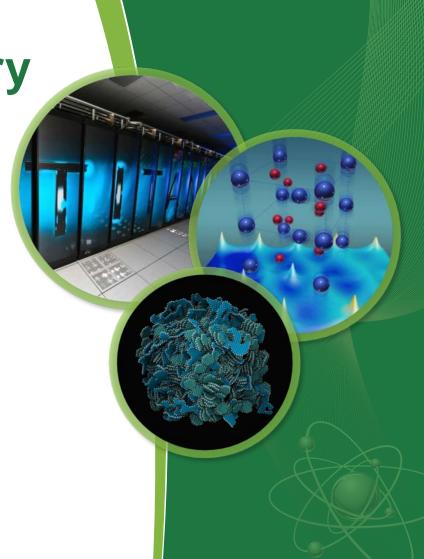
sCO₂ Panel Input from Oak Ridge National Laboratory

B. A. Pint

Corrosion Science & Technology Group Materials Science & Technology Division



ORNL's Materials Science and Technology Division Spans Basic to Applied Research

- ~\$100 M Budget
- ~230 staff + ~60 postdocs
- 20 research groups
- basic to applied research in structural and functional materials
- noted for technology development and transfer
 - 73 R&D 100 Awards
 - ~60 invention disclosures/year
 - ~30 patents filed/issued/year







ORNL has several capabilities for studying CO₂ corrosion



- 1. Automated Cyclic Rigs (1 bar)
- 2. 3-zone tube furnaces (1 bar)
- 3. 282 autoclave (300 bar)

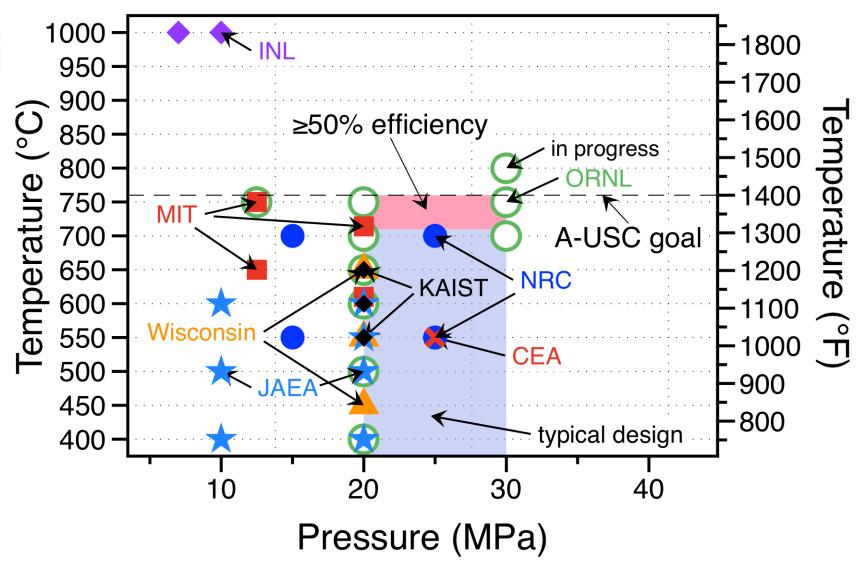








New temperature-pressure capabilities in new autoclave

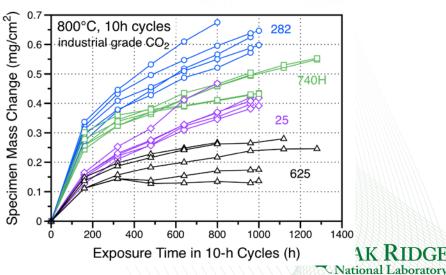




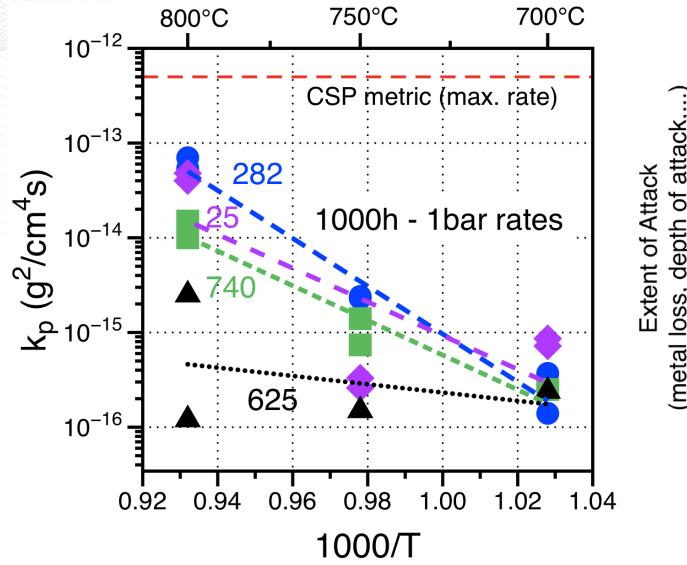
What now?

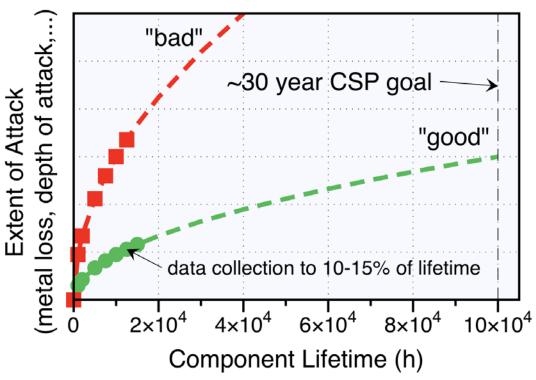
- Larger autoclave (more specimens at ±3°C)
- Controlled impurities (H₂O + O₂) at 300 bar
 - On-line detector system for H₂O and O₂ in sCO₂ at 300 bar
 - Operate system in FY17
- SunShot (CSP) Lifetime modeling project (FY16-FY18)
 - Long-term exposures:
 - 1 bar CO₂, 10-h cycles, 700°-800°C
 - 1 bar CO₂, 500-h cycles, 750°C
 - -300 bar CO₂, 500-h cycles, 700°-800°C





SunShot project has metrics to meet 30 year (100 kh) life







Key questions

- Does lower C solubility in Ni-base alloys make them "immune"?
- Limitation of Fe-base alloys (e.g. C-steel, low-Cr steel, 18-8 SS)
 - Does carburization restrict temperature? Fear of breakaway?
- Corrosion allowance for thin-walled components?
 - Oxide thickness may not be extent of damage
- Mechanical properties of thin-walled components (creep debit?)
- Is erosion the real problem?
 - Fluid or debris? Is exfoliation of reaction product an issue?
- How do these more complex issues get addressed?
 - "cross-cut" appears to be lacking coordination on a technical level