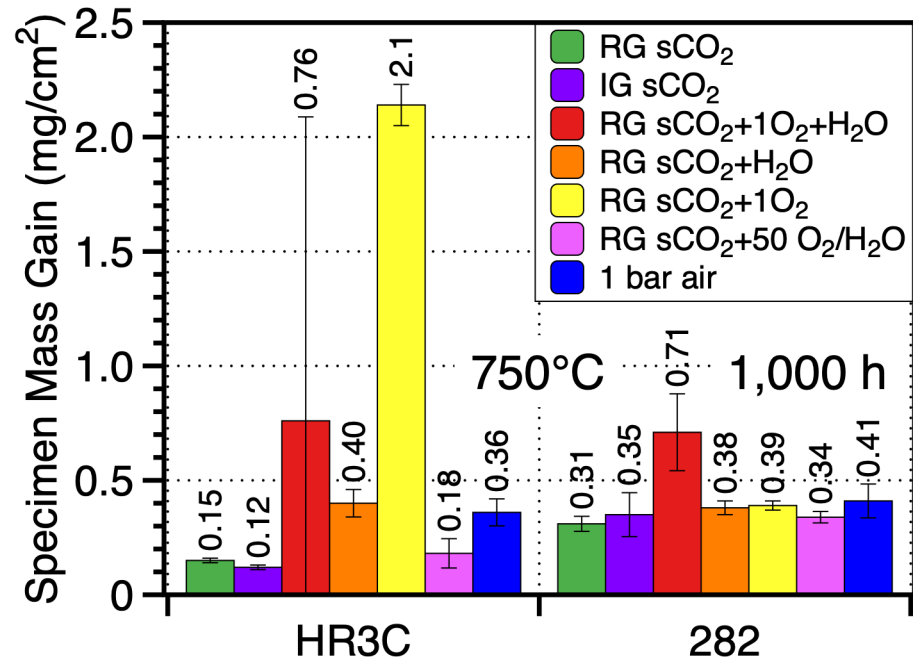


- Thinner coating formed on FCC 316H substrate
 - Again, coating made by company, no details on process
- Beneficial effect in RG sCO₂ at 650°C after 1,000 h
 - But Cr-rich precipitates are concerning for small Cr reservoir in coating

Summary: $s\text{CO}_2$ is a challenging environment

- All steels degrading at $550^\circ\text{-}650^\circ\text{C}$ in $s\text{CO}_2$
 - Increased attack when O_2 and H_2O impurities in $s\text{CO}_2$
 - Increased C uptake for Fe-rich oxides (vs. Cr-rich oxides)
 - Observed decrease in 25°C ductility for 316H in $s\text{CO}_2$ and 709 with impurities
- Opportunity for coatings?
 - Pack Cr and Al coatings evaluated on T91 and 316H
 - Processing (coating thickness + heat treatment) needs to be optimized
 - Cr-rich pack coating beneficial at 650°C
 - Thicker coating tested on T91 compared to 316H
 - Reduced mass gain in $s\text{CO}_2$ but internal carbides observed
 - With impurities at 650°C : significant Cr consumption after only 1000 h
 - Pack aluminized T91 (~20 wt.% peak Al)
 - Not forming a thin Al-rich scale on surface after 500 h at 650°C
- Longer exposures needed at lower temperatures (550°C)

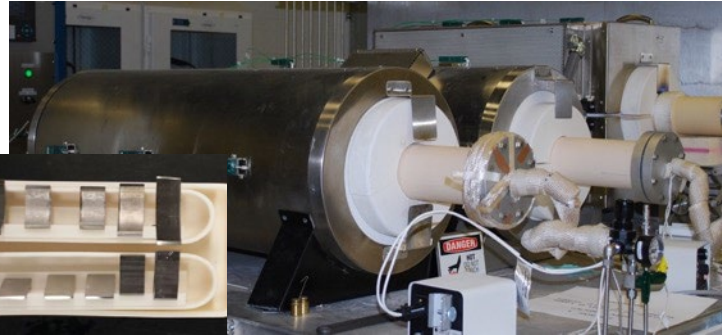
Thank you!



RG sCO₂+1%O₂+0.25%H₂O or +0.25%H₂O or +1%O₂

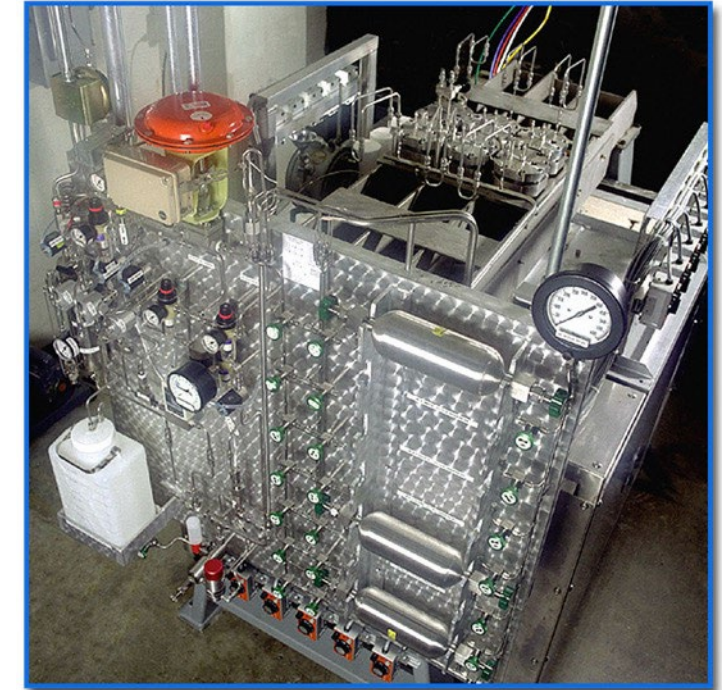
sCO₂ compatibility focused on steels at 450°-650°C

Autoclave: 300 bar sCO₂
500-h cycles



Same cycle frequency as autoclave

"Keiser" rig:
500-h cycles, 1-43 bar CO₂



Correct temperature and pressure

4-5 cm² alloy coupons

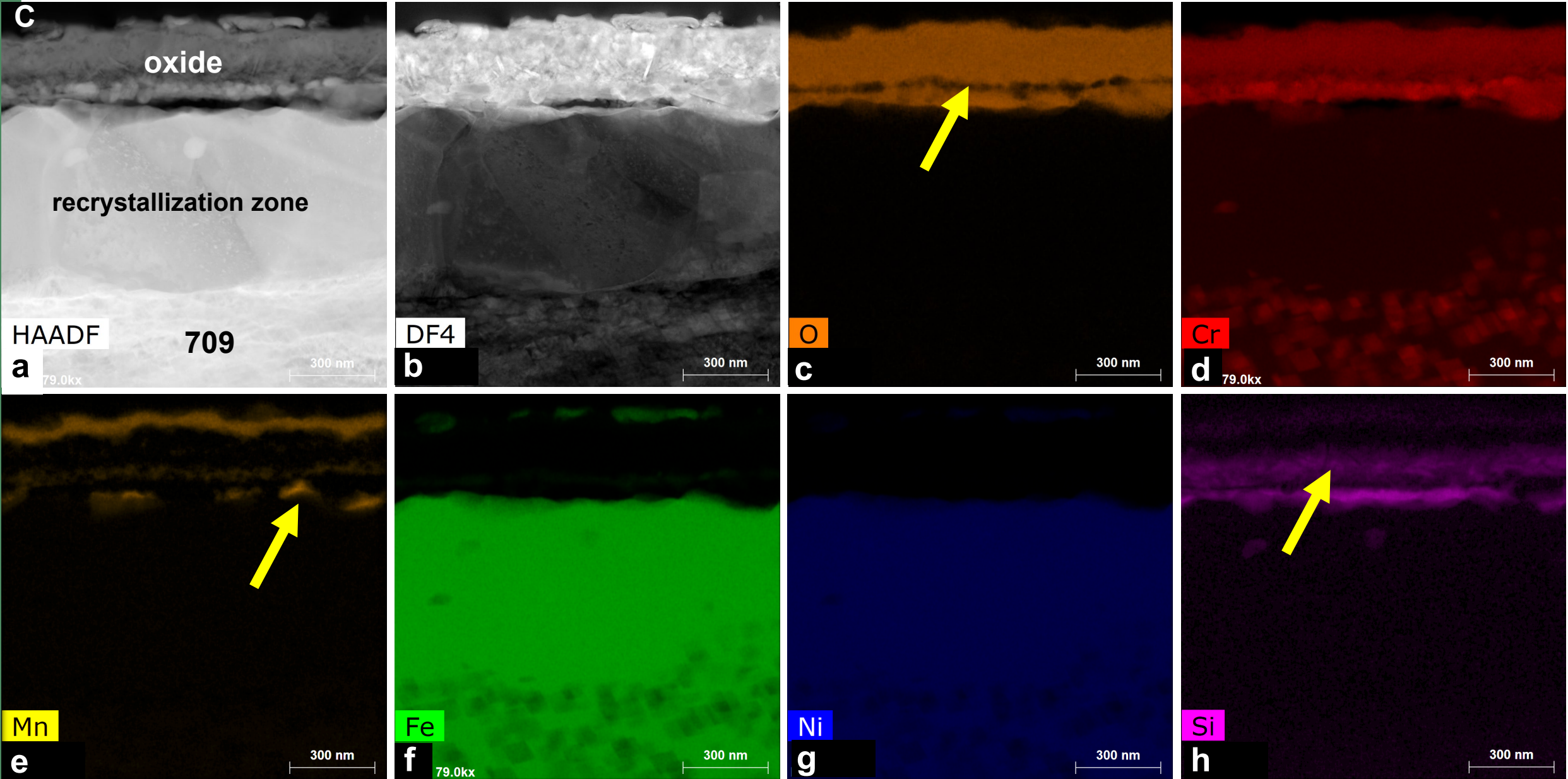


Box furnace:
Lab. Air
500-h cycles
(baseline)

Studies at 800°-1200°C

Baseline of research grade (RG) CO₂: ≤ 5 ppm H₂O and ≤ 5 ppm O₂
industrial grade (IG) CO₂: 18±16 ppm H₂O and ≤ 32 ppm O₂

709/sCO₂, 650°C/1000 h: Talos EDS maps of thicker multi-layer scale



709/sCO₂, 650°C/1000 h: Talos EDS line profile of multi-layer scale

